

Activated graphene: A new material for electrical energy storage

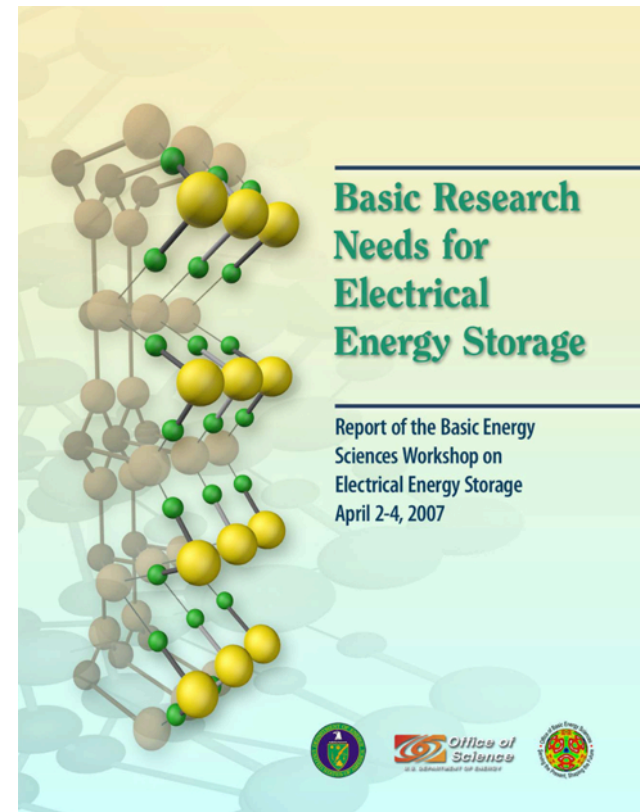
Eric A. Stach

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Brookhaven National Laboratory

Why do we need new materials for energy storage?

- **Electrical energy Storage refers to the ability to store electrical energy in a form that is readily available for later usage**
- **It is one of the most critical energy challenges**
 - **Consumer electronics**
 - Efficient, safe, cheaper is nice ...
 - **Transportation**
 - Lightweight, highly-efficient, safe
 - **Li-ion at present, what next?**
 - **Energy transmission – electric grid**
 - Cheap, reliable, Did I mention cheap?



"The performance of current EES technologies falls well short of requirements for using electrical energy efficiently in transportation, commercial, and residential applications."

Why do we need new materials for energy storage?

ELECTRONICS

Apple recalls first-gen iPod nano for bad batteries

Nov 14, 2011 10:15 AM

On Friday, Apple issued a recall for its first-generation iPod nano. The iconic digital music player has built-in rechargeable batteries that may overheat and pose a "safety risk."

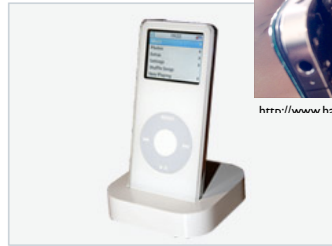
According to Apple, the problem affects only certain iPod nanos that contain defective batteries from one specific supplier. The faulty nanos were sold between September 2005 and December 2006. Says Apple:

“

While the possibility of an incident is rare, the likelihood increases as the battery ages.

”

<http://news.consumerreports.org/safety/2011/11/apple-recalls-first-gen-ipod-nano-for-bad-batteries.html>



<http://www.hanksville.com>

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Fiery Tesla crash torpedoes stock price



<http://www.telegraph.co.uk/news/1526424/Exploding-laptops->

17 January 2013 Last updated at 09:4



Li-ion batteries in personal electronics and transportation

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GM Lithium-Battery Explosion Sparks Fire at Company Test Lab

By Tim Higgins - Apr 11, 2012 4:06 PM ET

f t in +1 13 COMMENTS

+ QUEUE

General Motors Co. (GM), maker of the Chevrolet Volt plug-in hybrid sedan, said a test battery exploded at a research facility near its Detroit headquarters.

<http://www.bloomberg.com/news/2012-04-11/gm-lithium-battery-lab-explosion-injures-2-fire->

Dreamliner: Boeing 787 planes grounded on safety fears



There are 50 Boeing 787 Dreamliners at airlines around the world

All of Boeing's 50 flagship 787 Dreamliners have been temporarily taken out of service amid safety concerns.

Related Stories

Why do we need new materials for energy storage?

Integration of renewable energy into the 'smart grid'

➤ Intermittency



 SMARTGRID.GOV

Why do we need new materials for energy storage?

Weather leads to between \$18B and \$33B / year in losses*

- **Hurricane Sandy: between \$27B & \$55B**
- **90% of Suffolk County without power**
- **50 deaths attributed to power outages**

A 'smart grid' increases the resiliency of the grid

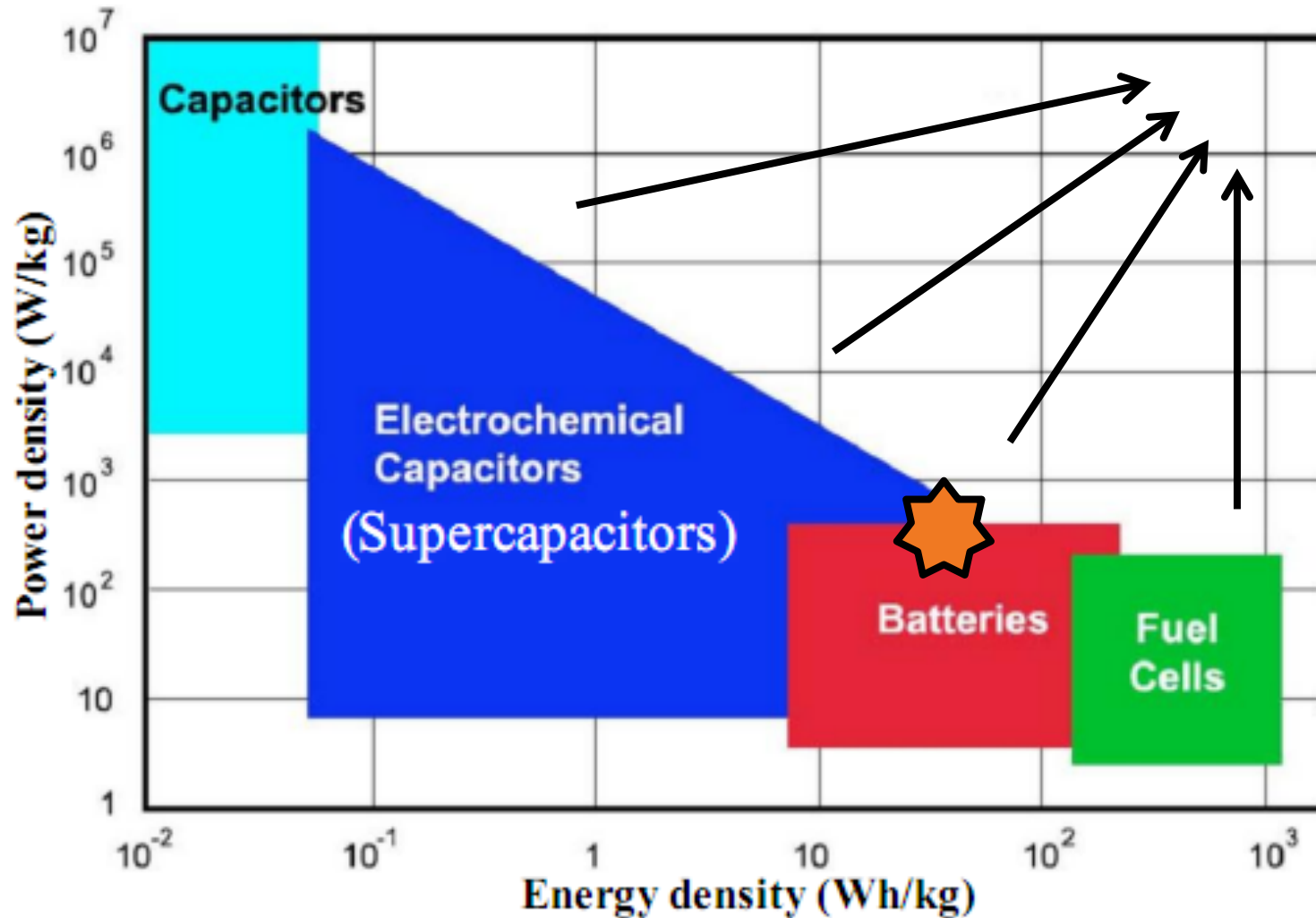


Breezy Point, Queens after Hurricane Sandy

* ECONOMIC BENEFITS OF INCREASING ELECTRIC GRID RESILIENCE TO WEATHER OUTAGES, Executive Office of the President, August 2013

http://energy.gov/sites/prod/files/2013/08/f2/Grid%20Resiliency%20Report_FINAL.pdf

Different available technologies



Back story – Part 1

- Rod Ruoff @ University of Texas at Austin
- Had synthesized a new material with “extraordinary properties” as a “supercapacitor”
- But what exactly was it?



