



Probing high local concentrations in track structure of high LET radiolysis

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Introduction



From stochastic system
... to deterministic solution.

Long term...

H₂

Corrosion

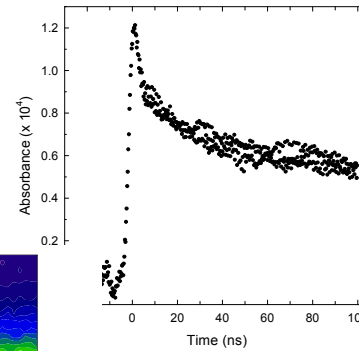
Solubilization (Waste management)

Radiobiological effects

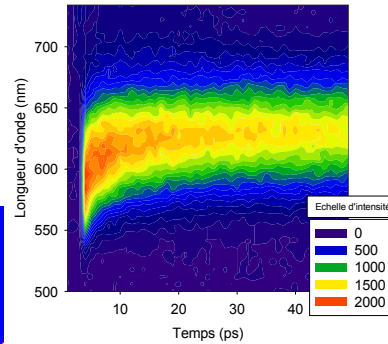
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Chemistry of solutions

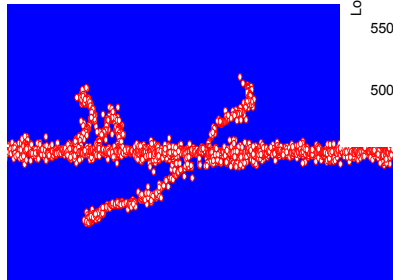
Chemical kinetics



Solvation



Ionization tracks

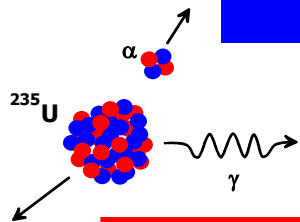


Local effect

Heterogeneity

Diffusion of species

time



0

10⁻¹⁵

10⁻¹²

10⁻⁹

10⁻⁶

1 s à 10000 years

LET

$$LET = - \left(\frac{dE}{dx} \right)_{\text{elec}} \quad \text{unit : keV}/\mu\text{m ou eV}/\text{nm}$$

Bethe formula :

$$-\frac{dE}{dx} = \frac{2\pi e^4 Z^2 N_z M}{E m} \ln \frac{4E m}{I M} = \frac{Z^2}{\beta^2} \frac{4\pi e^4 c^2 N_z}{m} \ln \frac{2m\beta^2}{Ic^2}$$

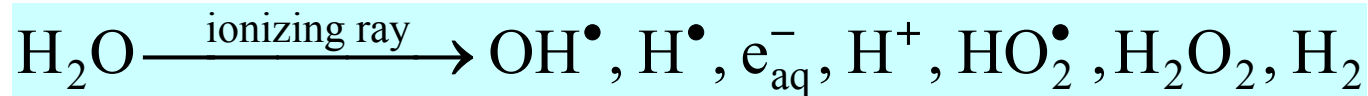
Orders of magnitudes in water:

e- (MeV) ou γ	: 0.2 – 0.5 eV/nm
β - (18 keV du ^3H)	: 2.7 eV/nm (mean value)
C^{6+} (1 GeV)	: 30 eV/nm
Ar^{18+} (3 GeV)	: 250 eV/nm
He^{2+} (5 MeV)	: 130 eV/nm (mean sur 35 μm)

Dose : Locally deposited energy for 1 kg of water
unit : Gy ou J/kg

Radiolytic yield (g) : conc. of produced (or consumed) species / dose
unit : mole/J

LET effect on the radiolytic yields



γ ^{60}Co rays
LET = 0.25 eV/nm

accelerated ions
LET > 0.25 eV/nm

e_{aq}^- $2.7 \times 10^{-7} \text{ mol/J}$
 OH^\bullet $2.7 \times 10^{-7} \text{ mol/J}$
 H^\bullet $0.6 \times 10^{-7} \text{ mol/J}$

e_{aq}^- $< 2.7 \times 10^{-7} \text{ mol/J}$
 OH^\bullet $< 2.7 \times 10^{-7} \text{ mol/J}$
 H^\bullet $< 0.6 \times 10^{-7} \text{ mol/J}$

HO_2^\bullet 0 mol/J

HO_2^\bullet $> 0 \text{ mol/J}$

H_2 $0.45 \times 10^{-7} \text{ mol/J}$
 H_2O_2 $0.7 \times 10^{-7} \text{ mol/J}$

