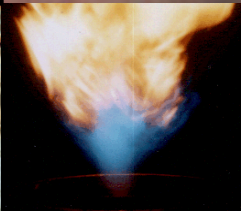
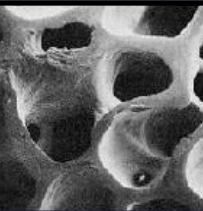
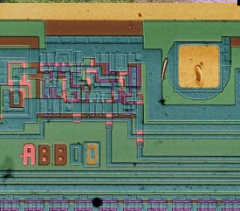
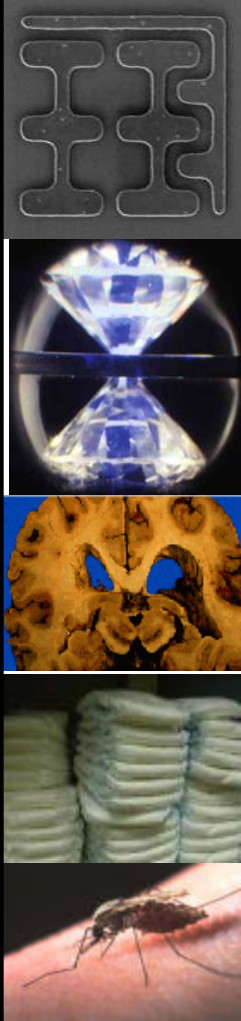


Light Source 101

With *many* thanks to Lisa Miller
Statistics by Kathy Nasta

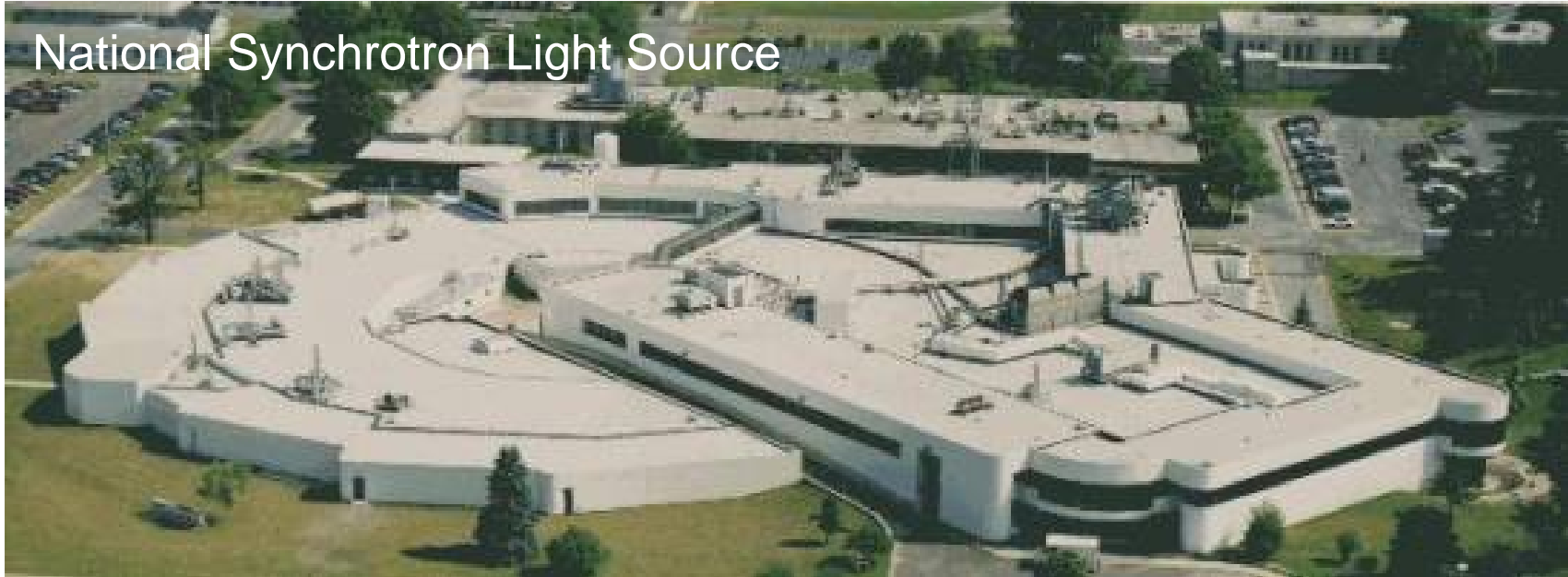
Jean Jordan-Sweet
December 9, 2010



Outline

- The Facility
- The Users
- The Science

Brookhaven National Laboratory



Commissioned in 1982(VUV)-1984(X-ray) , built for ~\$23M

What is a Synchrotron?

A synchrotron is a machine that produces tiny beams of very bright

LIGHT

by accelerating charged particles (electrons or positrons) around a “circular” track using magnetic fields

X-RAYS

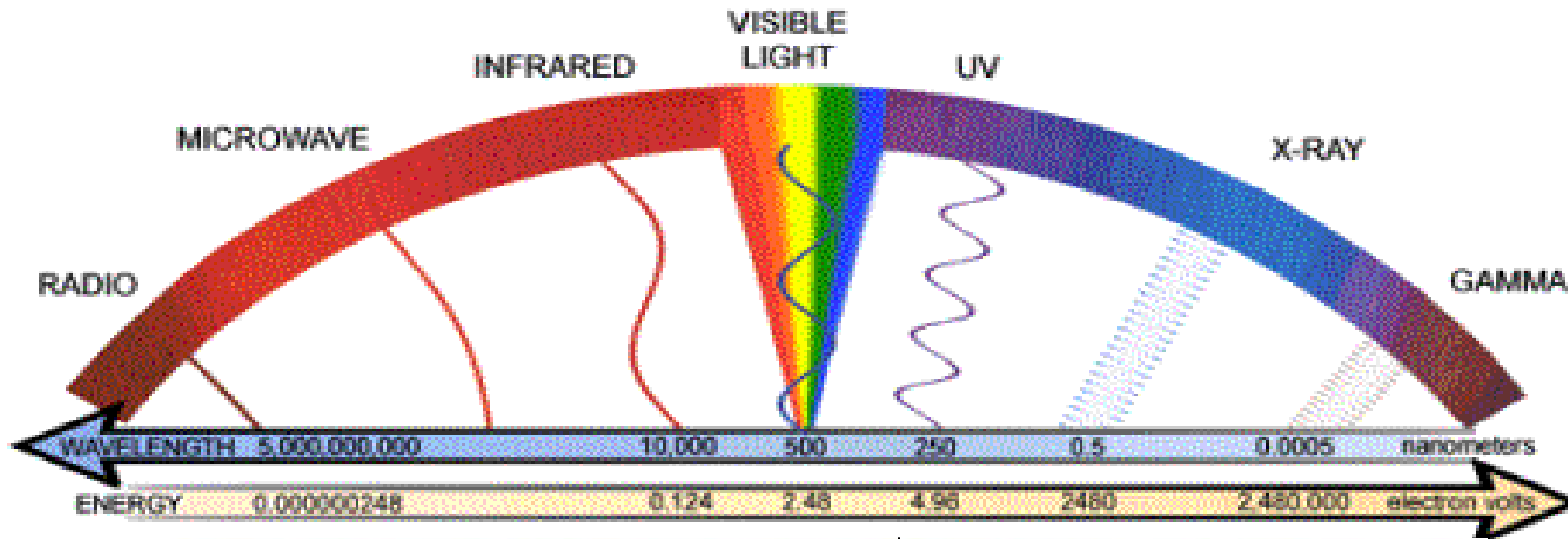
ULTRAVIOLET
LIGHT

VISIBLE
LIGHT

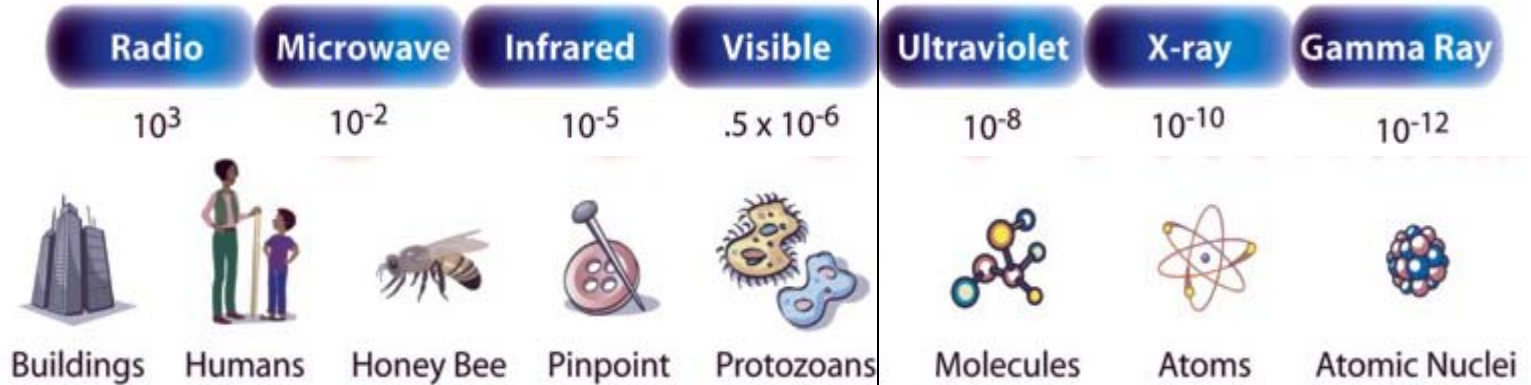
INFRARED
LIGHT

MICROWAVES

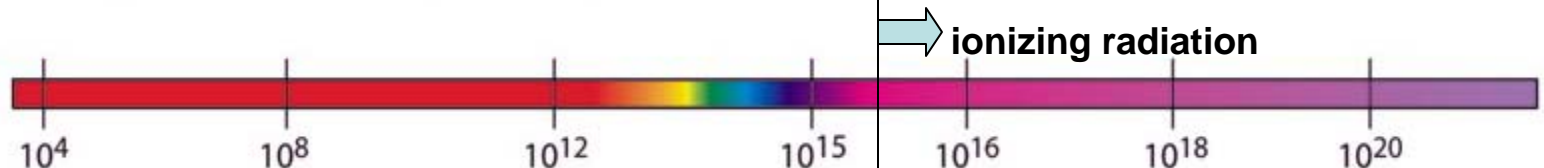
The Electromagnetic Spectrum



Wavelength (meters)



Frequency (Hz)



Early Synchrotrons

- First particle accelerators (cyclotrons) were used to “split the atom”.
- First synchrotron was built in 1947 by General Electric.
- Synchrotron radiation was given off by these accelerators and was seen to be a nuisance.
- In the 1960’s it was realized that this would be a useful source of radiation – a new light source.



North American Synchrotrons



4 are operated by the U.S. DOE

1st → 2nd → 3rd generations

The World of Synchrotron Radiation

Click an area or button for further listings of synchrotrons around the world.