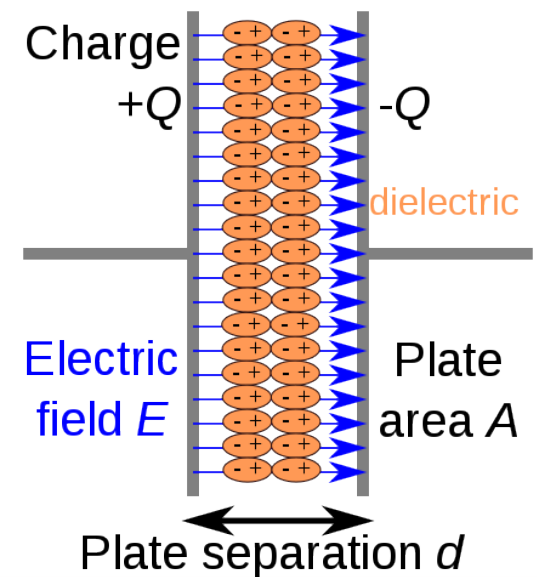
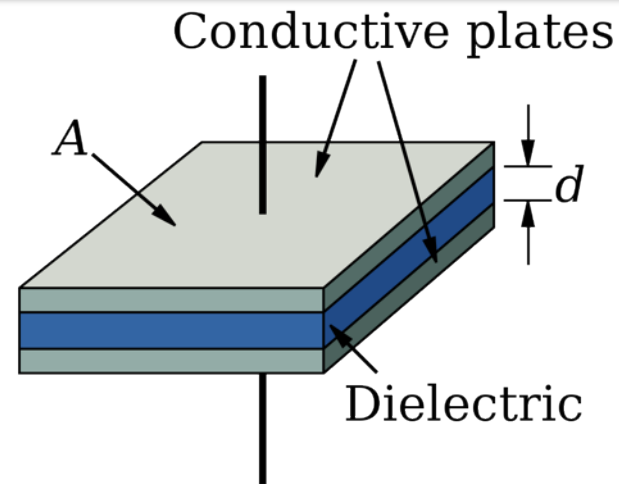
THE WHO'S WHO IN SCIENCE

Honoring the **hottest** researchers and **hottest** research papers of 2012.

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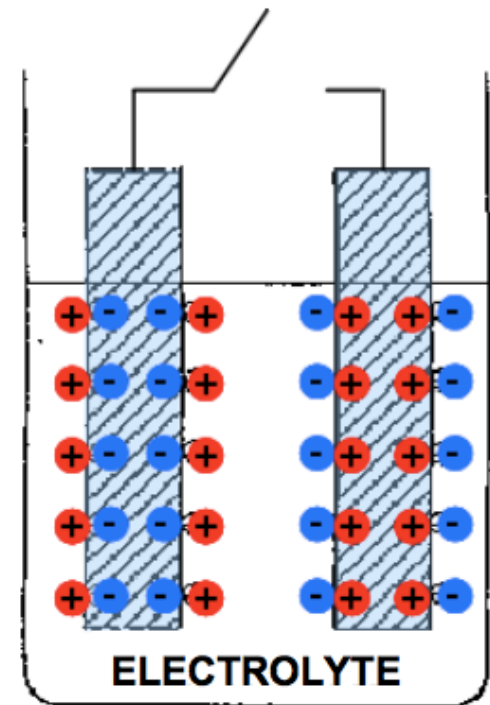
Capacitors

- Capacitors store electrical charge in a dielectric media
- This is conventionally a ceramic material
 - **Ceramics do have electrons that move freely in the material**
 - **Thus they do not conduct electric charge**
- Good dielectrics store charge by having electrons that can shift (“polarize”) in an applied field



Electrochemical Capacitors: “Supercapacitors”

- Batteries transfer charged ions from one electrode to a second electrode by an fluid (electrolyte).
 - **A chemical reaction occurs at the interface to store the charge**
- Supercapacitors store charge physically at interfaces in the material
 - **Application of a voltage separates charge in the electrolyte to the interfaces**
 - **When voltage is removed the charge is stored**
- **Advantages:**
 - **Rapid charge & discharge**
 - **No chemical degradation – huge numbers of cycles**
- **Disadvantage:**
 - **Low energy density – can’t store as much as a battery**

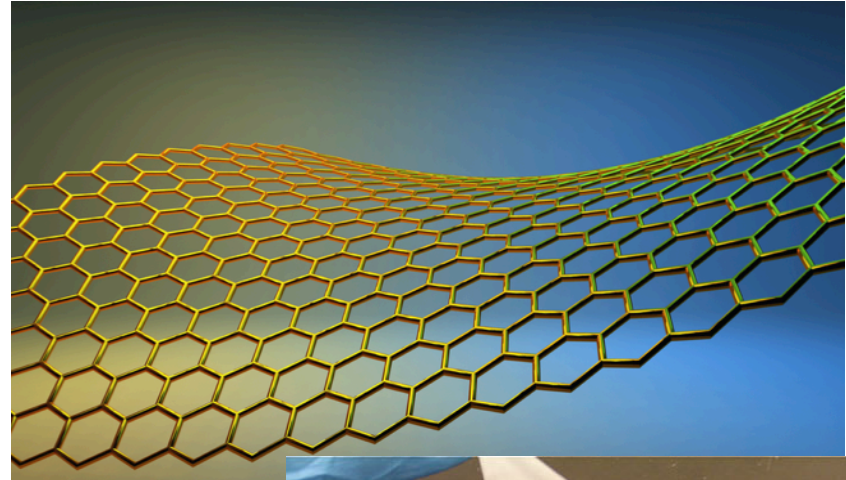


- **+** positive charge
- **-** negative charge

Charged double layer persists after power source is removed, providing a source of stored electric energy.

Back story – Part 2: Graphene

- Graphene ‘discovered’ in 2003
 - A new form of carbon – a sheet of carbon atoms one atomic layer thick
- Amazing properties
 - Super strong & elastic
 - Highly electrically conductive
 - Highly thermally conductive



The Nobel Prize in Physics 2010

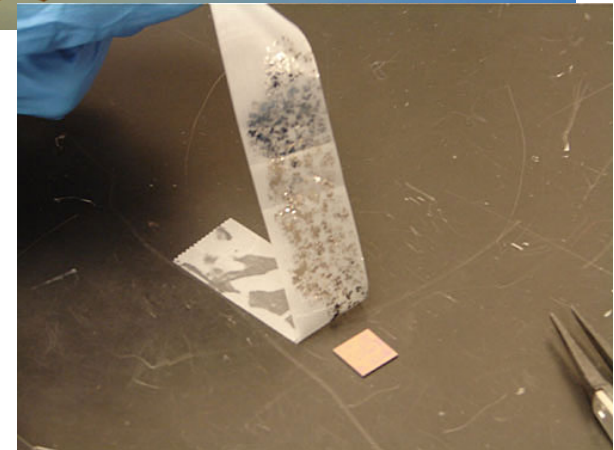


Photo: U. Montan
Andre Geim



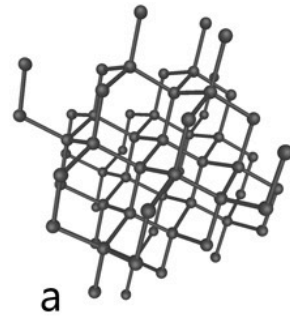
Photo: U. Montan
Konstantin Novoselov

The Nobel Prize in Physics 2010 was awarded jointly to Andre Geim and Konstantin Novoselov "for groundbreaking experiments regarding the two-dimensional material graphene"