H_2 $> 0.45 \times 10^{-7} \text{ mol/J}$
 H_2O_2 $> 0.7 \times 10^{-7} \text{ mol/J}$

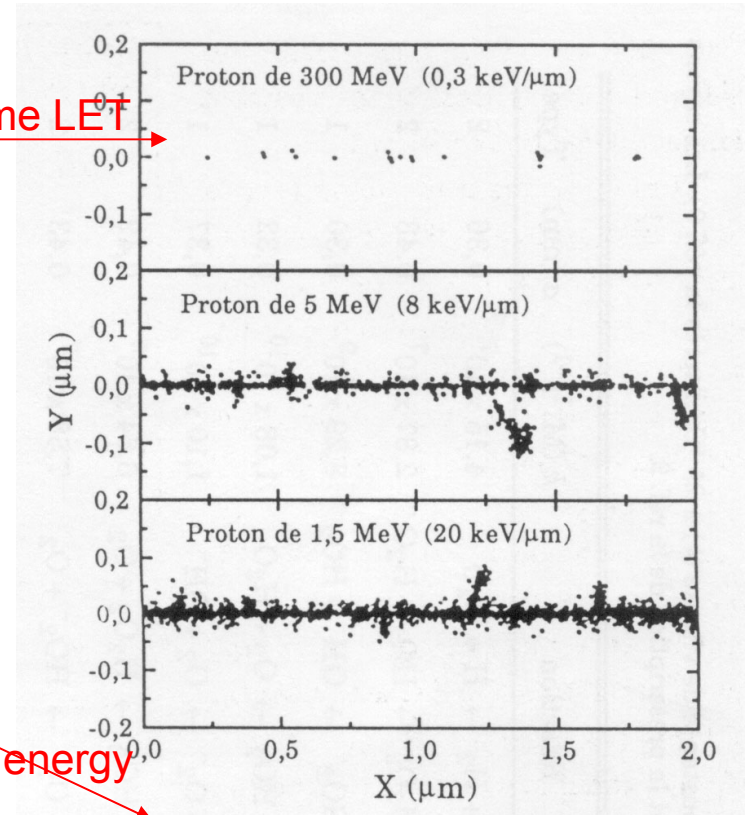
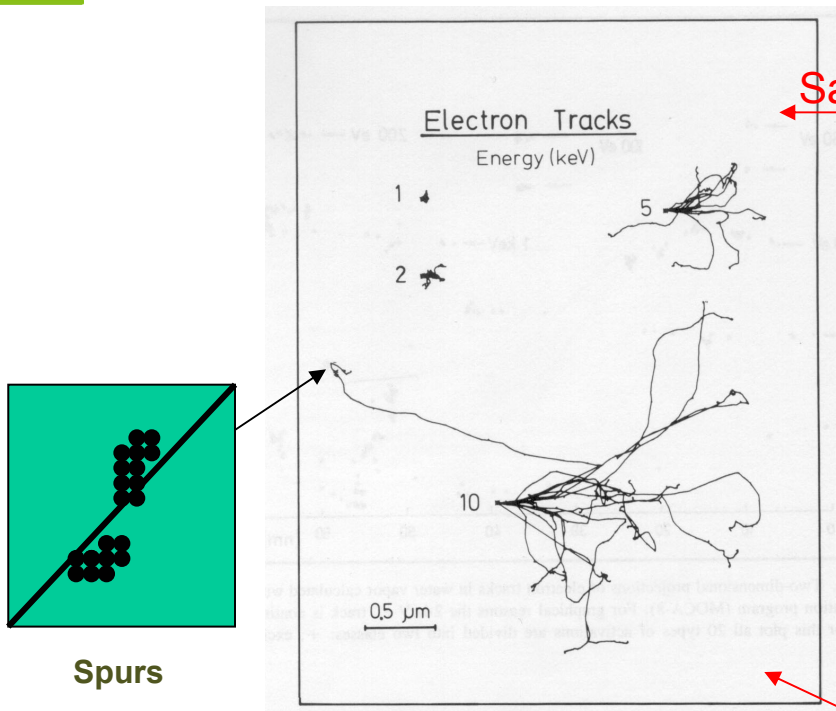
Structure of the tracks and LET values

When LET do not explain the results... how to access to structure of tracks?



Atomic number (Z)

Energy (β)



Same LET

Same energy

Freeman, G., 1987,
« Kinetics of non homogeneous processes »

Cobut *et al.*, J. Chim. Phys., 1995, (93) 93-100.

What is the chemistry with high concentration?
How to probe local high concentration at fs-ps?