- N. America
- S. America
- Europe
- Russia-Far East
- Japan
- Australia
- World



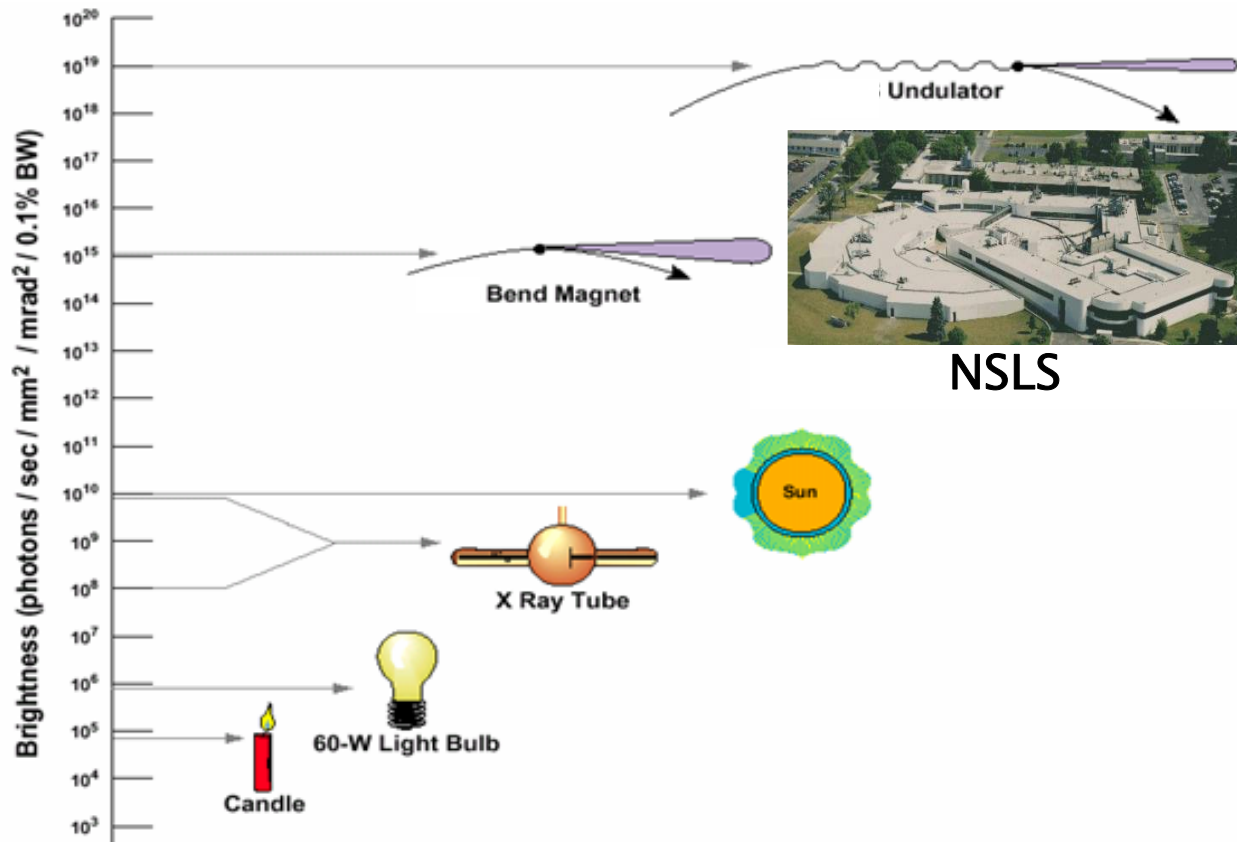
http://srs.dl.ac.uk/srworld/world_sr.html

Properties of Synchrotron Light

- **Broad band** – a wide energy range of photons is available (tunable)
- **Polarized** – this minimizes background scattering, improves sensitivity and enables measurement of circular dichroism
- **Pulsed** – the electron bunches produce nanosecond light pulses, enabling process kinetics to be followed and ‘movies’ of reactions to be made.

Properties of Synchrotron Light

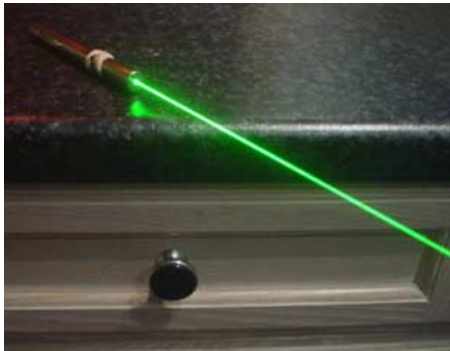
Brightness – many orders of magnitude brighter than conventional sources, enabling quick experiments on small samples.



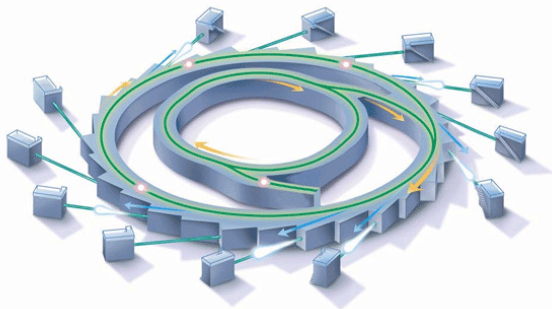
The **BRIGHTNESS** of Synchrotron Light



$$100 \text{ Watts} / 1000000 \text{ mm}^2 (1 \text{ m}^2) = \\ 0.0001 \text{ Watts} / \text{mm}^2$$

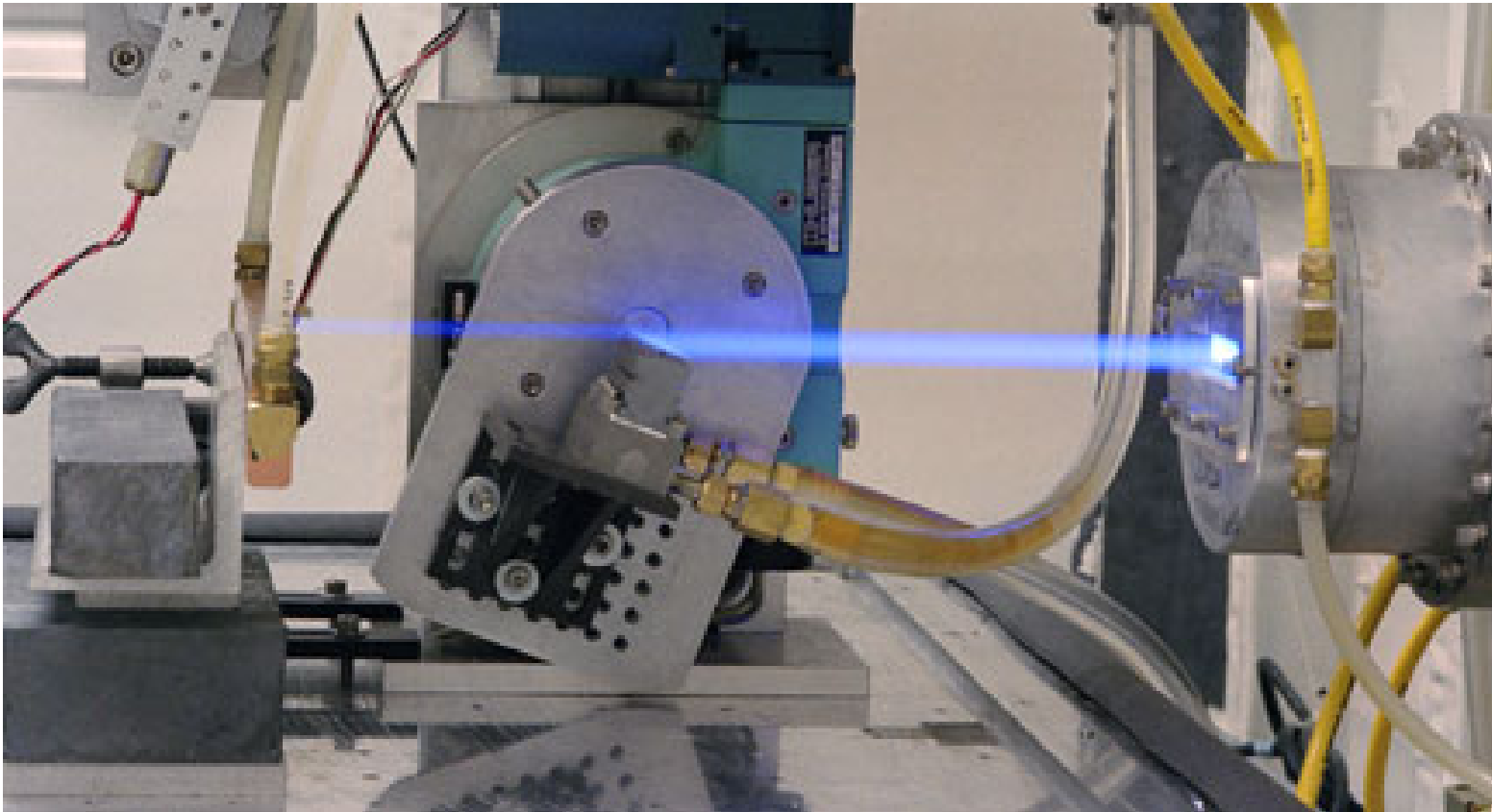


$$0.005 \text{ Watts} / 1 \text{ mm}^2 = \\ 0.005 \text{ Watts} / \text{mm}^2$$



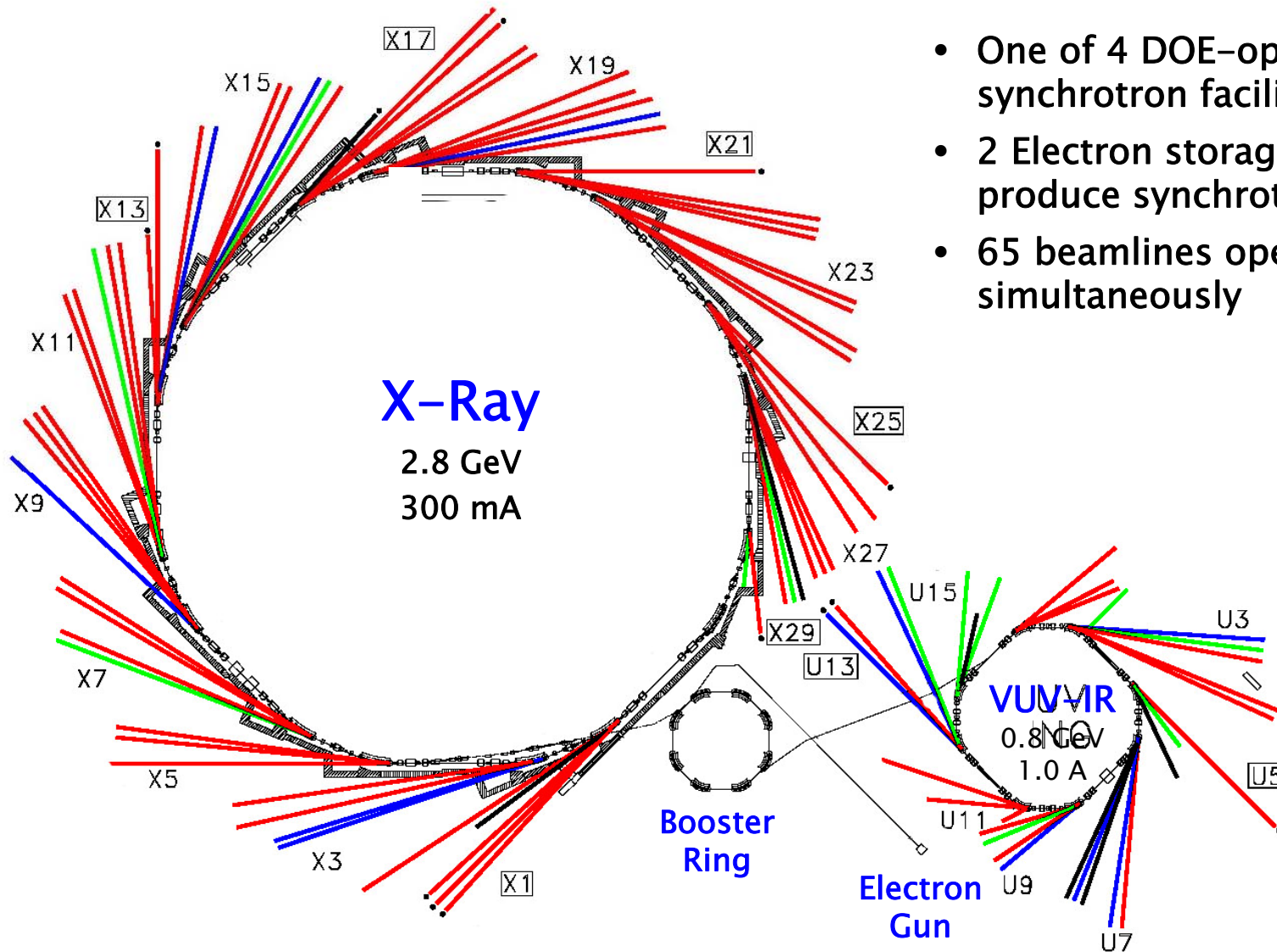
$$30 \text{ Watts} / 0.01 \text{ mm}^2 = \\ 3000 \text{ Watts} / \text{mm}^2$$

This is Synchrotron Light



(actually it's air (nitrogen) that has been excited by x-ray light)

How Do We Make Synchrotron Light?