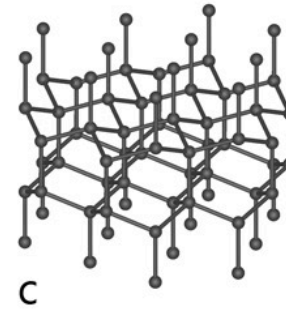
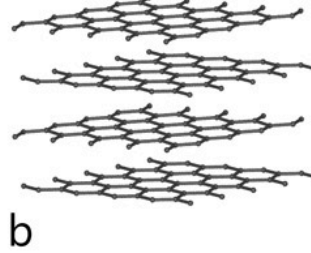


Back story – Part 2: Graphene

Diamond

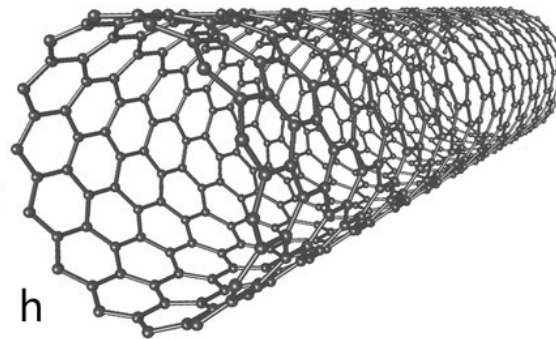
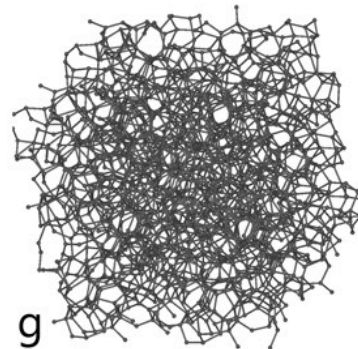
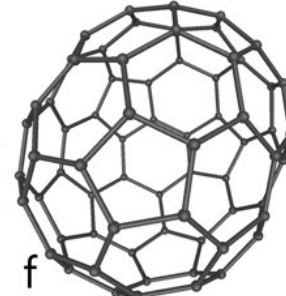
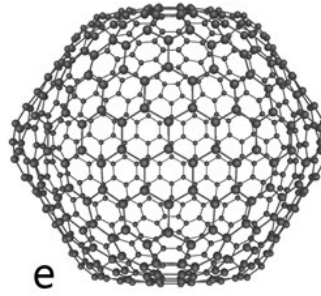
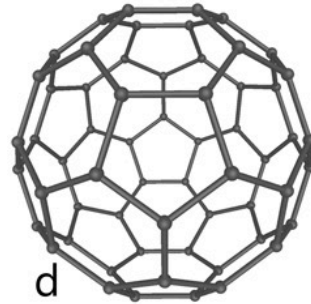


Graphite



“Bucky-balls”

Nobel Prize in Chemistry to Smalley, Kroto and Curl in 1996

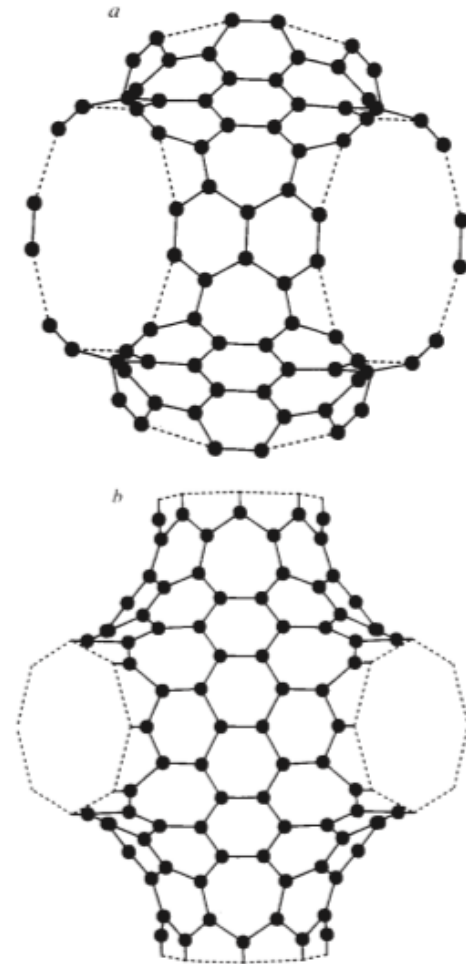
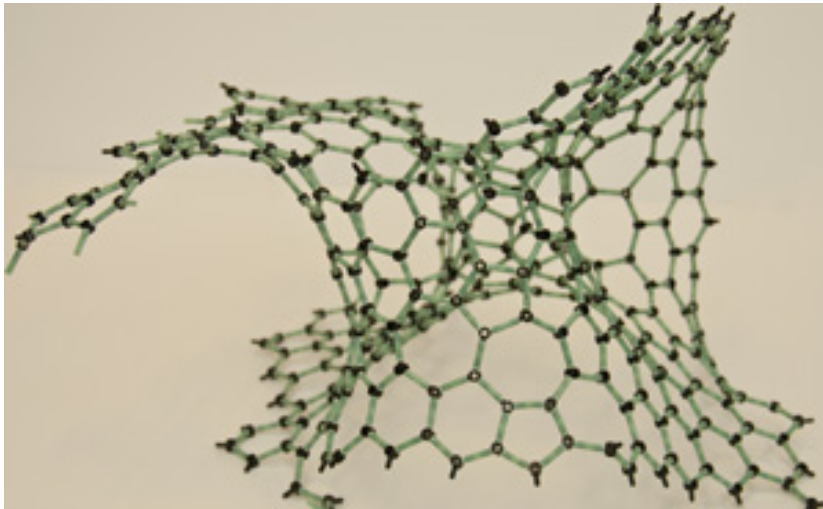


Carbon Nanotubes

Inaugural “Kavli Prize” for Nanoscience, to Sumio Iijima in 2008

Back story – Part 2: Graphene

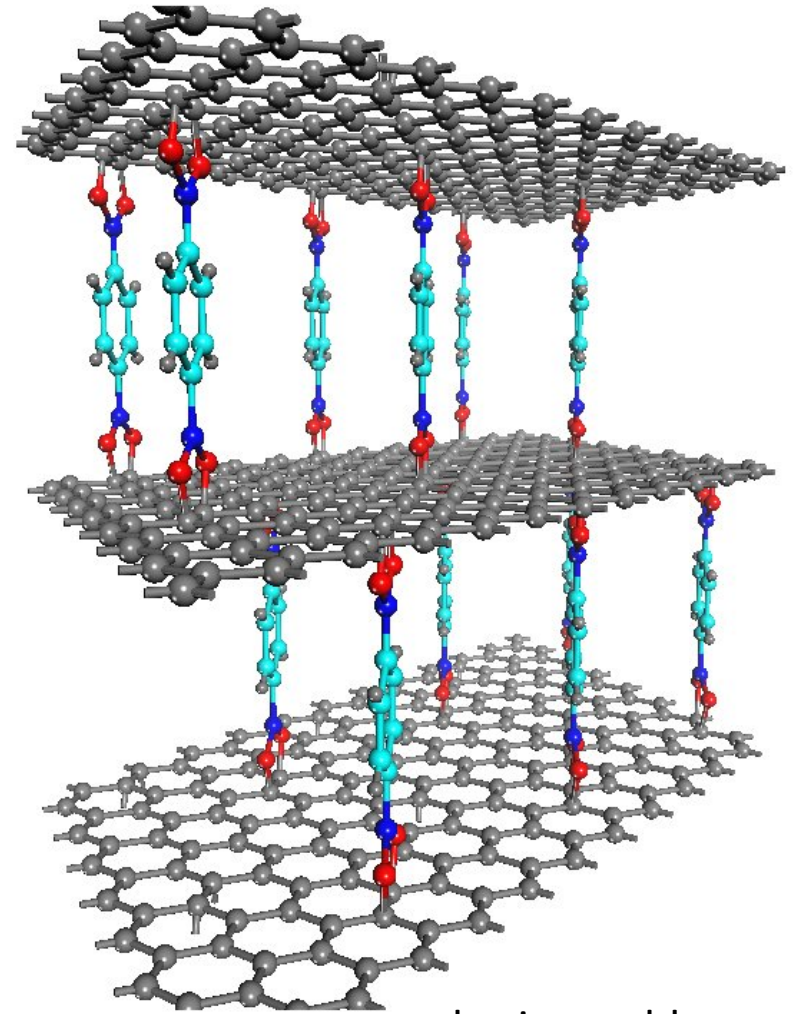
- Why not ‘negative curvature’ carbon?
- Theoretically predicated to be energetically favorable*
- Despite 20+ years since predication, no prior observations