Impossible to use « final products analysis » by using scavenging method

The concentration of scavengers should be $> 5M$

Possible direct effect on the solute

So, it remains real time methods : transient species

- Pulse radiolysis (absorption, fluorescence...), « μs -ns-ps limited »

Pump probe spectroscopy

Hydrated electron

An ideal probe of the track structure



Formation in the fs-ps time range

Properties: $\lambda_{\max} = 720 \text{ nm}$, $\varepsilon = 20000 \text{ M}^{-1} \text{ cm}^{-1}$

Many studies exist

Few time-resolved studies at high LET, direct detection

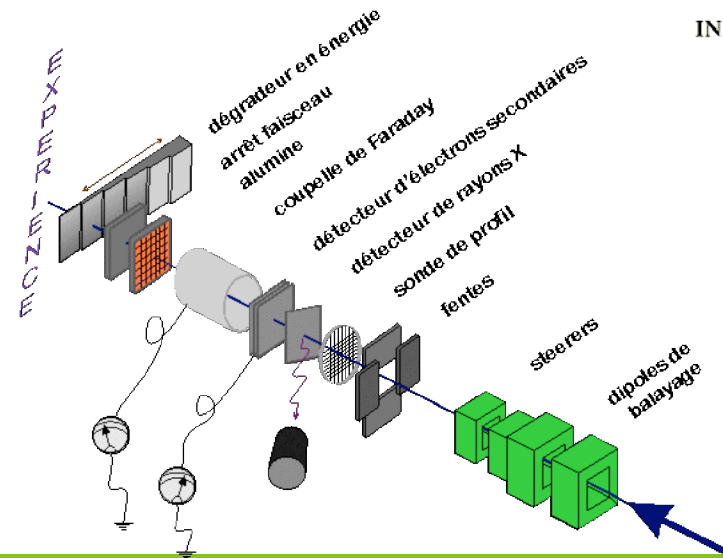
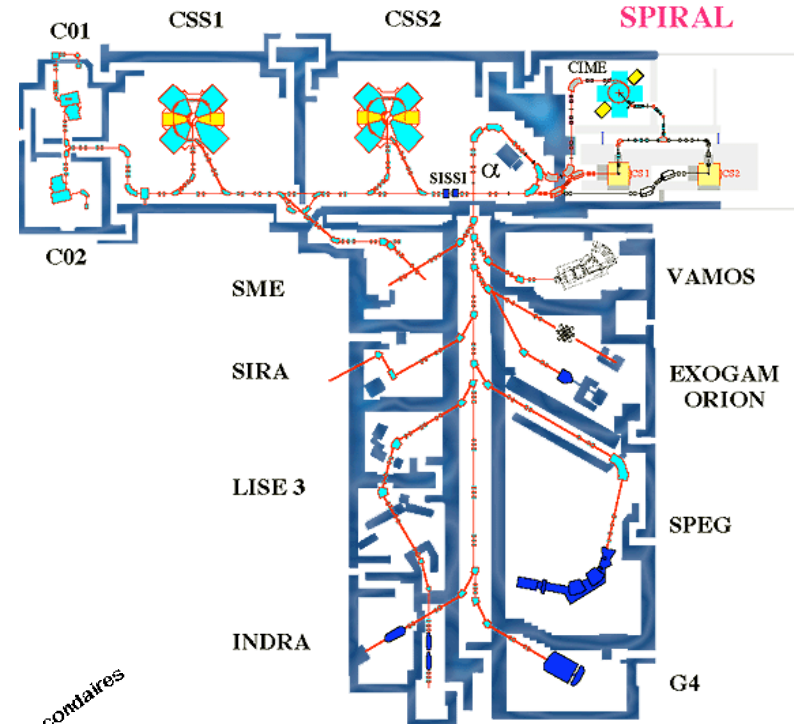
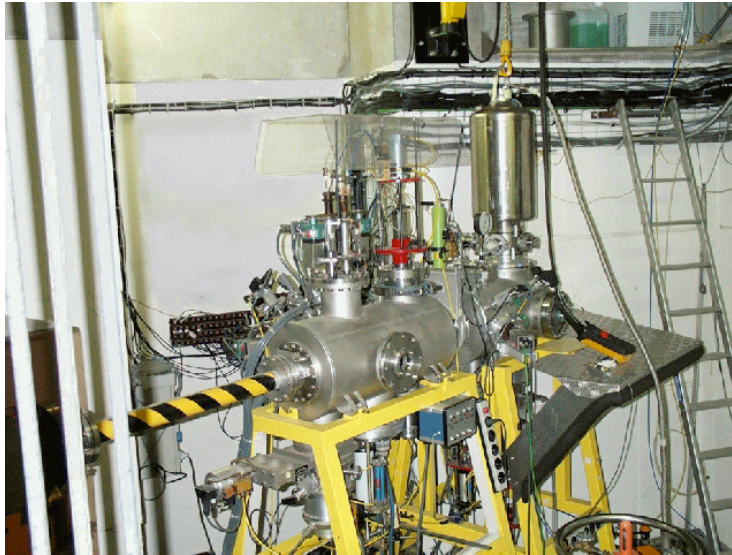
helions $10 \mu\text{s}$: Sauer et *al.*, 1977;