- One of 4 DOE-operated synchrotron facilities
- 2 Electron storage rings that produce synchrotron light
- 65 beamlines operate simultaneously

Building a Synchrotron 101

1) Take evacuated beam pipe

ADD:

2) Bends (dipoles) to form e-beam trajectory (& as SR sources)

3) Quadrupole magnets to focus e-beam transversely

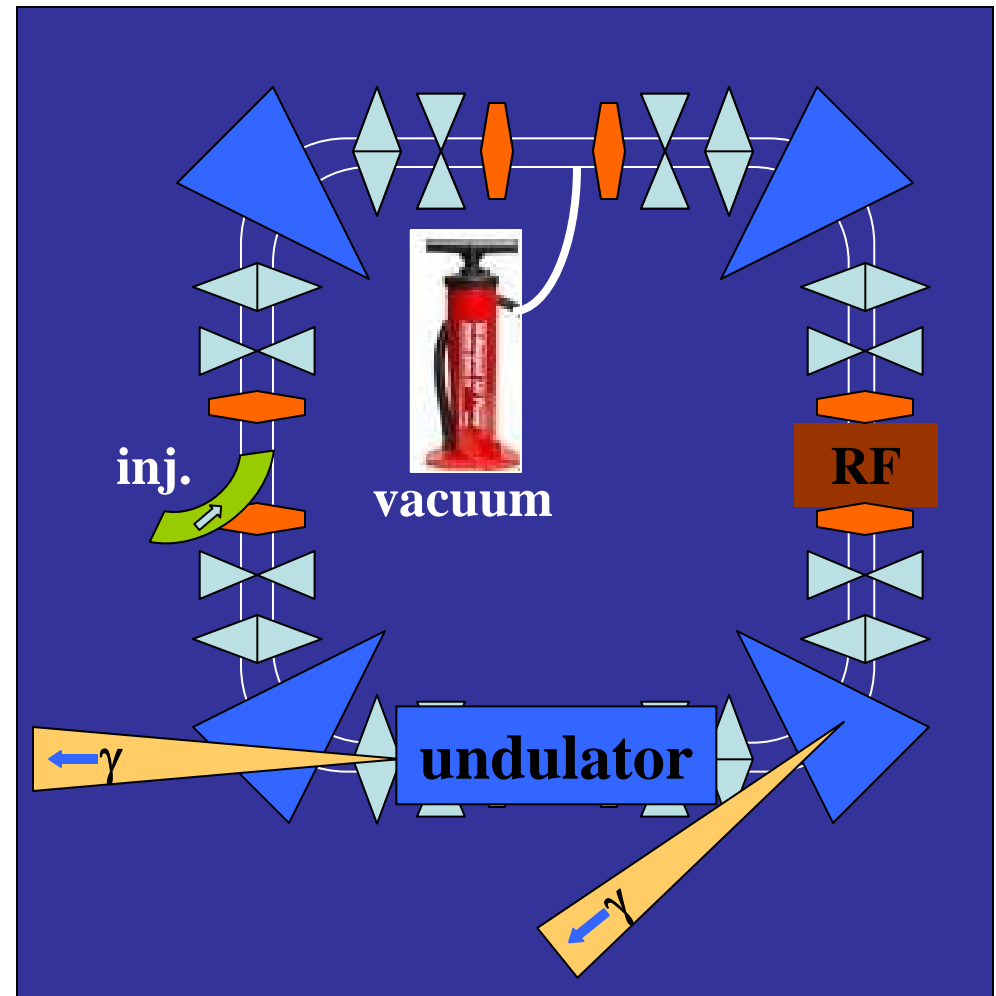
4) Sextupoles for achromatic focusing

5) RF to make up for energy loss; also provides longitudinal focusing (bunching)

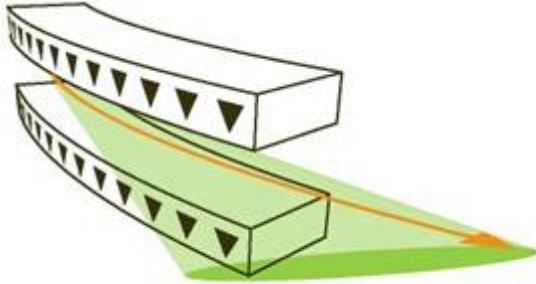
6) Injection system

7) IDs into avail. straight sections

8) Beamlines to deliver photons to the Users



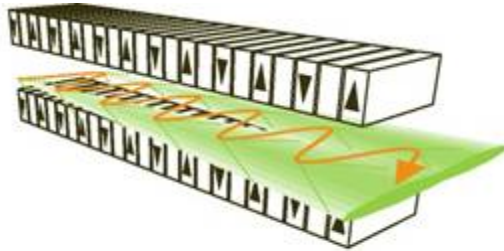
Types of Light-Generating Sources



Bending magnet

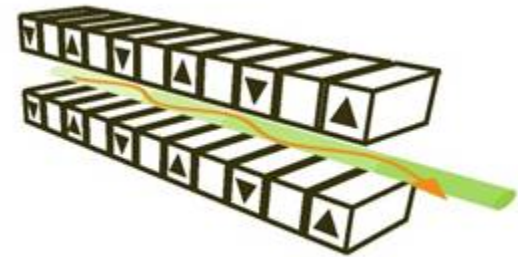
- Sweeping searchlight
- At each deflection of the electron path a beam of radiation is produced

Insertion devices: inserted into “straight sections” of the ring and produce higher intensity of light



Wiggler

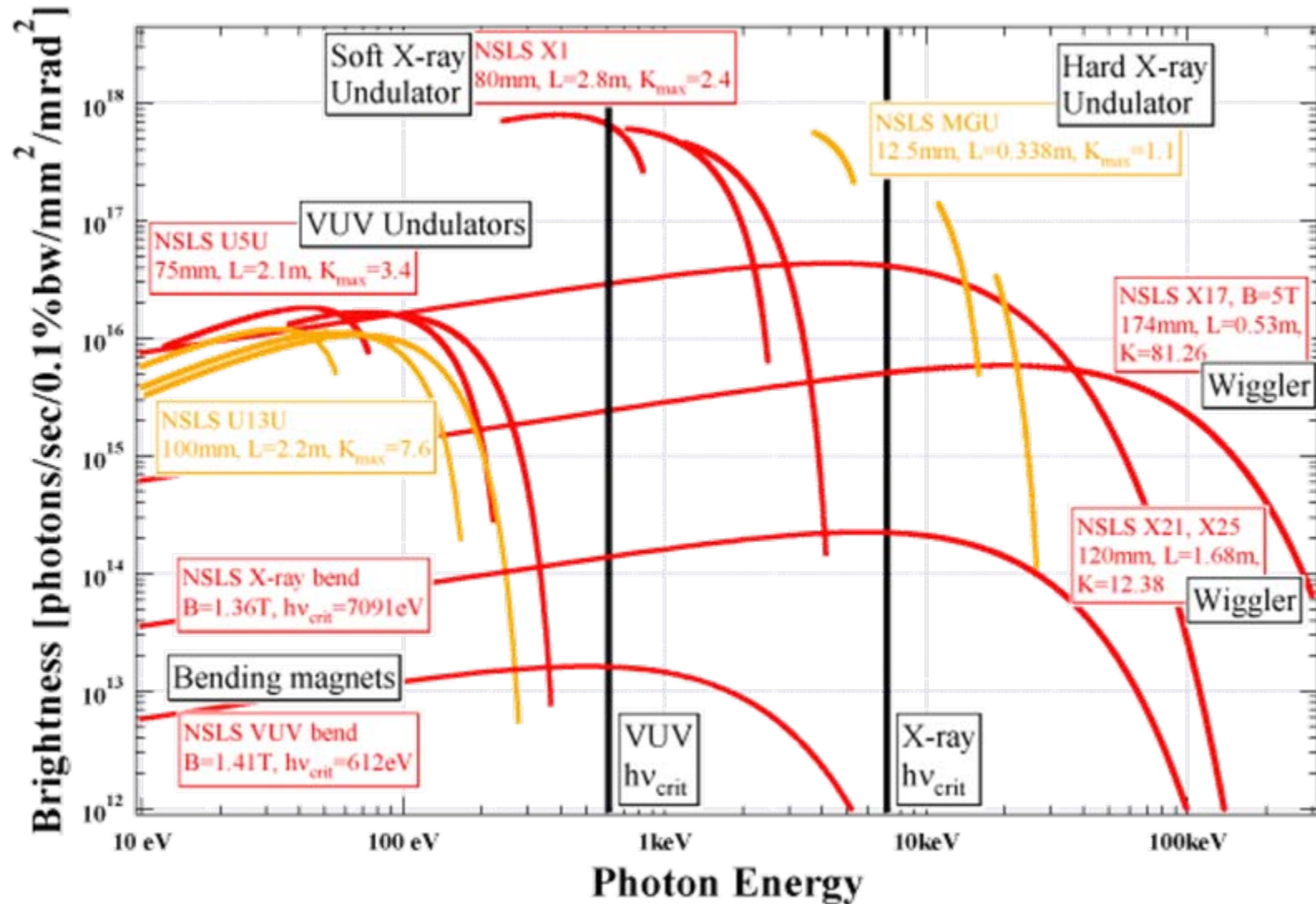
- Beams emitted at each pole reinforce each other and appear as a broad beam of incoherent light