*Lenosky, et al., Nature, 355, 1992

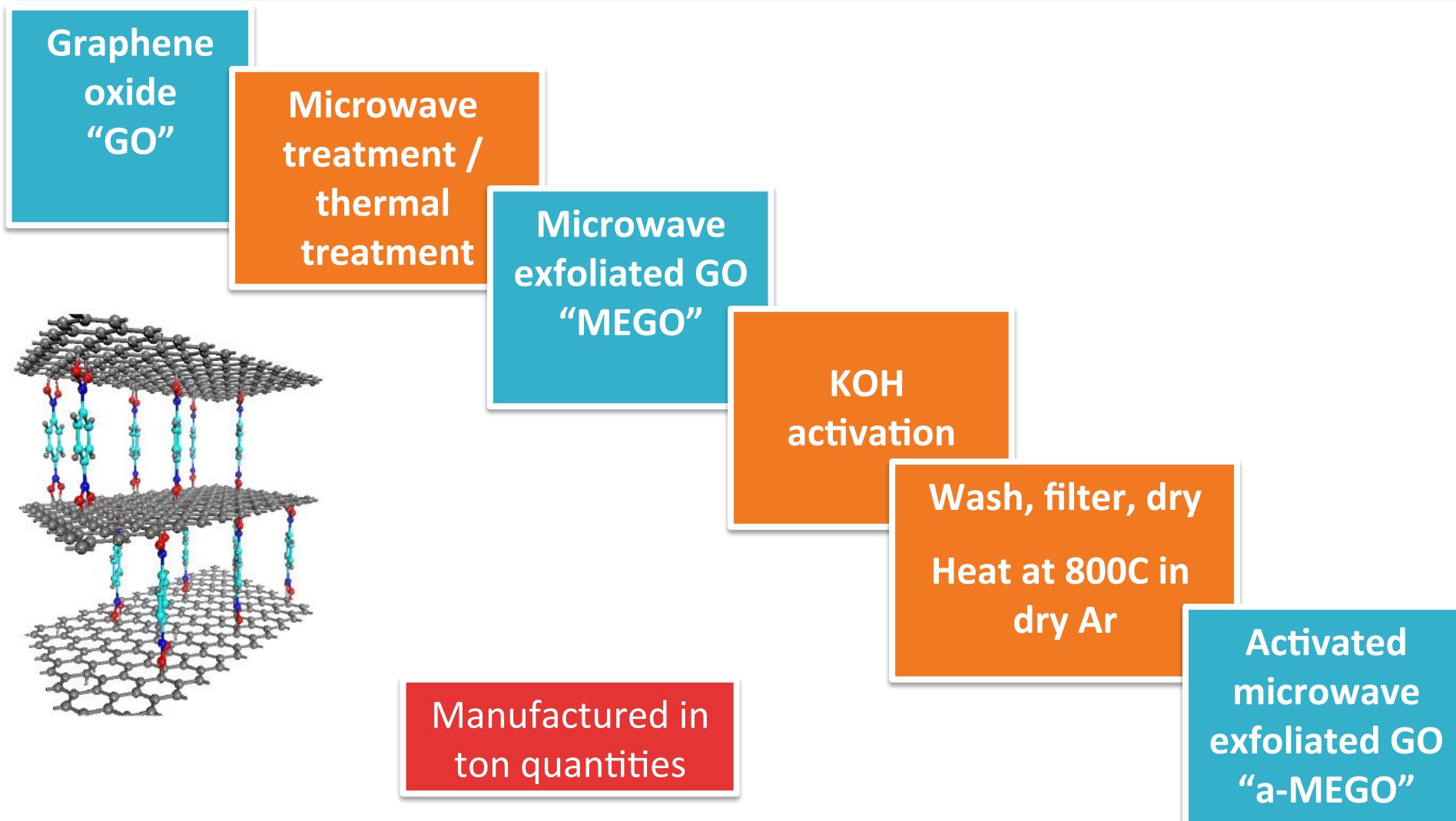
Back story – Part 3: Graphene oxide

- Graphene oxide is a material that is available in ton quantities
- Oxygen ‘linking groups’ hold together graphene sheets
 - These groups can be broken through heat or chemical attack



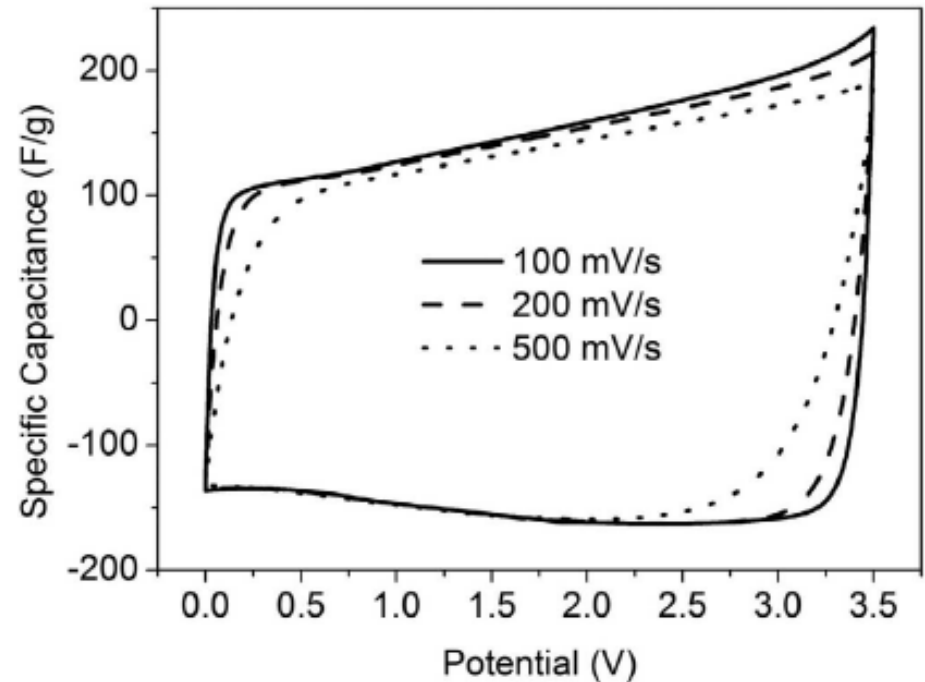
www.physicsworld.com

Ruoff group's synthesis approach



Electrochemical performance

- Extraordinary performance as a supercapacitor
 - **Estimated practical energy of a packaged device of 20 Wh/kg**
 - 4 times higher than existing supercapacitors based on activated carbons
 - $\approx 70\%$ of a lead-acid battery



What did the Ruoff group make?