protons 1 ns : Buxton et *al.*, 1981

But: low concentrations at high LET

But: high concentration at early times

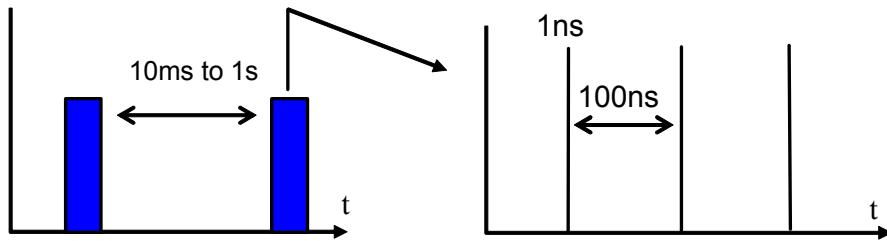
High energy ions at GANIL



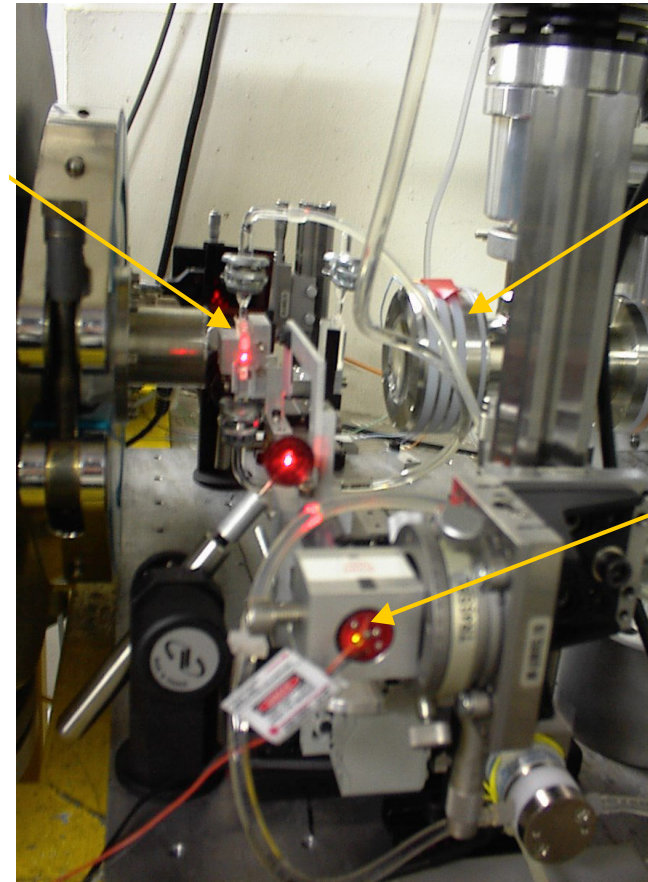
Pulse radiolysis setup



Temporal structure of the beam



Duration of the pulses : 1ns or $> 5\mu\text{s}$



Collimator & irradiation cell

Faraday Cup

Laser Diode 670 nm

Formation and decrease of hydrated electron

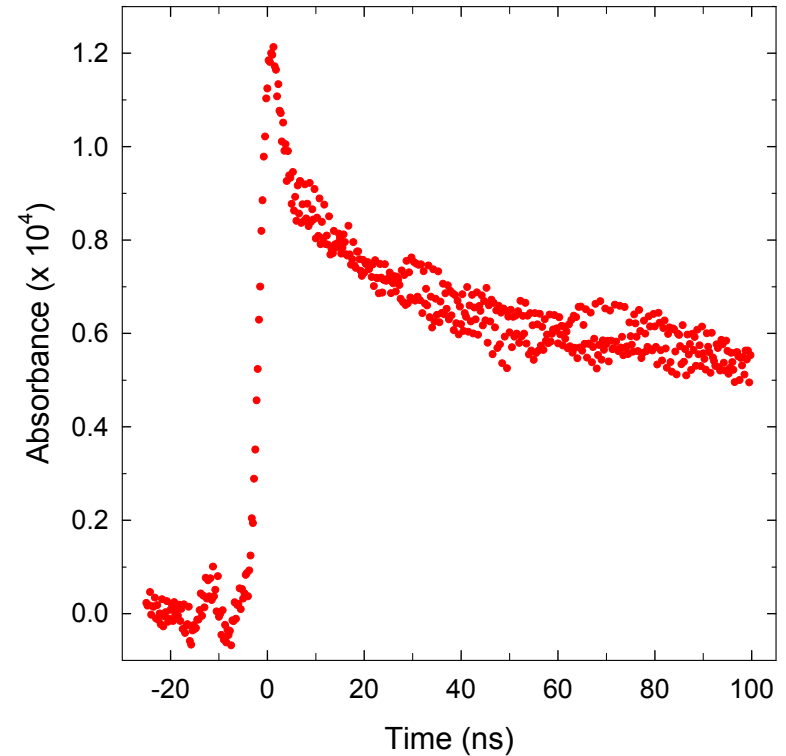


Experimental conditions