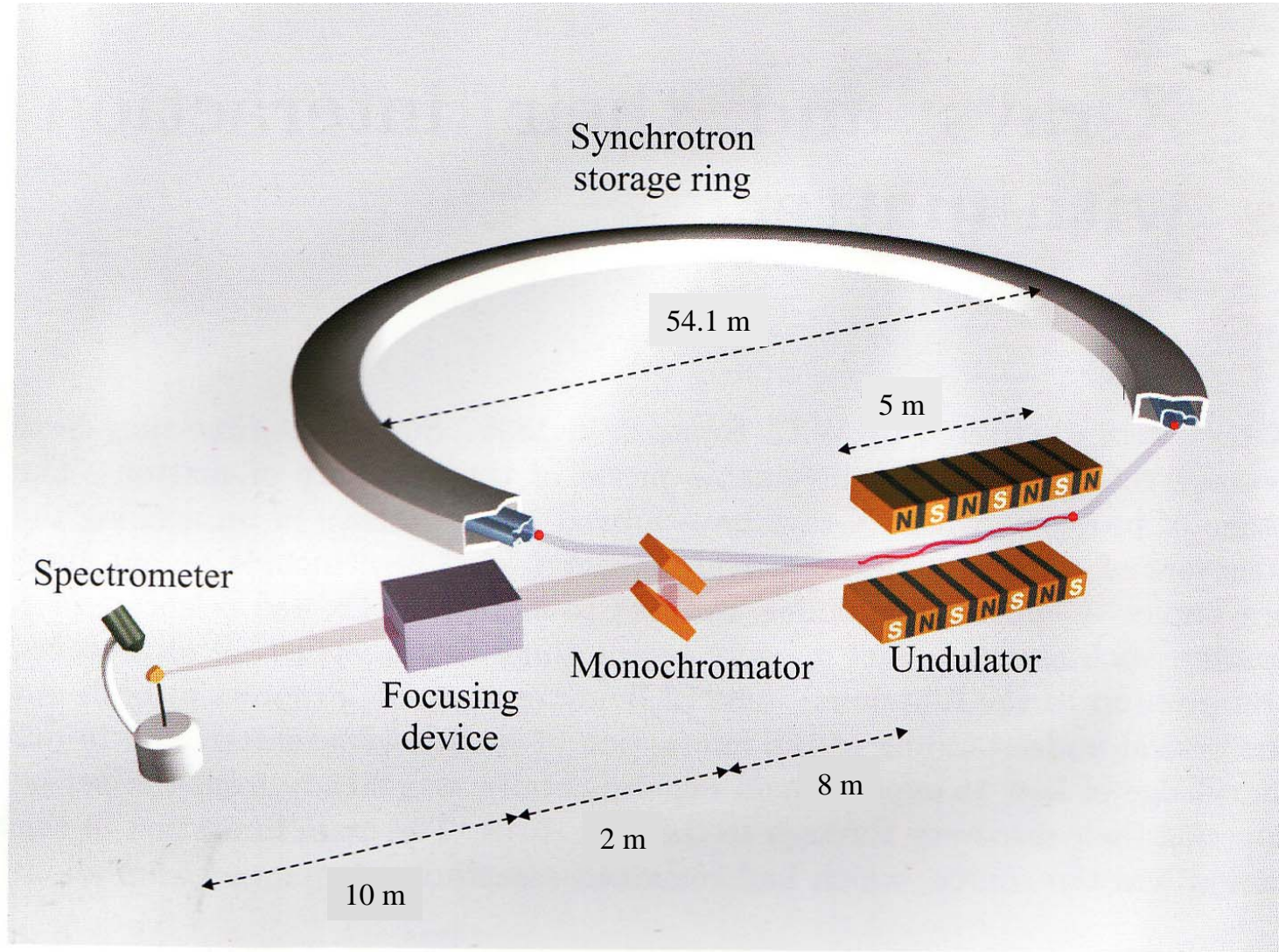
Undulator

- Produces a very narrow beam of coherent light, amplified by up to 10,000x

Source output



Typical Beamline



Outline

- The Facility
- The Users
- The Science

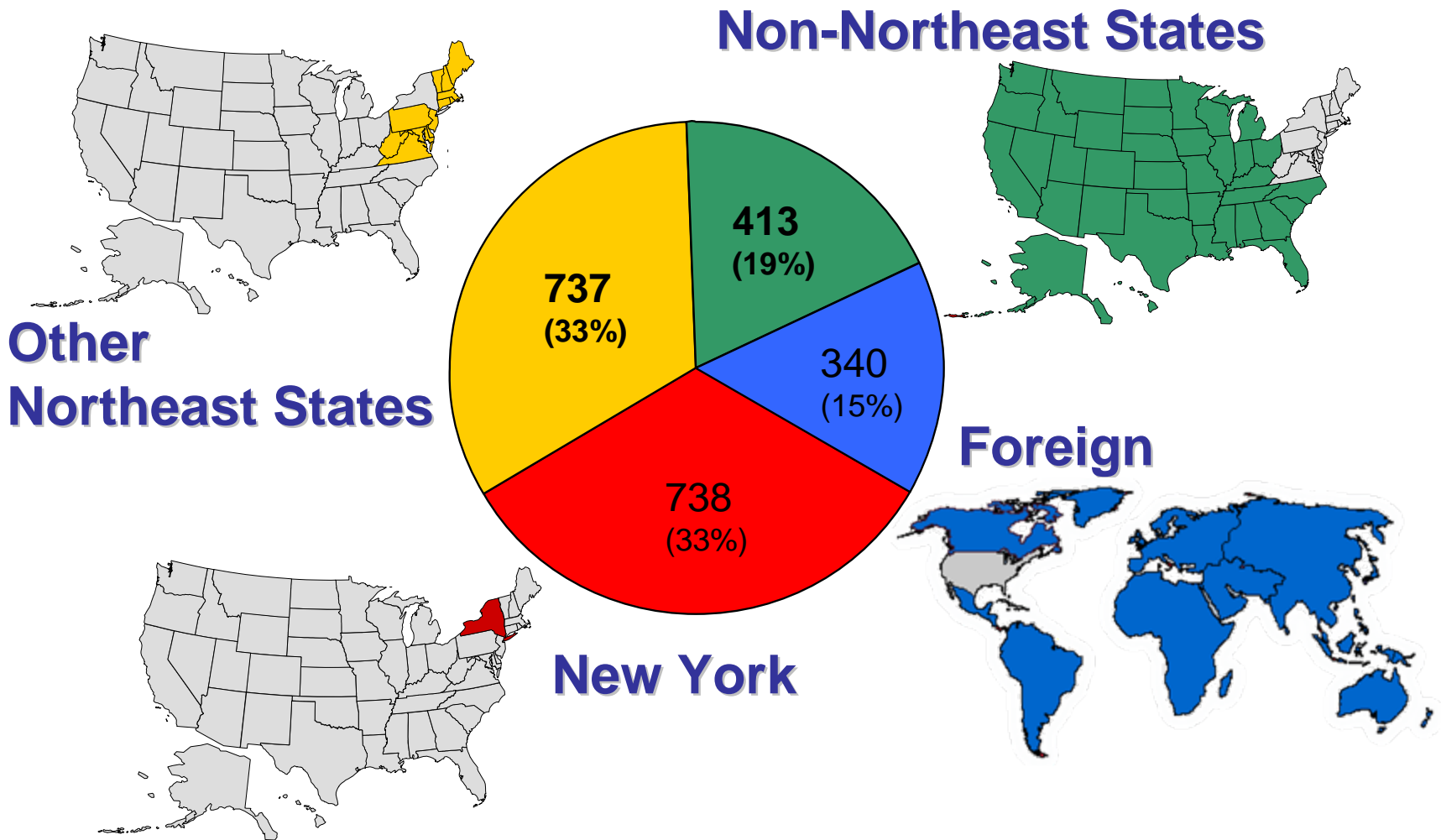


- Facility operates 24 hours/day, 7 days/week, ~10 months/year
- >2200 users per year (~1/3 are new users)
- Typical stay is 2–4 days (onsite housing)

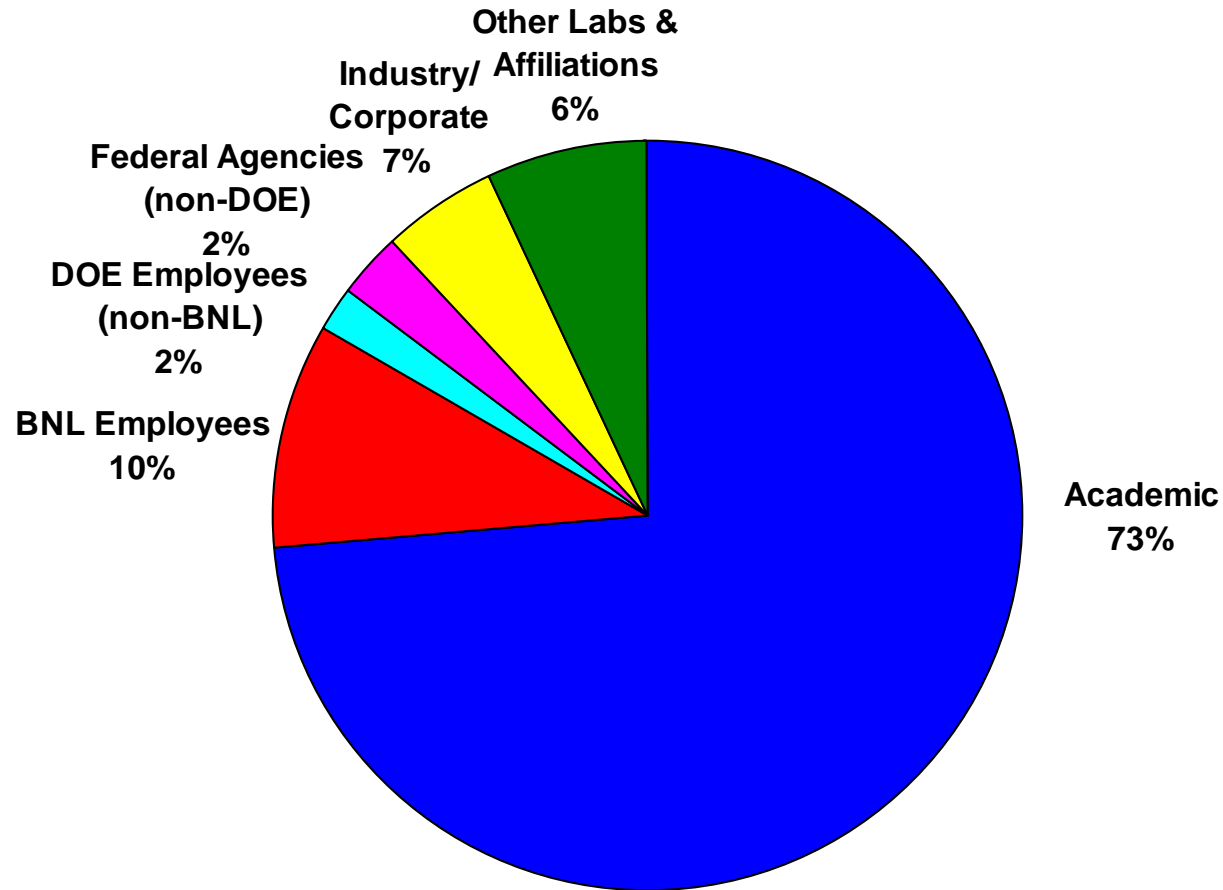
How to Get Beamtime

- Scheduled in three 4-month cycles
- User access via 2 primary mechanisms:
 - General User Proposals: peer-reviewed proposal system
 - Participating Research Teams (75%)
- Some proprietary research is done (full cost-recovery rate, ~\$100/hr)