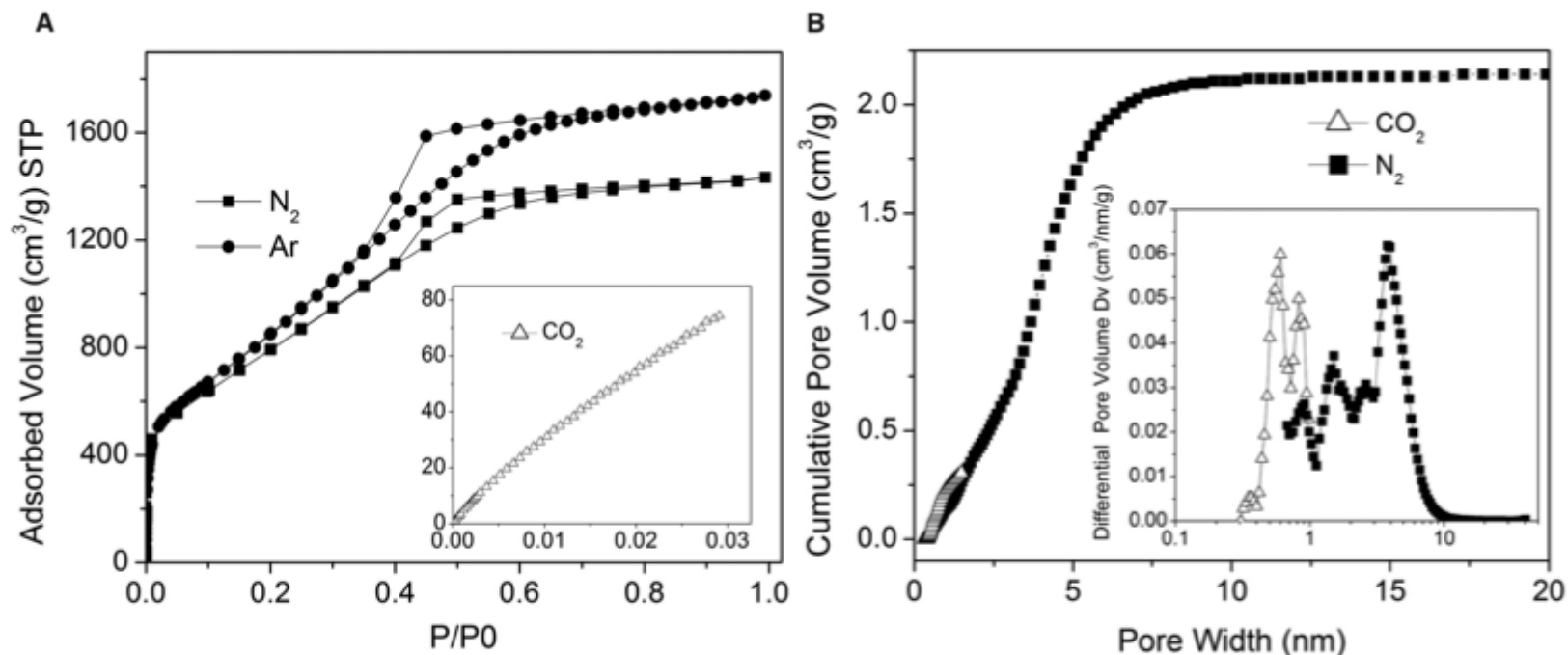


Fig. 3. Gas adsorption/desorption analysis of an a-MEGO sample (SSA ~ 3100 m²/g). **(A)** High-resolution, low-pressure N₂ (77.4 K) and Ar (87.3 K) isotherms. (Inset) The CO₂ (273.2 K) isotherm. **(B)** Cumulative pore volume and (inset) pore-size distribution for N₂ (calculated by using a slit/cylindrical NLDFT model) and CO₂ (calculated by using a slit pore NLDFT model).

- By measuring infiltration of N₂, Ar and CO you can determine the size of pores
- It was found to be a highly porous material
 - **Distribution of pore sizes between 0.6 nm and 5 nm**

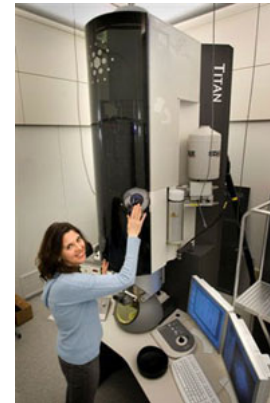
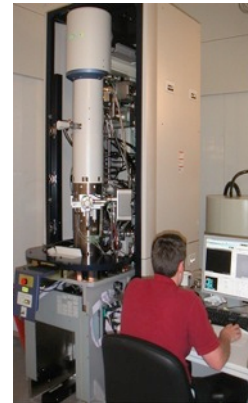
This is where BNL joined in ...

- **DOE Labs have amazing resources**
 - **NSLS – bright source of x-rays**
 - **CFN – powerful electron microscopes**
 - **National Center for Electron Microscopy at Lawrence Berkeley National Lab**

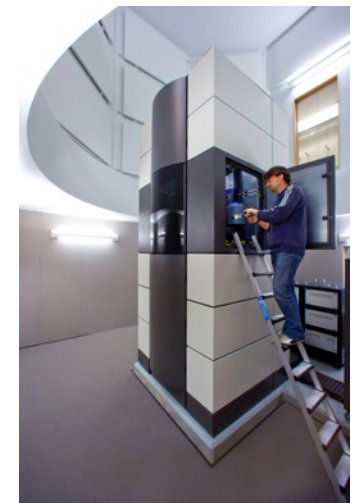
National
Synchrotron
Light Source
at BNL



Two 'aberration corrected'
electron microscopes at the CFN



The TEAM instrument at the
National Center for Electron
Microscopy



Diffraction of x-rays & electrons

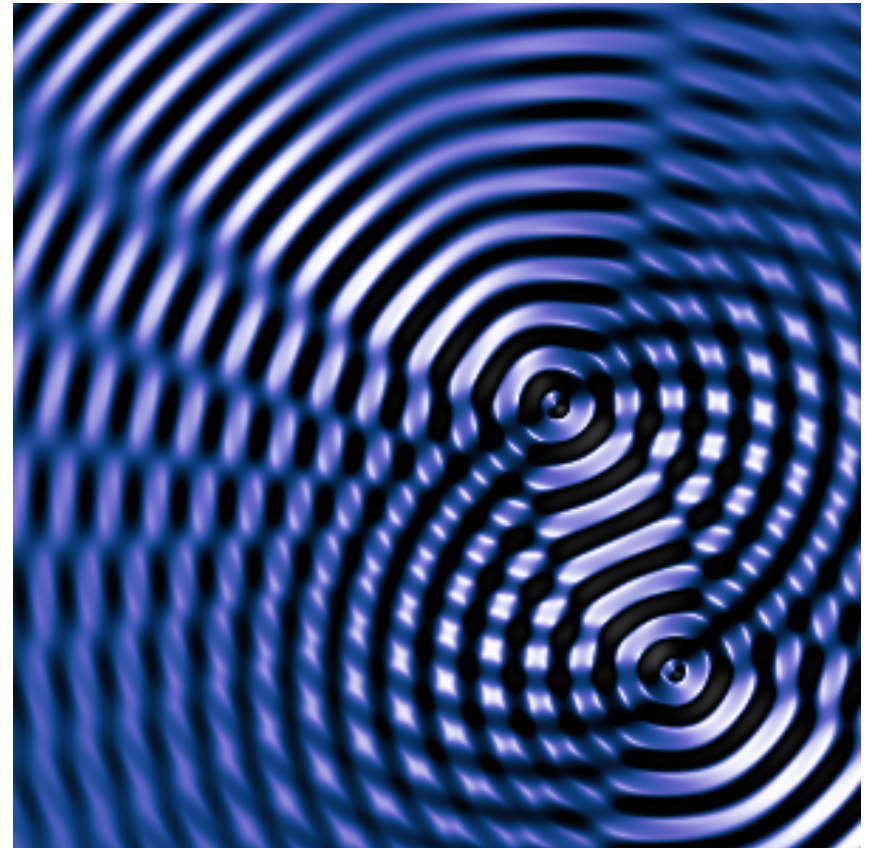
- Incident waves of x-rays / electrons bounce off of atoms.
- If atoms are arranged in a ordered structure there are specific angles where the waves interfere constructively