- C^{6+} , 95 MeV/amu, 27 eV/nm
- 1 ns, 1 kHz, 10^6 acquisitions
- De-aerated pure water
- dose = $6 \cdot 10^{-3}$ Gy/pulse

Results

- Low levels of absorption / conc.
- Fast reactions in 10^{-7} s
- $g_{1ns}(e_{aq}^-) = 4.5 \cdot 10^{-7}$ mol/J



Baldacchino et al., NIMB (2003) 209, 219-223

With 2 ions with same velocity



Incident ions

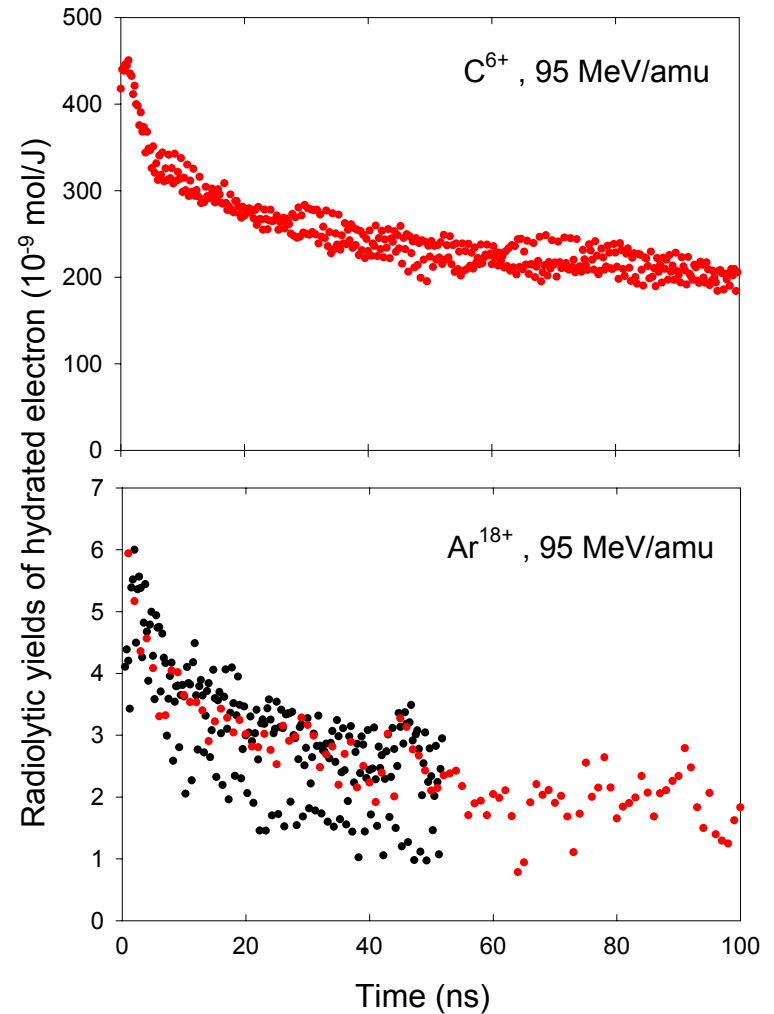
- C^{6+} , 95 MeV/amu, 27 eV/nm
- Ar^{18+} , 95 MeV/amu, 250 eV/nm
- => same velocity: $\beta = v/c = 0.32$
- => $TEL[Ar^{18+}] \# 10 \times TEL[C^{6+}]$