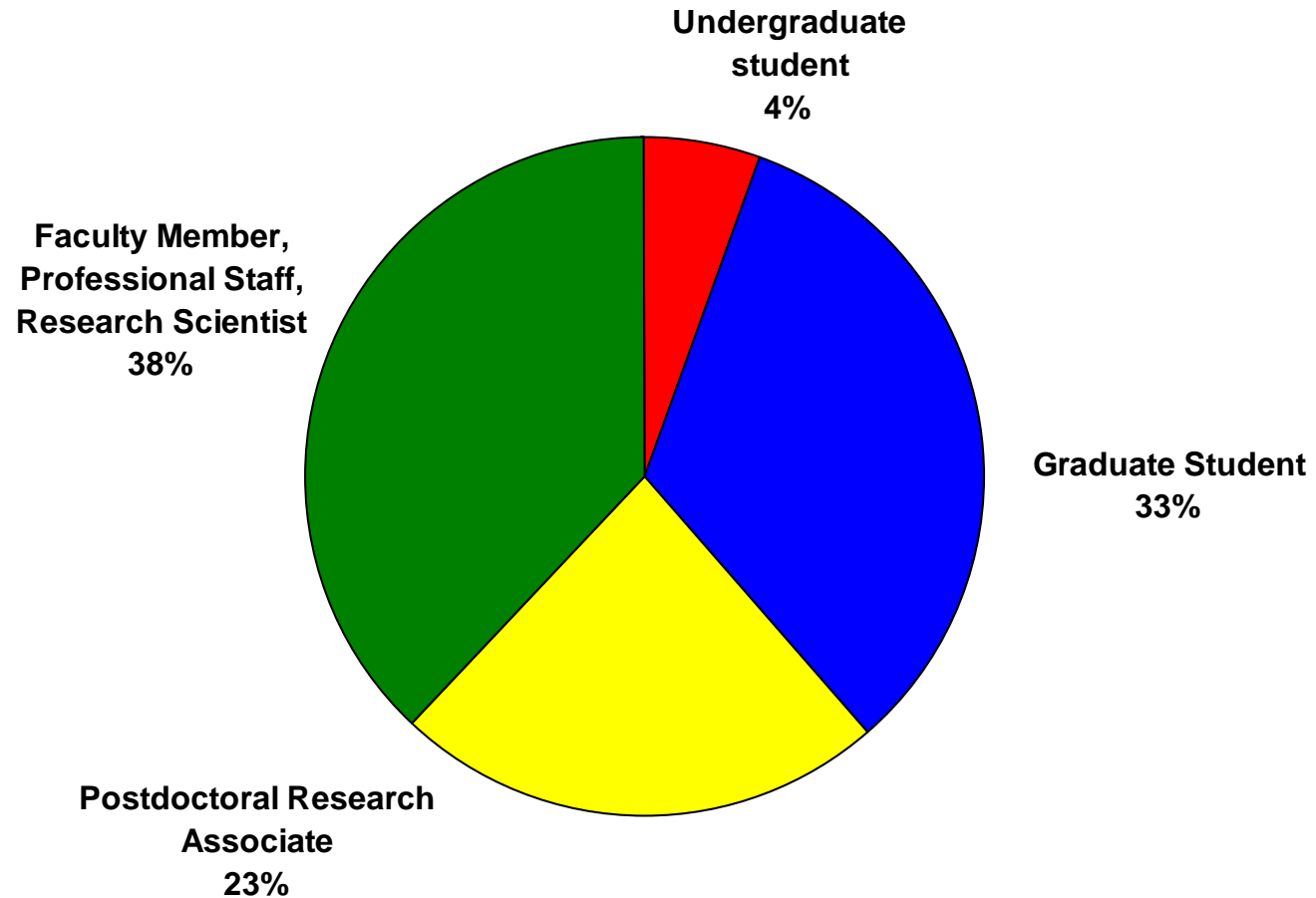
NSLS User Distribution FY 2010



NSLS Users by Research Institution FY 2010

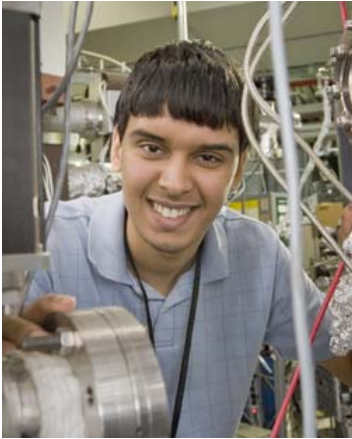


NSLS Users by Employment Level FY 2010



- A large number of students and postdocs work at NSLS

Student Research



Bishnu Panigrahi
college student

“heavy metal transport in plants”
Sayville, NY



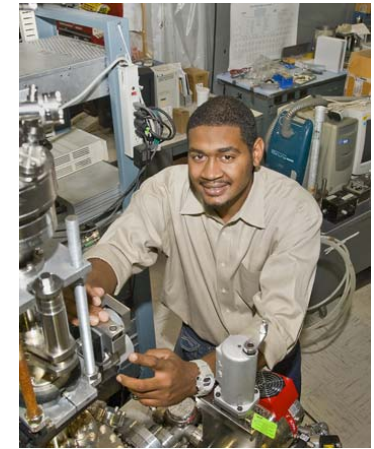
Ashley Jones
college student

“Arsenic toxicity in the kidney”
Saginaw, MI



Megan Bourassa
PhD student

“metal homeostasis in ALS”
Phoenix, AZ



Jeff Ambrose
college student

“nitrogen fixation in soils”
New Orleans, LA



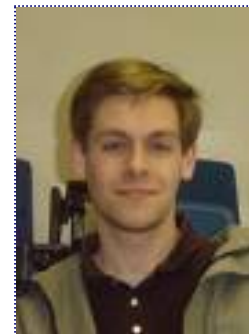
Matt Engel
PhD student

“protein structure in hepatitis C virus”
Stony Brook, NY



Andreana Leskovjan
PhD student

“metal uptake in Alzheimer’s disease”
Tampa, FL



Alvin Acerbo
PhD student

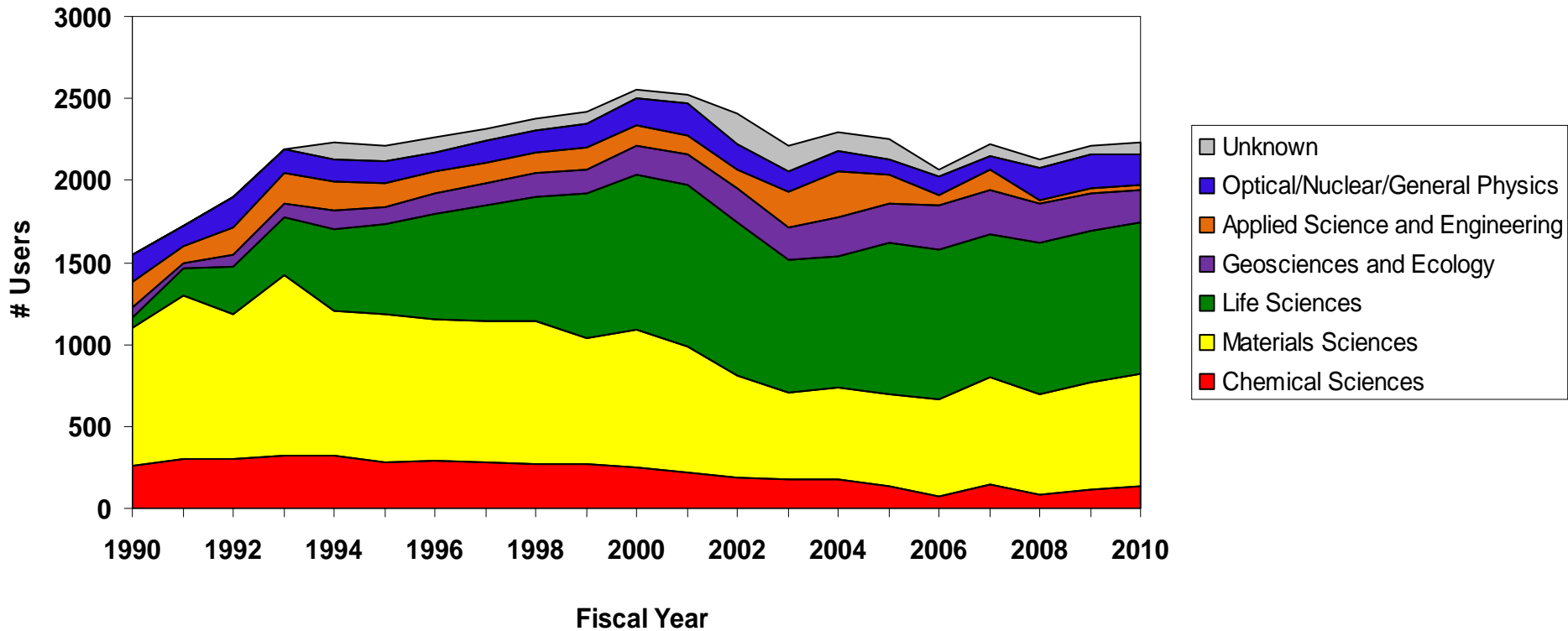
“improving imaging resolution”
The Netherlands



Shirin Mortazavi
college student

“unique methods to crystallize proteins”
Bellport, NY

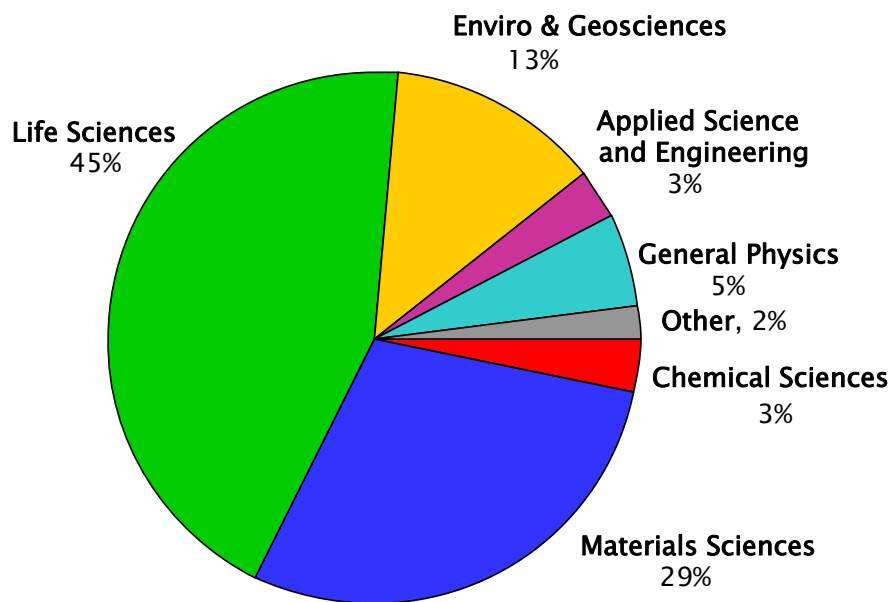
NSLS User Community



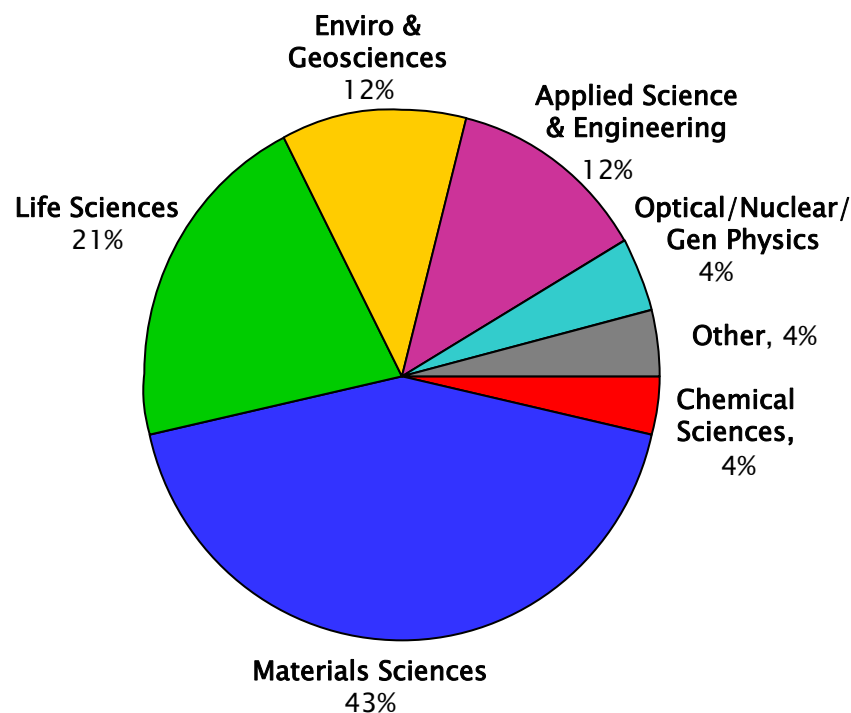
- 2228 scientists in FY 2010
- >400 academic, industrial, government institutions
- ~750 new users register each year
- Strongest growth in life sciences
- Largest groups are materials and life sciences

Beamtime Used by Field of Research

Number of Users



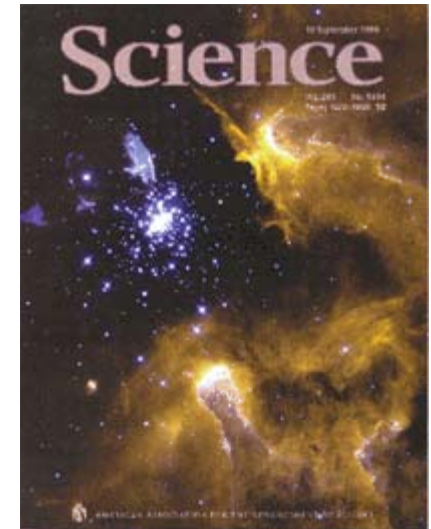
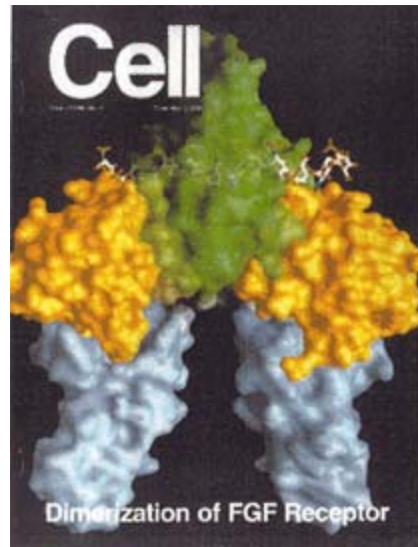
Beamtime Used*



- More life science users, shorter beamtime, high-throughput
- Materials science experiments are more complicated, time-consuming

NSLS Publications

- almost 1000 publications per year
- > 200 publications in premier journals

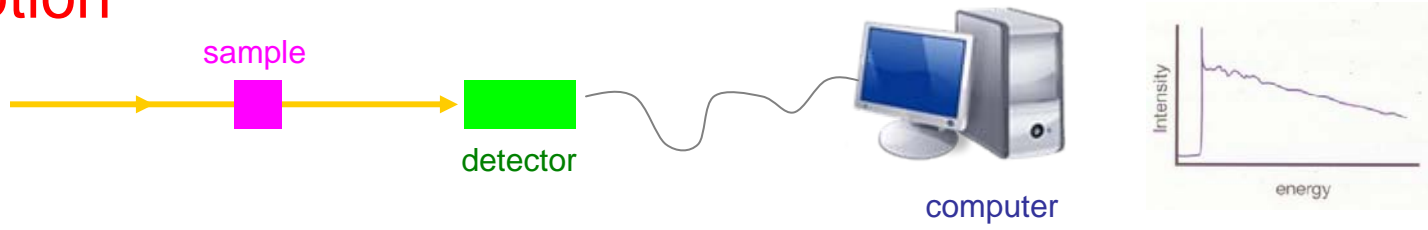


Outline

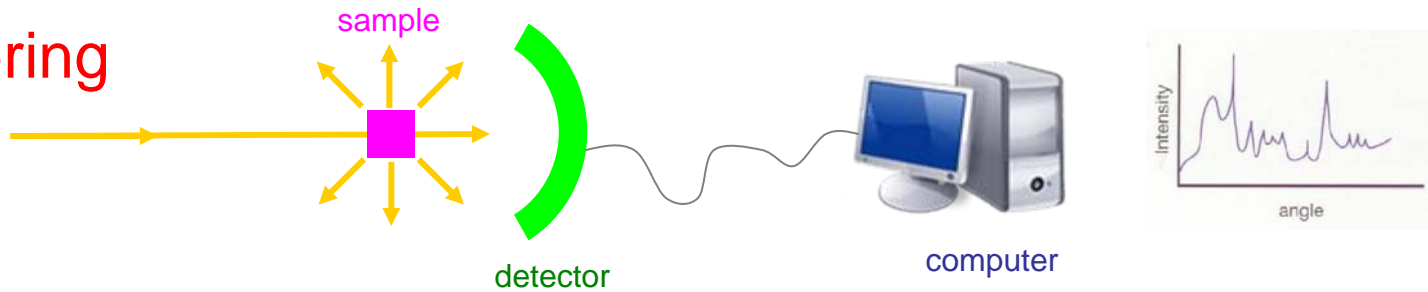
- The Facility
- The Users
- The Science

What do we do with Synchrotron Light?