- $2d \sin\theta = \lambda$

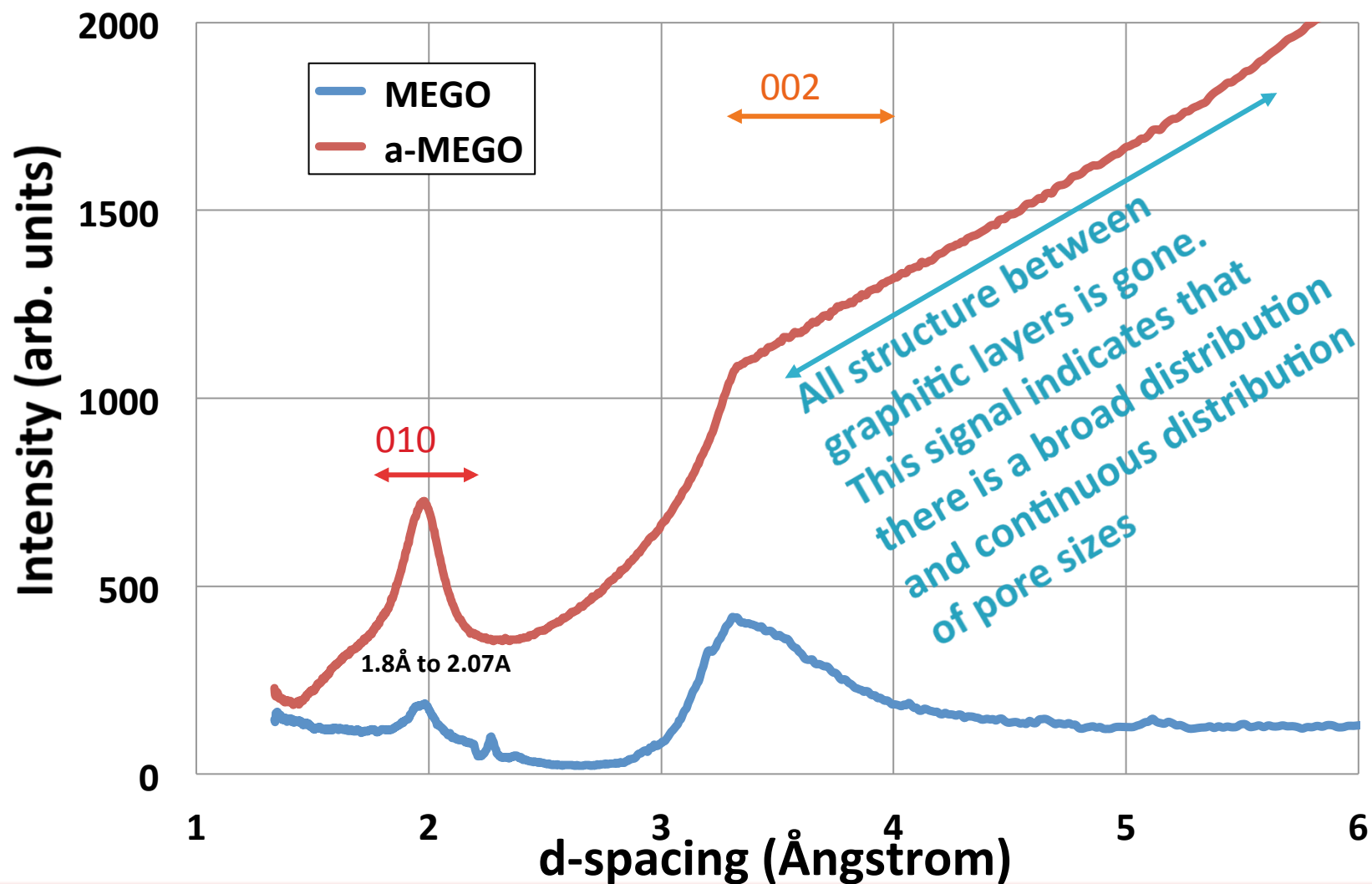
Distance
between
atomic planes

Angle

Wavelength



NSLS: Diffraction from MEGO & a-MEGO



Transmission electron microscope

Transmission electron microscope

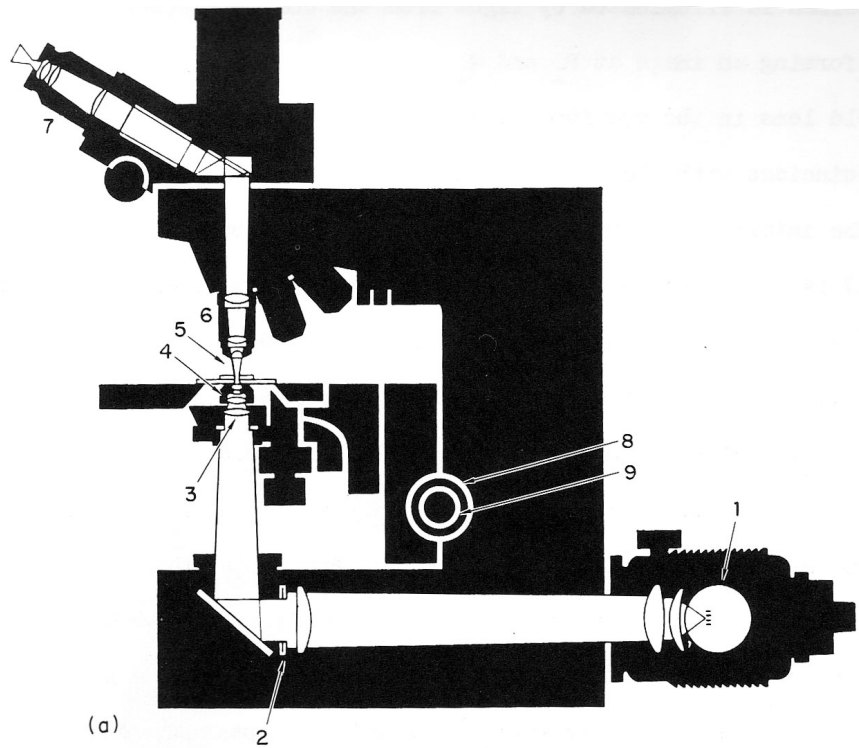
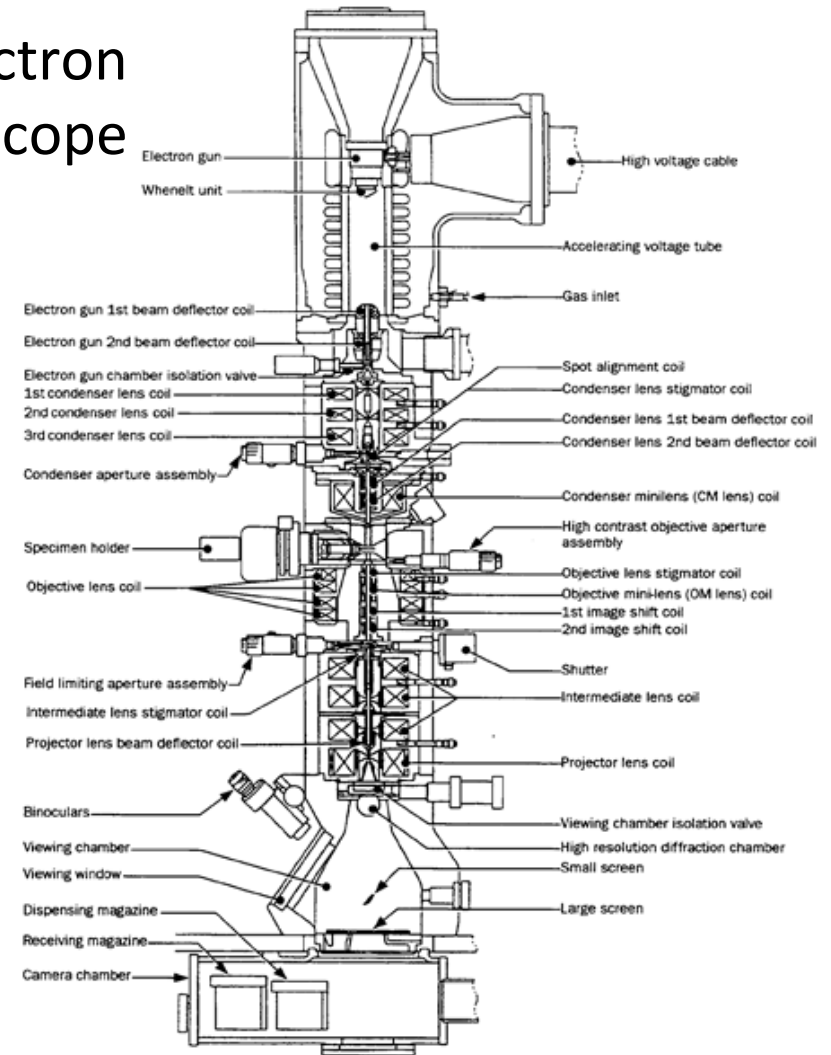


Table-top optical microscope



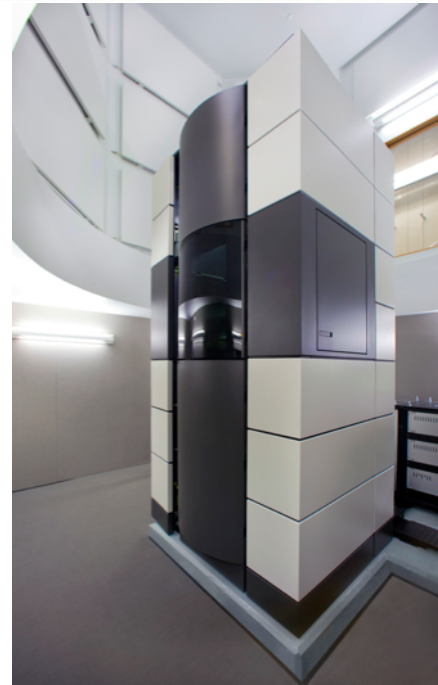
The best microscopes in the world ...