(i.e.: Bethe formula)

- => $Z^2[Ar^{18+}] \# 10 \times Z^2[C^{6+}]$

Results for e^-_{aq}

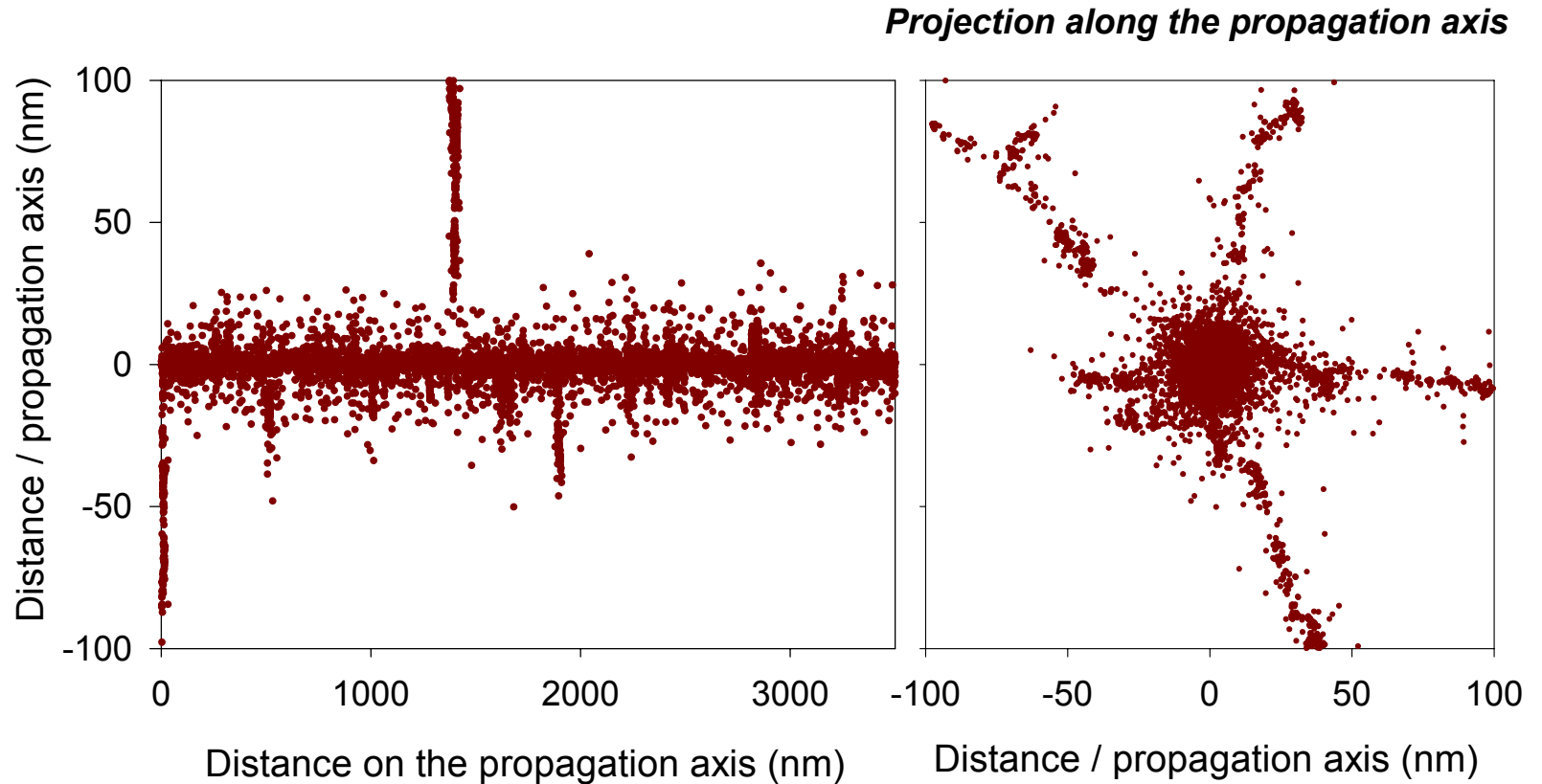
- $g_{1ns}[C^{6+}] \# 100 \times g_{1ns}[Ar^{18+}]$
- In 100 ns,
 - $g/3$ for Ar^{18+}
 - $g/2$ for C^{6+}