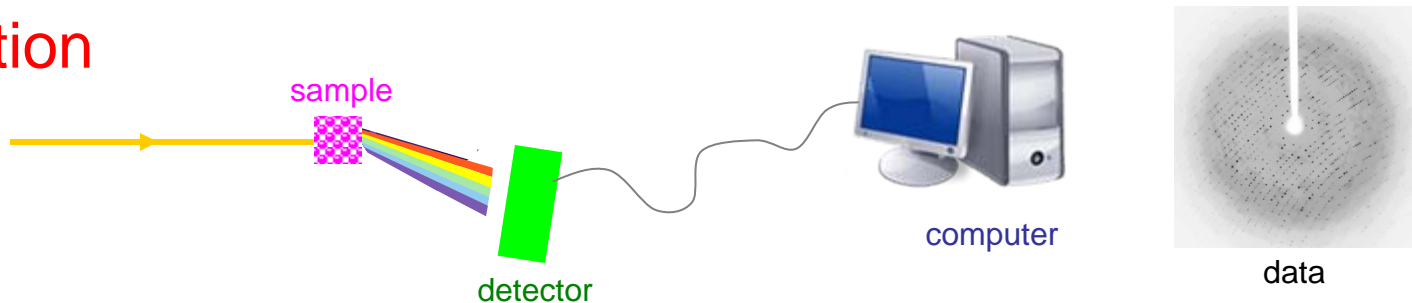
Absorption



Scattering



Diffraction



Major Synchrotron Techniques

SPECTROSCOPY

- Infrared spectroscopy
- Photoelectron spectroscopy
- X-ray absorption spectroscopy
- X-ray emission spectroscopy

DIFFRACTION/SCATTERING

- Protein crystallography
- Small molecule crystallography
- Powder diffraction
- Small-angle x-ray scattering
- X-ray microdiffraction
- High momentum resolution x-ray scattering

IMAGING

- Infrared microspectroscopy
- Soft X-ray scanning microscopy
- Hard X-ray microprobe
- X-ray microtomography
- Diffraction-enhance imaging

OTHER

- Micro-machining
- X-ray footprinting

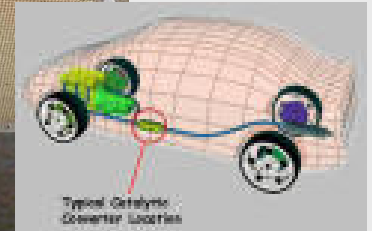
Chemistry



Corrosion

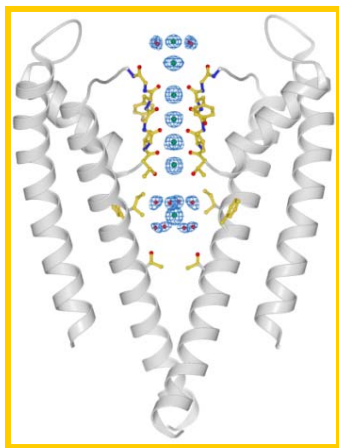


**Rechargeable
batteries**



Catalytic converters

2003 and 2009 Nobel Prizes in Chemistry



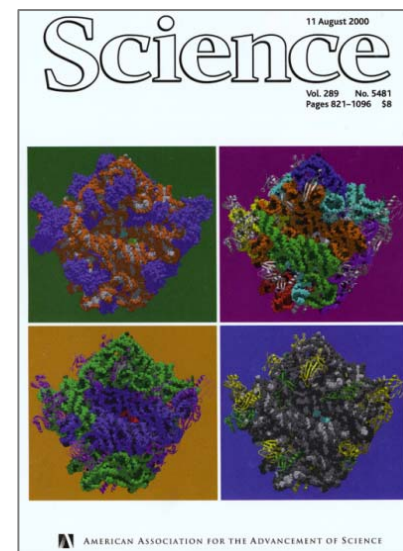
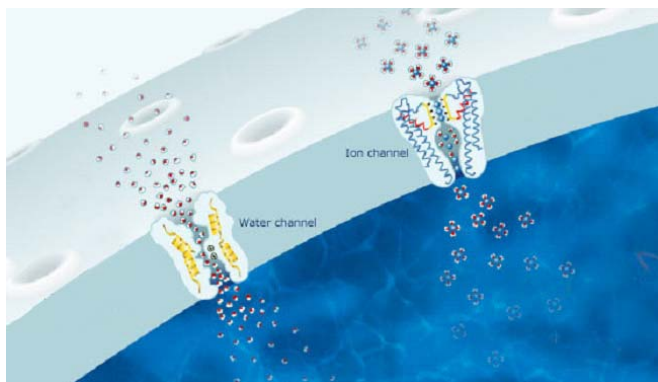
Rod MacKinnon



Venkatraman Ramakrishnan



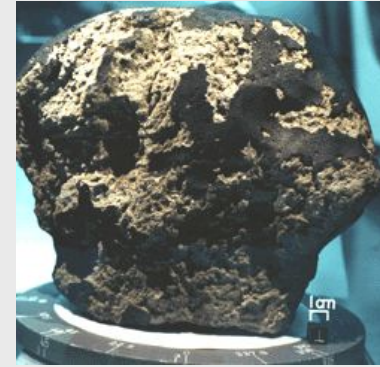
Thomas Steitz



Geology and Environmental Science



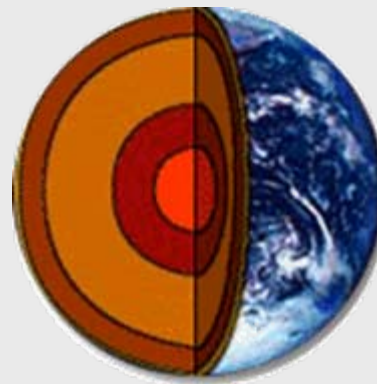
Environmental cleanup



Mars meteorites



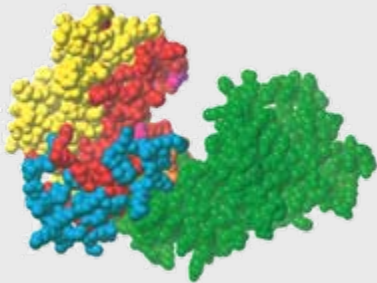
Space dust



Earth's core



Biology and Medicine



Anthrax



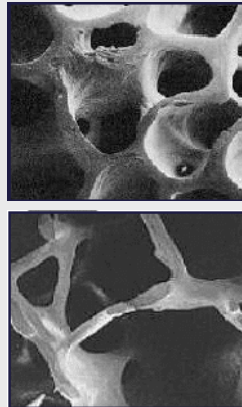
Malaria



Lyme's disease



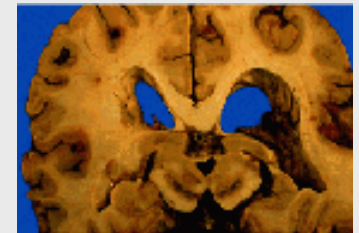
Arthritis



Osteoporosis



HIV



Alzheimer's disease

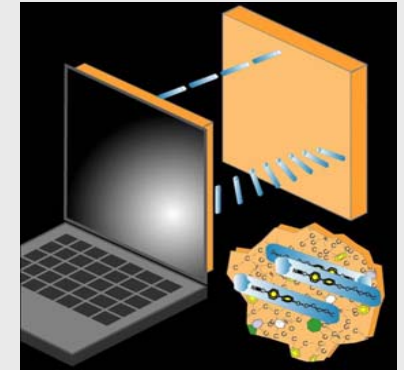
Physics and Materials Science



Data storage



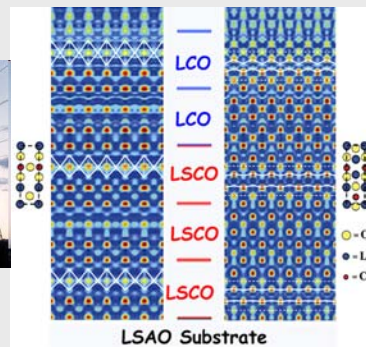
Improved polymers



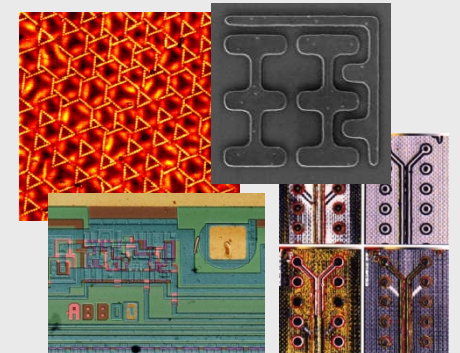
Liquid crystal displays



Nonstick coatings

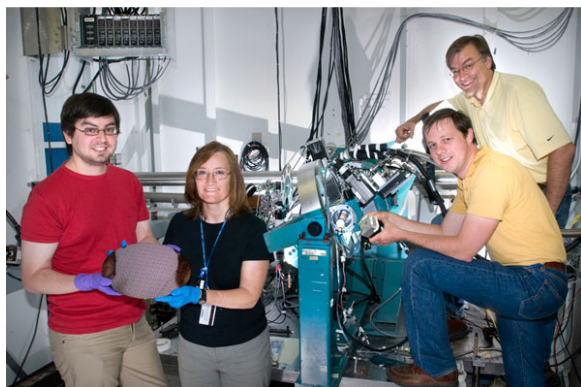
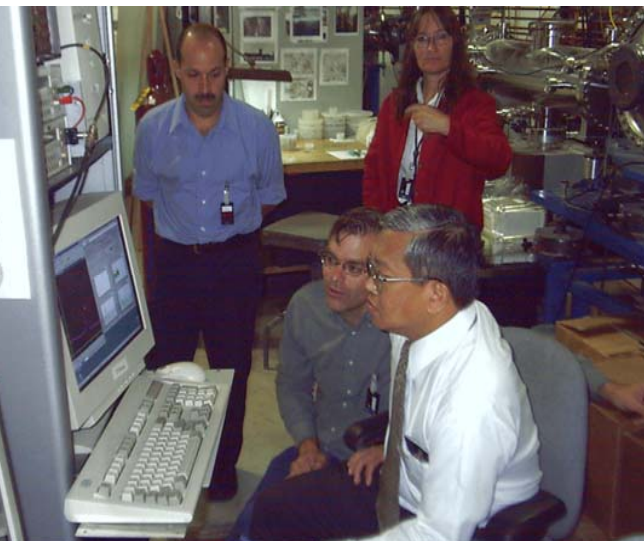
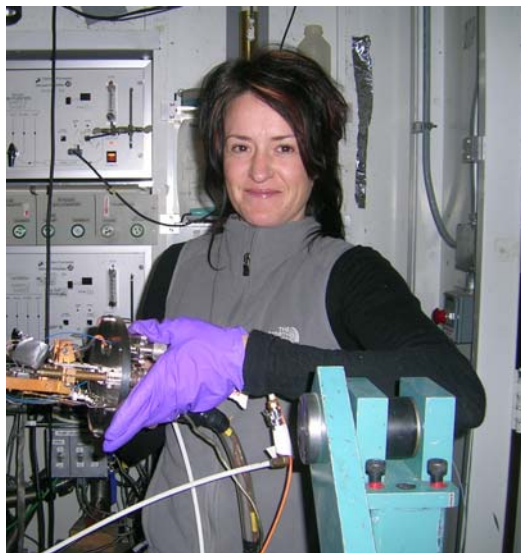