Hitachi HD2700C STEM
Probe corrector at the
CFN @ BNL



Titan 80-300 - ETEM
Image corrector at
the CFN @ BNL

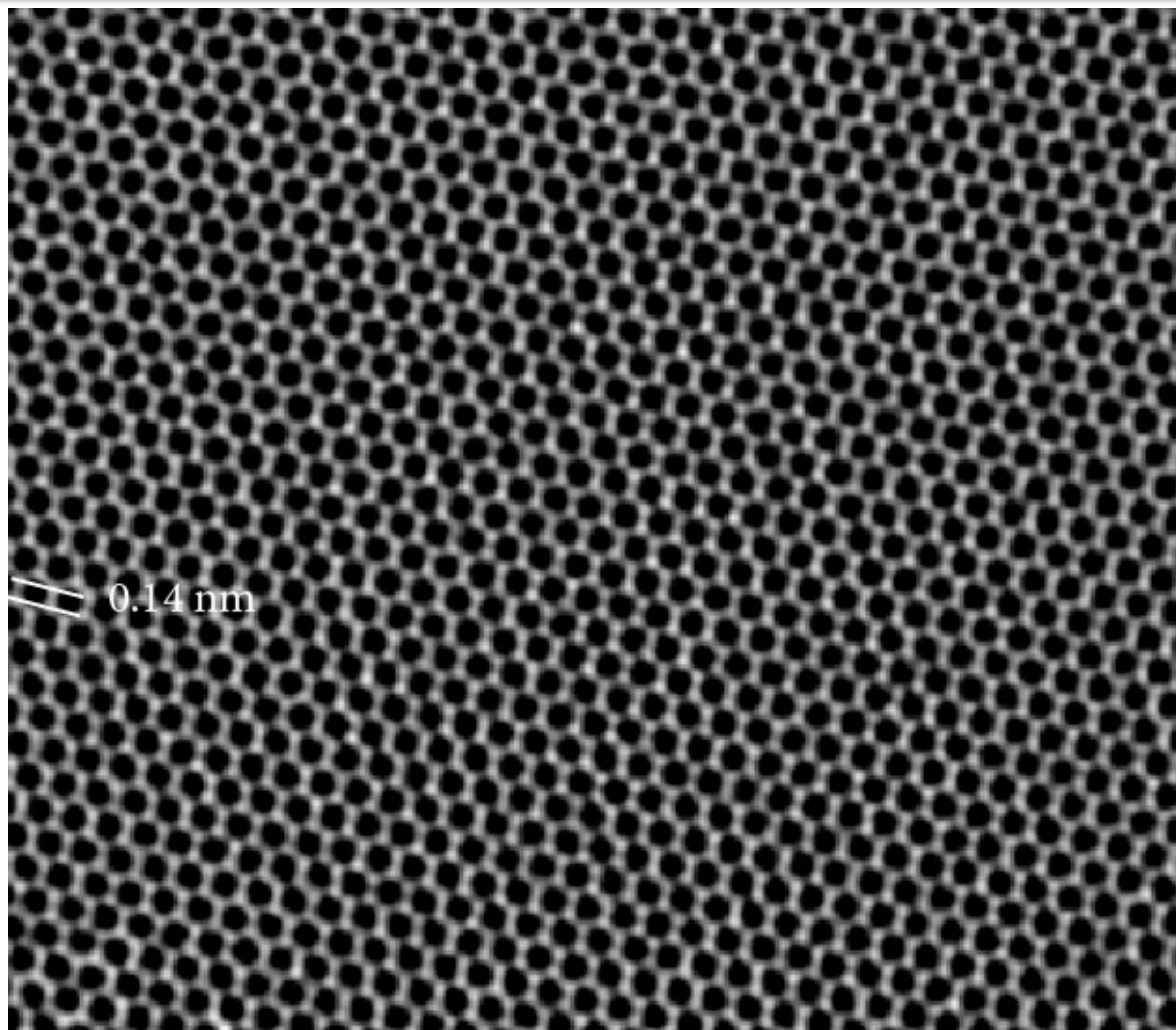


TEAM 1.0 instrument
@ the National
Center for Electron
Microscopy @ LBNL



“Triple-C”
instrument at AIST in
Tsukuba, Japan

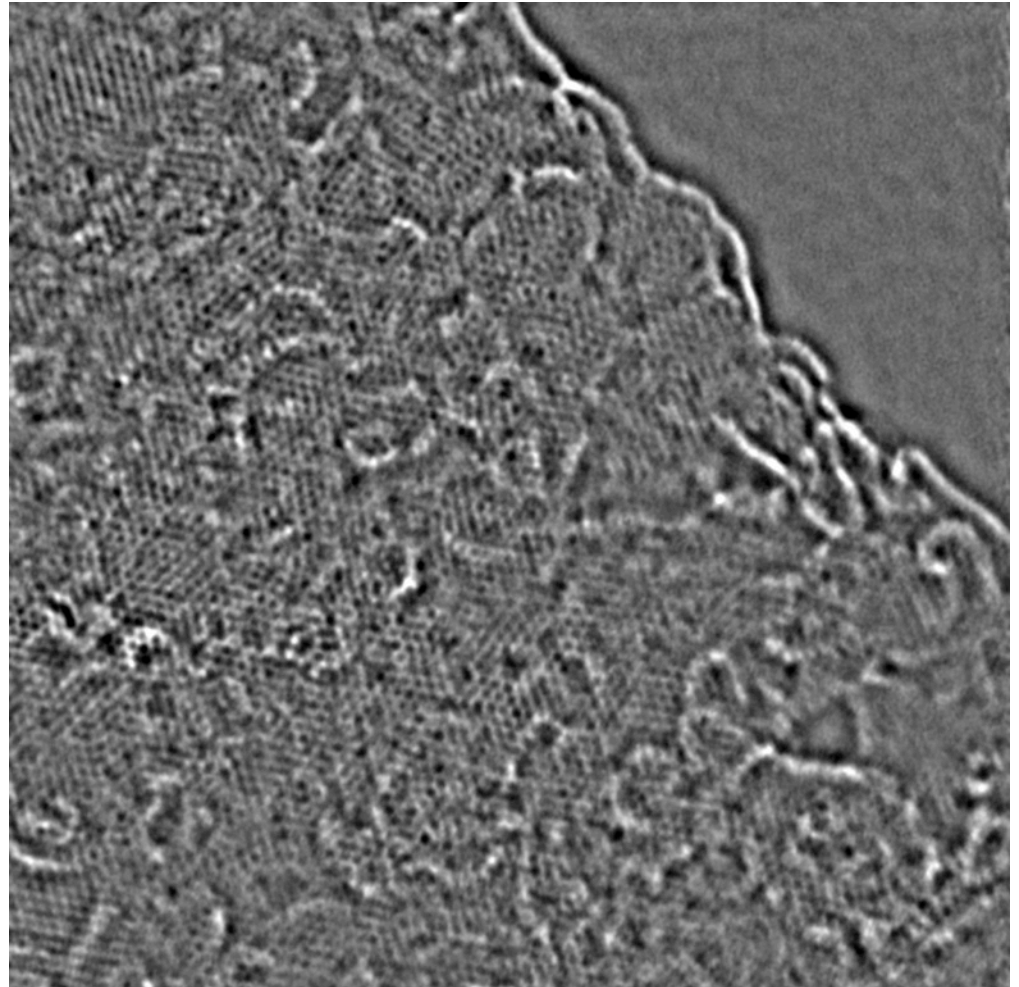
What does graphene look like in the TEM?