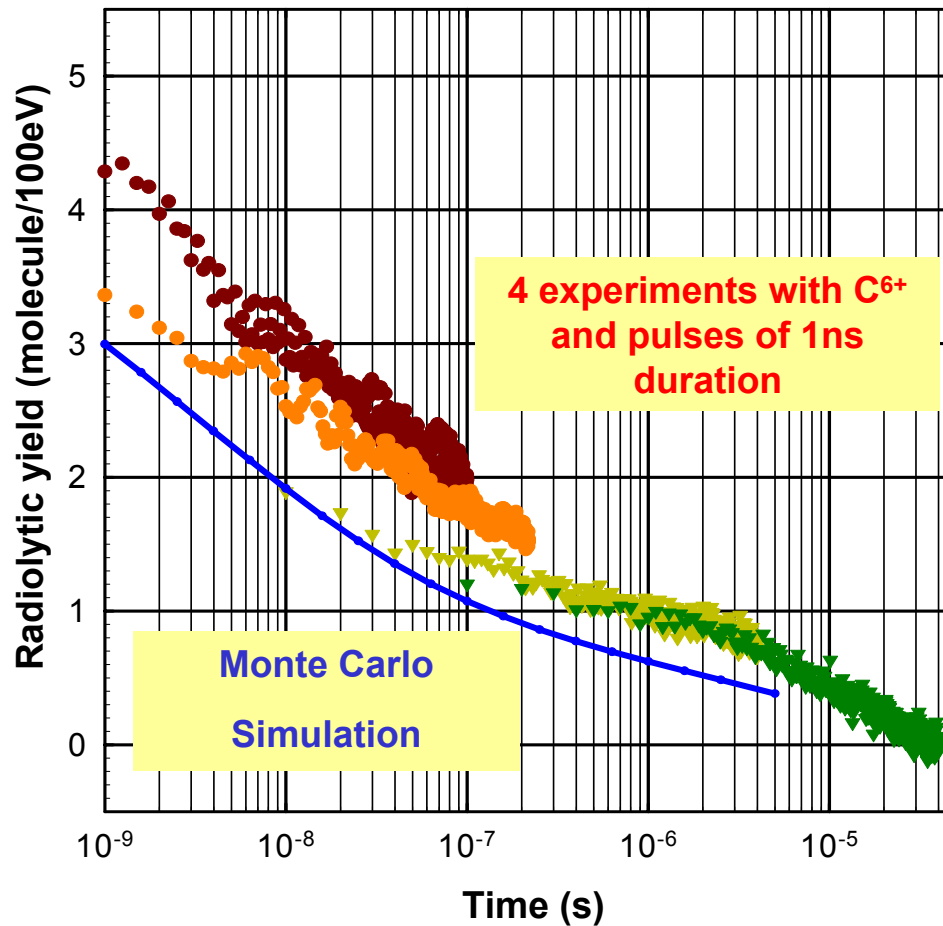
Baldacchino et al., Chem. Phys. Lett. (2004) 385, 66-71