Exotic inorganics

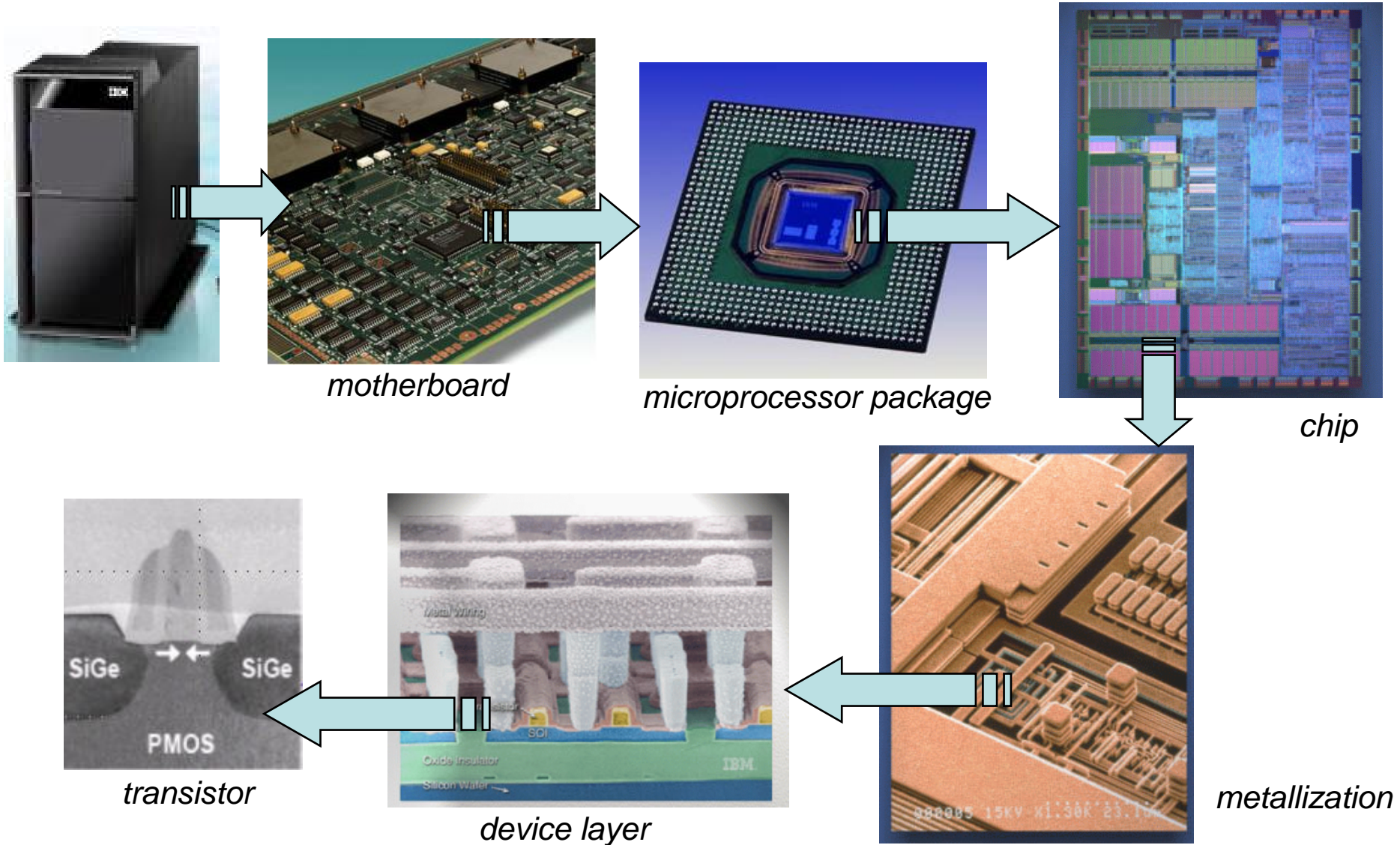


Nanomaterials

IBM and collaborators at NSLS

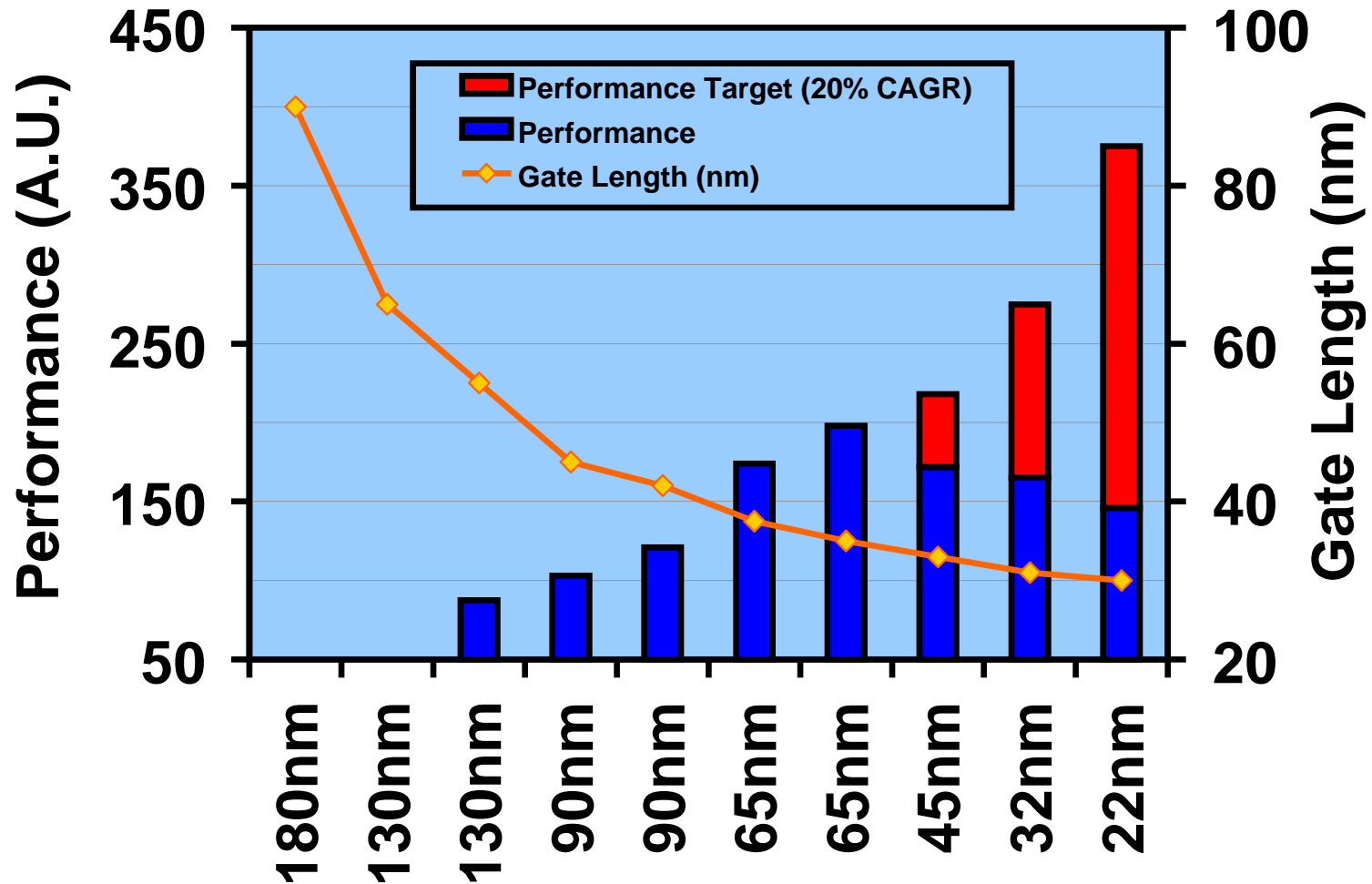


Computer hardware



Faster, Smaller, Cheaper

With every generation beyond 45nm, performance scaling is becoming increasingly difficult and will require new performance-enhancing elements



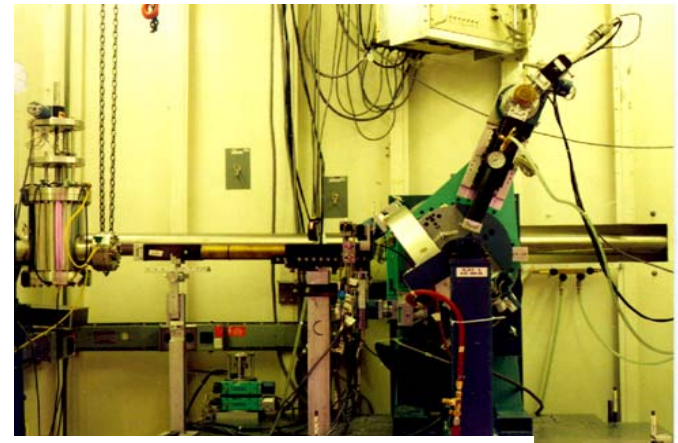
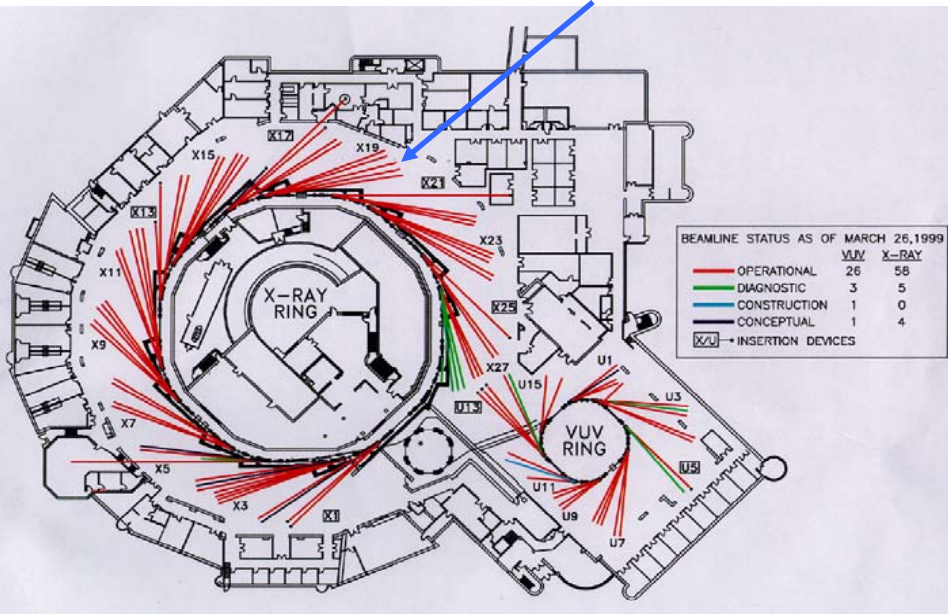
“We are not just shrinking lithography and buying the next-best stepper. **Everything in the last five years has included changes in basic materials and process engineering**, whether it’s strain engineering, embedded silicon germanium in our Sol, our Airgap technology, our high-k metal gate—these are fundamental R&D, not just shrinking.” *Tom Reeves VP of semiconductor and technology services at IBM*, interview in *Electronic News*, 6/19/2007

Before 90's																					
hydrogen 1 H																	helium 2 He				
Since the 90's																					
Beyond 2006																					
lithium 3 Li 6.941	beryllium 4 Be 9.0122															boron 5 B	carbon 6 C	nitrogen 7 N	oxygen 8 O	fluorine 9 F	neon 10 Ne 20.180
sodium 11 Na 22.990	magnesium 12 Mg															aluminum 13 Al	silicon 14 Si	phosphorus 15 P	sulfur 16 S 32.065	chlorine 17 Cl	argon 18 Ar 39.948
potassium 19 K 39.098	calcium 20 Ca	scandium 21 Sc	titanium 22 Ti	vanadium 23 V	chromium 24 Cr	manganese 25 Mn 54.938	iron 26 Fe 55.845	cobalt 27 Co	nickel 28 Ni	copper 29 Cu	zinc 30 Zn	gallium 31 Ga 69.723	germanium 32 Ge	arsenic 33 As	selenium 34 Se 78.96	bromine 35 Br	krypton 36 Kr 83.80				
rubidium 37 Rb 85.468	strontium 38 Sr	yttrium 39 Y	zirconium 40 Zr	niobium 41 Nb	molybdenum 42 Mo	technetium 43 Tc [98]	ruthenium 44 Ru	rhodium 45 Rh	palladium 46 Pd	silver 47 Ag 107.87	cadmium 48 Cd 112.41	indium 49 In 114.82	tin 50 Sn 118.71	antimony 51 Sb	tellurium 52 Te	iodine 53 I 126.90	xenon 54 Xe [222]				
cesium 55 Cs 132.91	barium 56 Ba	57-70 *	lutetium 71 Lu 174.97	hafnium 72 Hf	tantalum 73 Ta	tungsten 74 W	rhenium 75 Re 186.21	osmium 76 Os 190.23	iridium 77 Ir	platinum 78 Pt 195.08	gold 79 Au 196.97	mercury 80 Hg 200.59	thallium 81 Tl 204.38	lead 82 Pb 207.2	bismuth 83 Bi [209]	polonium 84 Po [209]	astatine 85 At [210]	radon 86 Rn [222]			
francium 87 Fr [223]	radium 88 Ra [226]	89-102 **	lawrencium 103 Lr [262]	rutherfordium 104 Rf [261]	dubnium 105 Db [262]	seaborgium 106 Sg [266]	bohrium 107 Bh [264]	hassium 108 Hs [269]	meitnerium 109 Mt [268]	ununnium 110 Uun [271]	ununium 111 Uuu [272]	ununbium 112 Uub [277]	ununquadium 114 Uuq [289]								
		*lanthanoids	lanthanum 57 La	cerium 58 Ce	Praeseodymium 59 Pr	neodymium 60 Nd	promethium 61 Pm [145]	samarium 62 Sm	europium 63 Eu	gadolinium 64 Gd	terbium 65 Tb	dysprosium 66 Dy	holmium 67 Ho	erbium 68 Er	thulium 69 Tm	ytterbium 70 Yb					
		**actinoids	actinium 89 Ac [227]	thorium 90 Th 232.04	protactinium 91 Pa 231.04	uranium 92 U 238.03	neptunium 93 Np [237]	plutonium 94 Pu [244]	americium 95 Am [243]	curium 96 Cm [247]	berkelium 97 Bk [247]	californium 98 Cf [251]	einsteinium 99 Es [252]	fermium 100 Fm [257]	mendelevium 101 Md [258]	nobelium 102 No [259]					

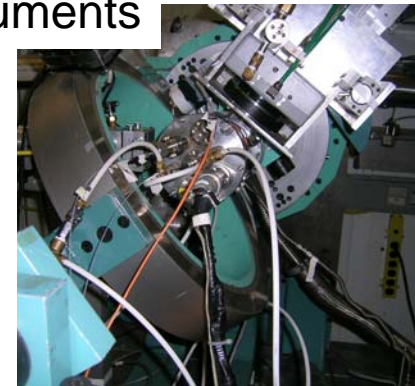
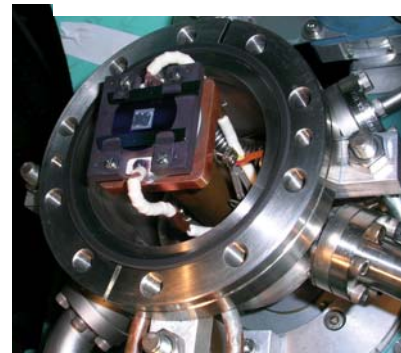
Why IBM comes to the NSLS:

IBM scientists travel to the NSLS to use x-rays to look at the arrangement of atoms in crystalline materials. The x-rays can tell us the type of crystal, if it is under stress, or if it is unstable during heating or reacts with other materials. We can also use x-rays to measure the thickness of films, how rough they are, and their “texture”.

