

Cesium-induced Ethanol Synthesis by CO₂ Hydrogenation on Cu/ZnO(000 $\bar{1}$) Surface

Scientific Achievement

Researchers demonstrated that Cs doping on a Cu/Zn(000 $\bar{1}$) surface, a model CO₂ conversion catalyst, opened a direct catalytic pathway to ethanol synthesis by tuning selective binding of key intermediates to enable C-C coupling and to improve methanol synthesis.

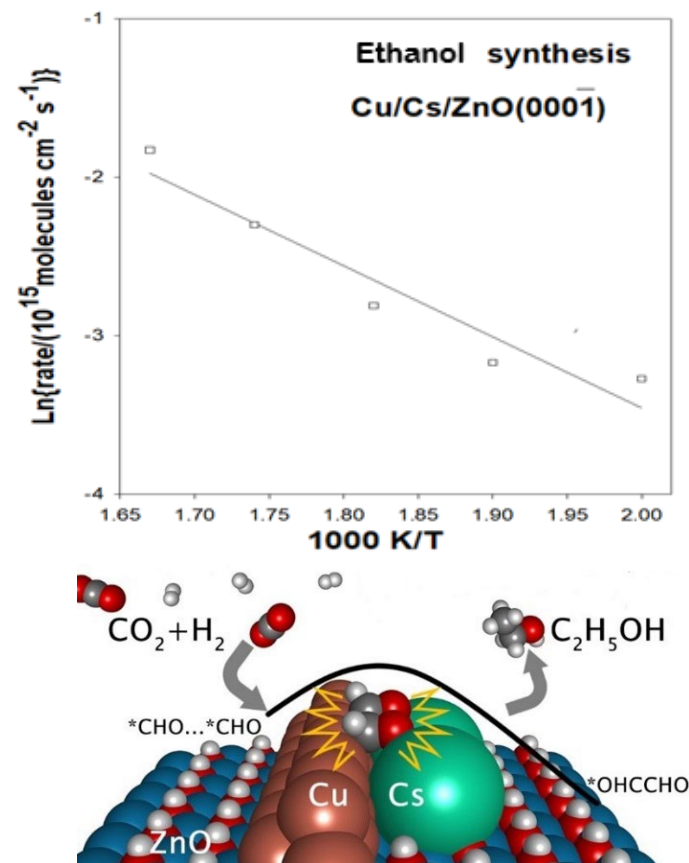
Significance and Impact

Conversion of greenhouse gas CO₂ into higher alcohols is desirable for sustainable fuels. The Cu/ZnO/Al₂O₃ catalyst is the dominant industrial catalyst for methanol synthesis by CO₂ hydrogenation. This work shows Cs doping of the catalyst has promise for higher alcohols and gives a mechanistic basis for further improvements.

Research Details

- Catalytic testing confirmed ethanol synthesis activity.
- X-ray photoemission spectroscopy (XPS) identified a change in the reaction mechanism for CO₂ to oxygenates by Cs deposition on Cu/ZnO(000 $\bar{1}$).
- Density functional theory (DFT) and kinetic Monte Carlo (KMC) simulation showed Cs-tuned binding of *CHO to facilitate further hydrogenation to methanol and open a new route toward ethanol synthesis via *OHC-CHO coupling.

X. Wang, P. J. Ramírez, W. Liao, J. A. Rodriguez, and P. Liu, *JACS*. 143, 13103 (2021) (Front cover) .



Top: Arrhenius plot for ethanol synthesis from CO₂ and H₂ at 0.5 atm of CO₂ and 4.5 atm of H₂; Bottom: Structures of Cu-Cs-ZnO interface on Cu/Cs/ZnO(000 $\bar{1}$) surface, active for C-C coupling and ethanol synthesis.