Monte Carlo simulation

C^{6+} ions, position at $t=1ps$

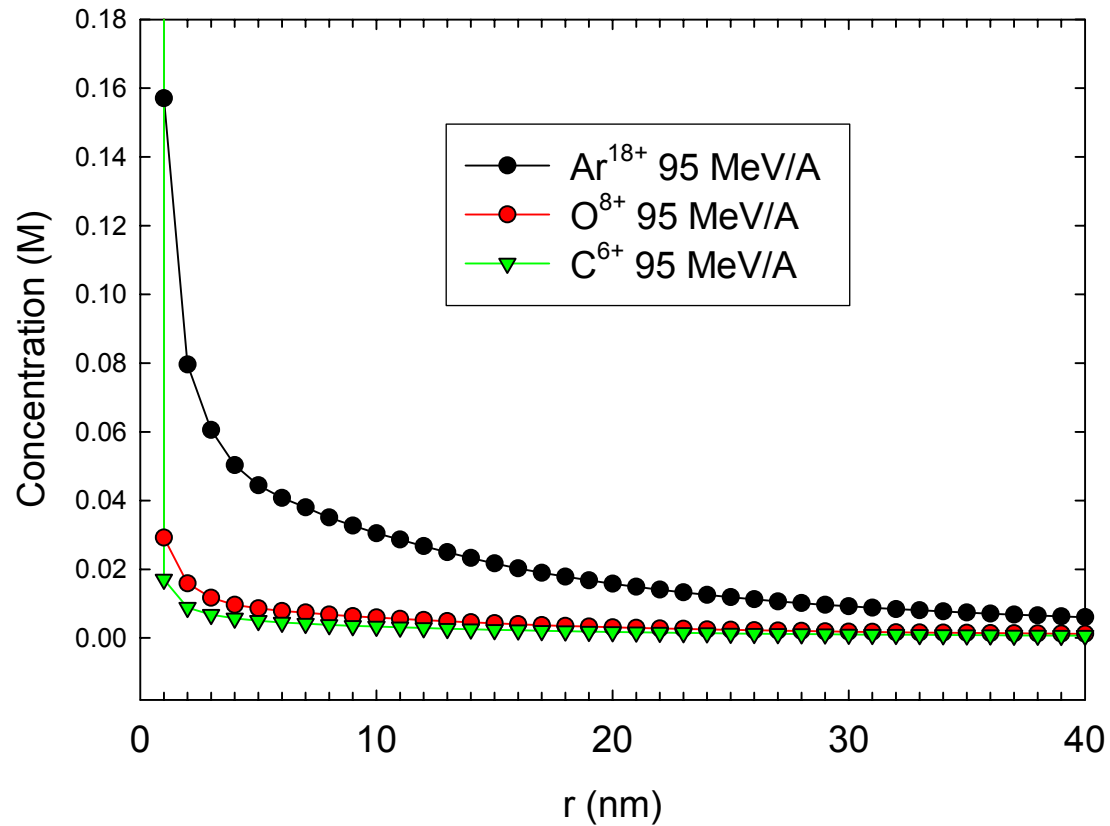


Comparaison Experiments / MC Simulation



Baldacchino et al., NIMB (2003) 209, 219-223

Concentrations of free radicals in the tracks



- **Results at 1ns:**

With C⁶⁺ to Ar¹⁸⁺

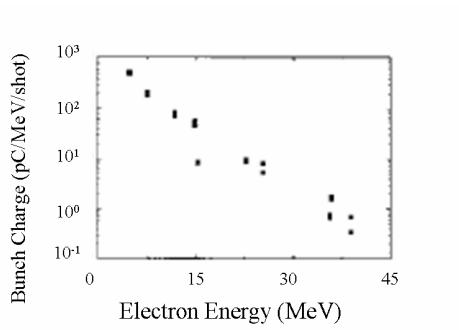
- **1ns – 100ns : complex kinetics**

Heterogeneous processes

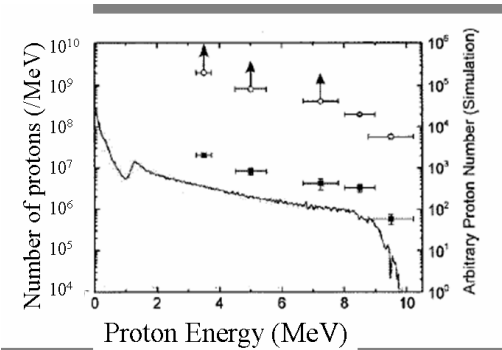
High concentrations at early times

- **Necessity to probe it with fs-ps pulses**

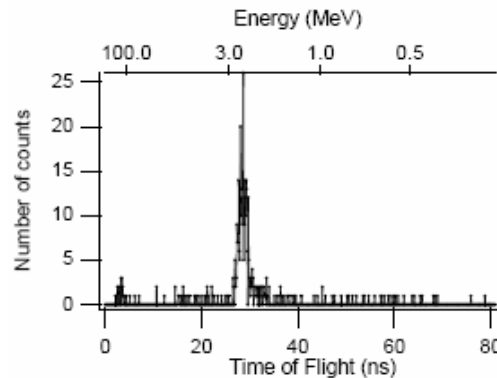
Electrons



Protons



Neutrons



Unique source able to produce e⁻, p, ions, neutrons, γ ?

Acceleration: $T_{eV} \propto E \times L$; laser: $E \uparrow L \downarrow$ proximité source cible

Projects of pulsed high energy ions accelerators



Femto-pico proton / neutron

- Based on T³-laser at DRECAM/Saclay (Cf Stanislas Pommeret)
- Pump probe experiments (« single shot » method?)

Over pulsed accelerators

Long term projects and intermediate time-resolution

Cyclotrons (He²⁺) : ns – μs (GANIL, Subatech Nantes)

**Developments of Linear Accelerator of p⁺, d⁺ :
high energy and 100 ps (GANIL)**