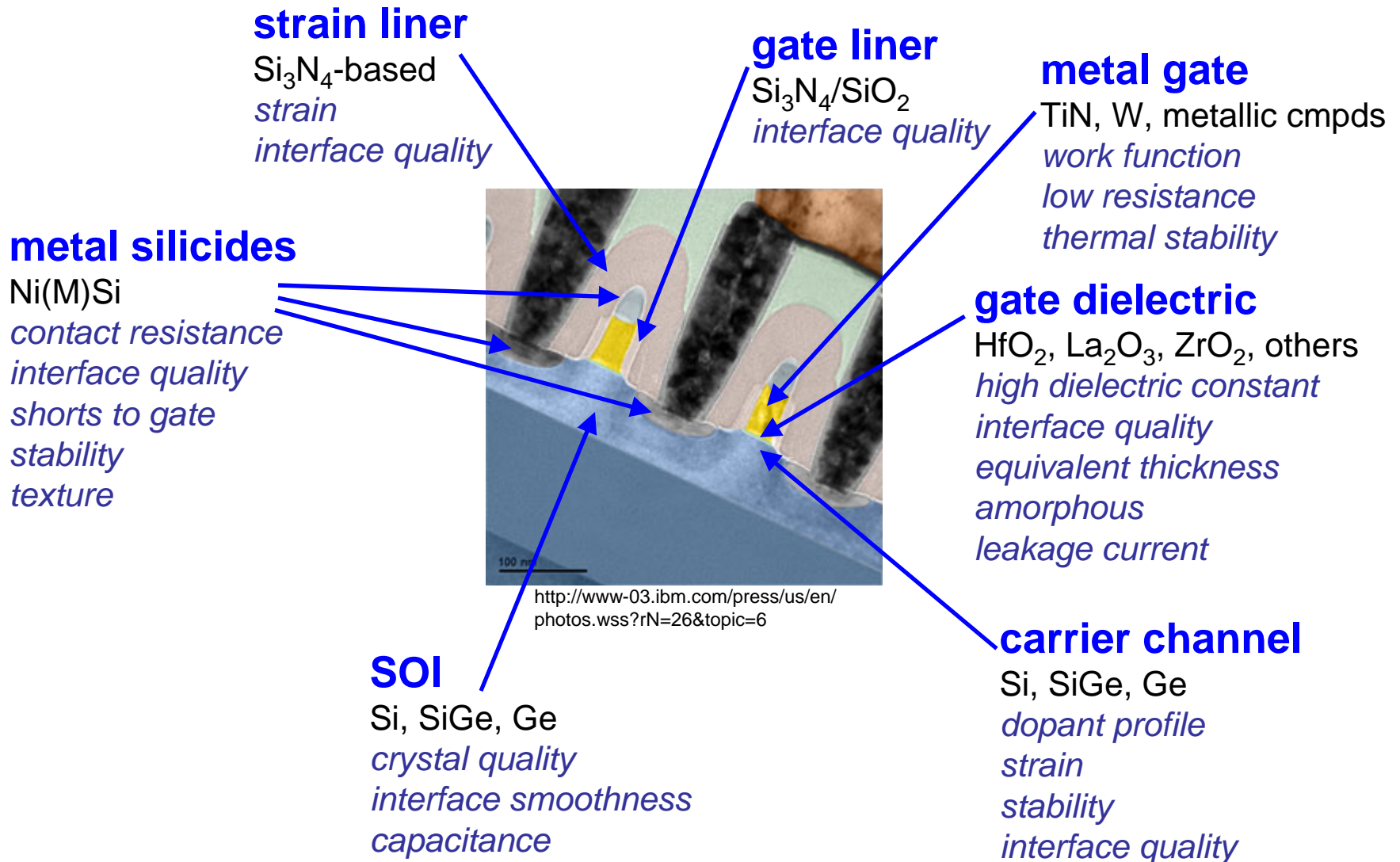
IBM's X20 beamlines



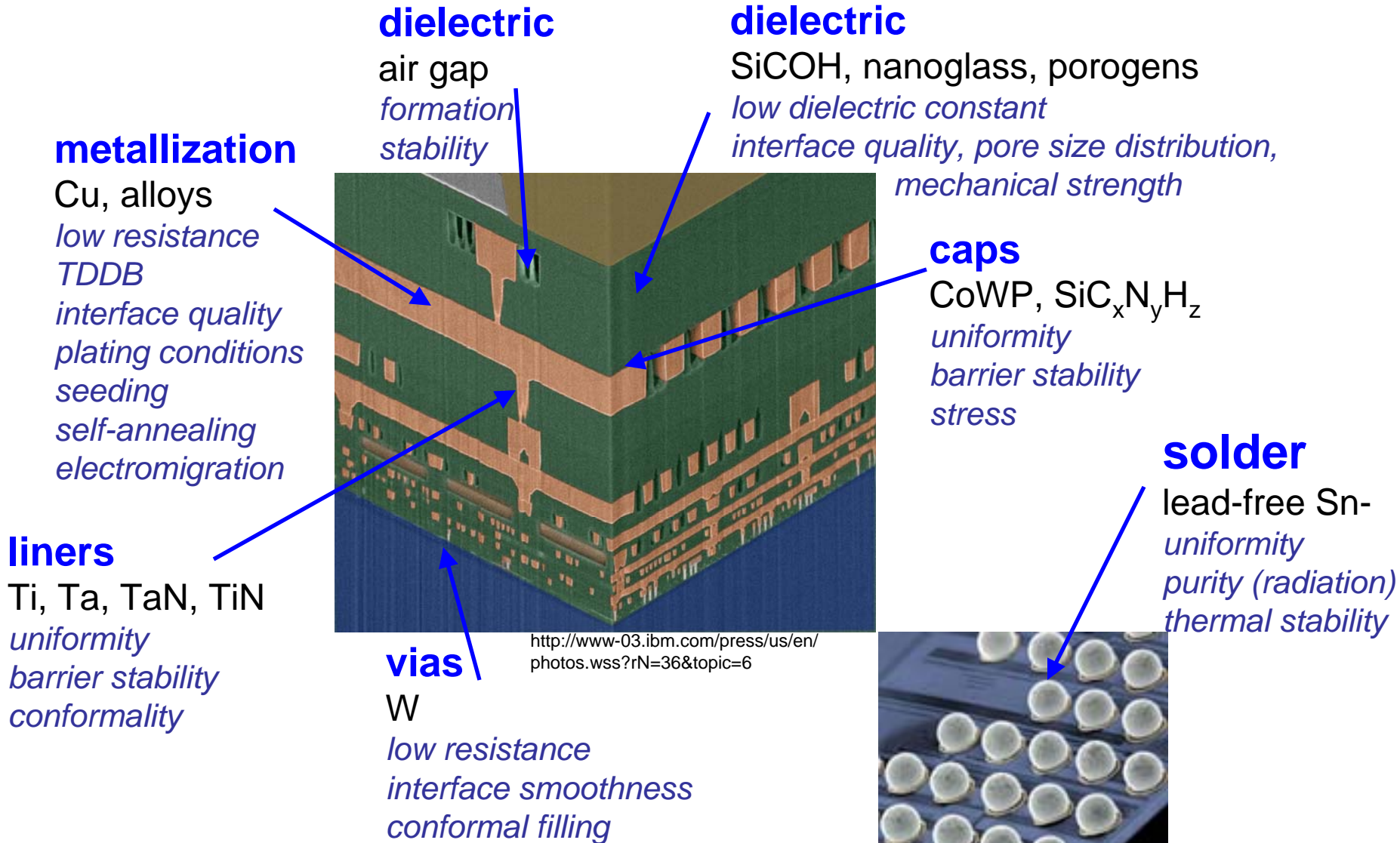
beamline instruments



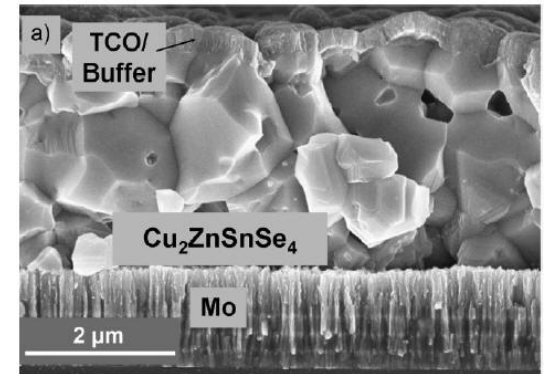
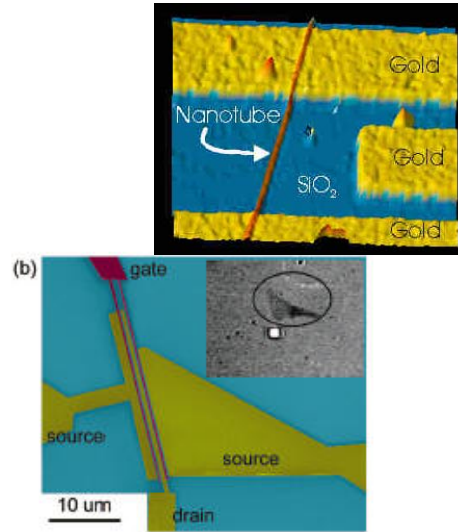
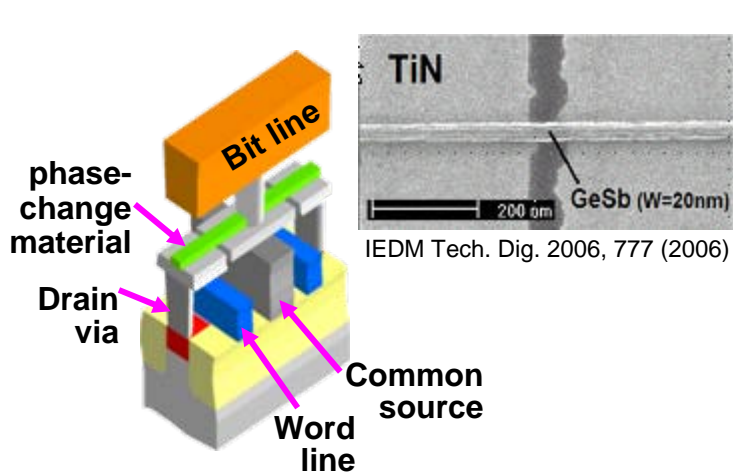
FEOL materials and challenges



BEOL / MOL materials and challenges



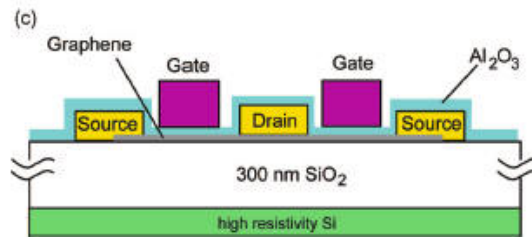
New devices, exotic materials



Phase-change memory

GeSb, GeSbTe, GeTe, others

crystallization Temp
amorphization Temp
resistance change
transition speed
scalability, cyclability



C-based transistors

graphene, CNT
switching speed
carrier mobility
manufacturability

solar cells

CIGS, CZSSS, Si
ohmic contact
efficiency
low cost
stability
manufacturability

The End