<http://ncem.lbl.gov/frames/TEAM0.5.htm>

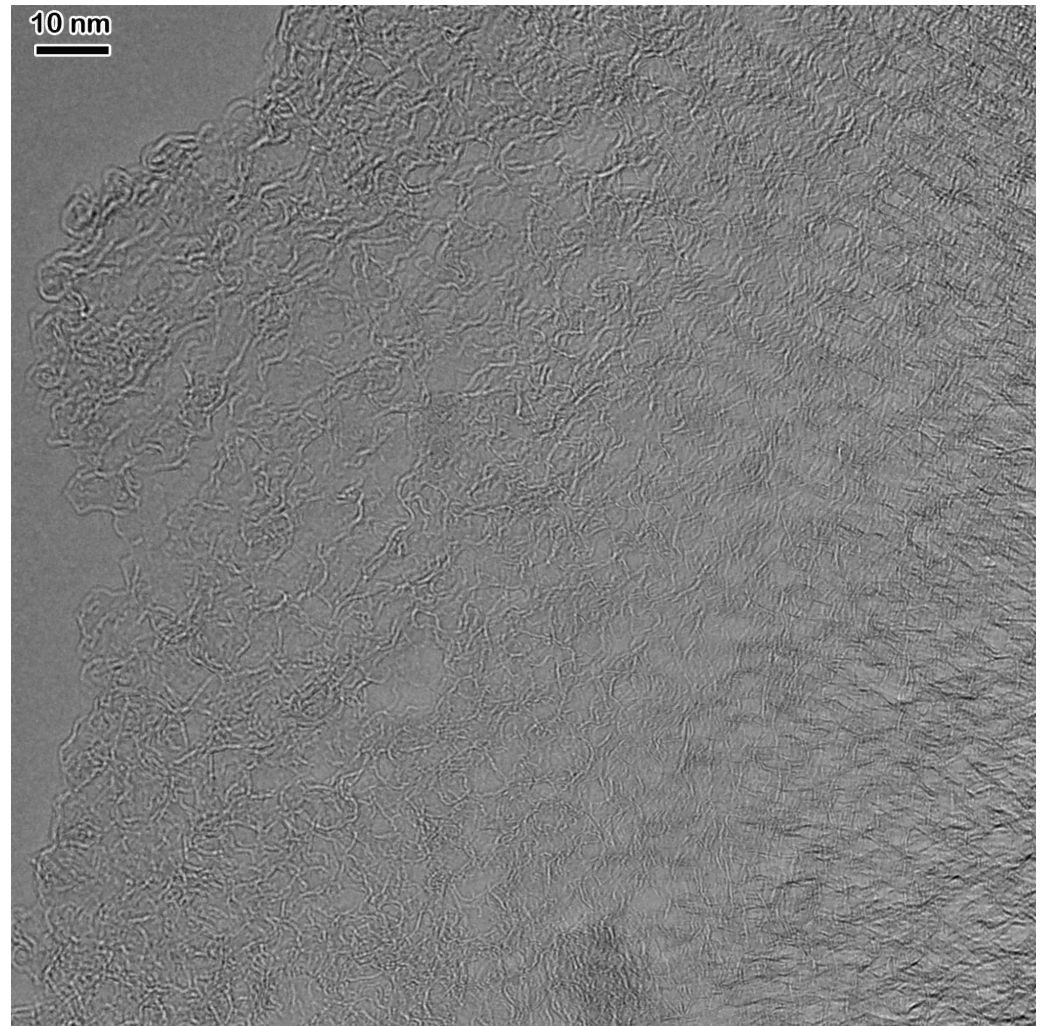
HRTEM images of MEGO

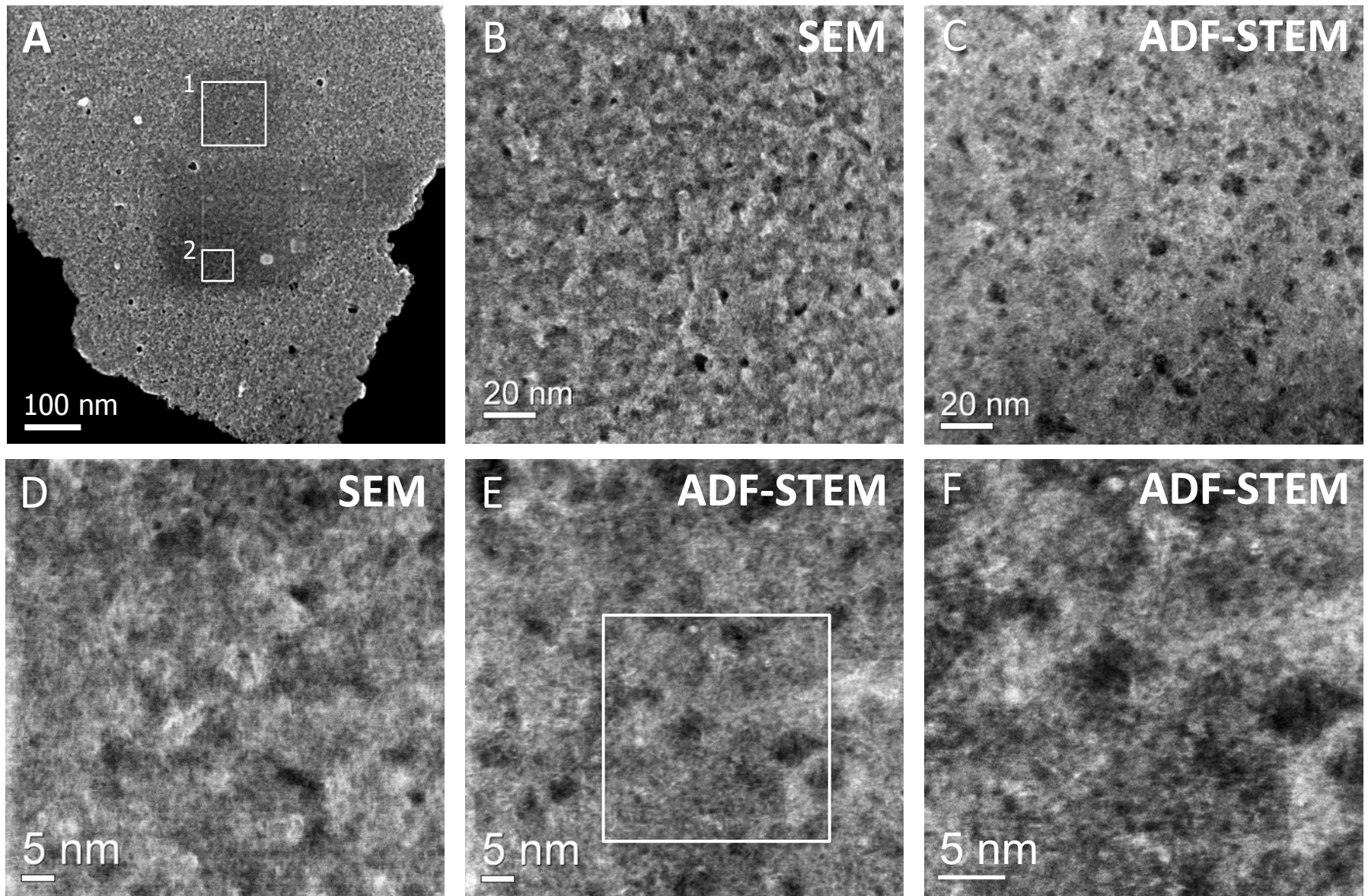
- Exit wave reconstructed image of MEGO
 - **TEAM 1.0: Operated at 80kV**
- Quite defective in plane
- A bit of crinkle out of plane
- Image consistent with diffraction results



Phase contrast imaging of pores

- **Movie shown is a focal series of images @ 80 kV**
 - **Taken using CFN Titan electron microscope**
- **A continuous network of pores visible**
 - **Size range from $\approx 5\text{\AA}$ to 5 nm**
- **Consistent with BET measurements**
 - **Indicate a specific surface area of $\approx 3100\text{ m}^2/\text{g}$**
 - **At the theoretical limit for flat graphene!**



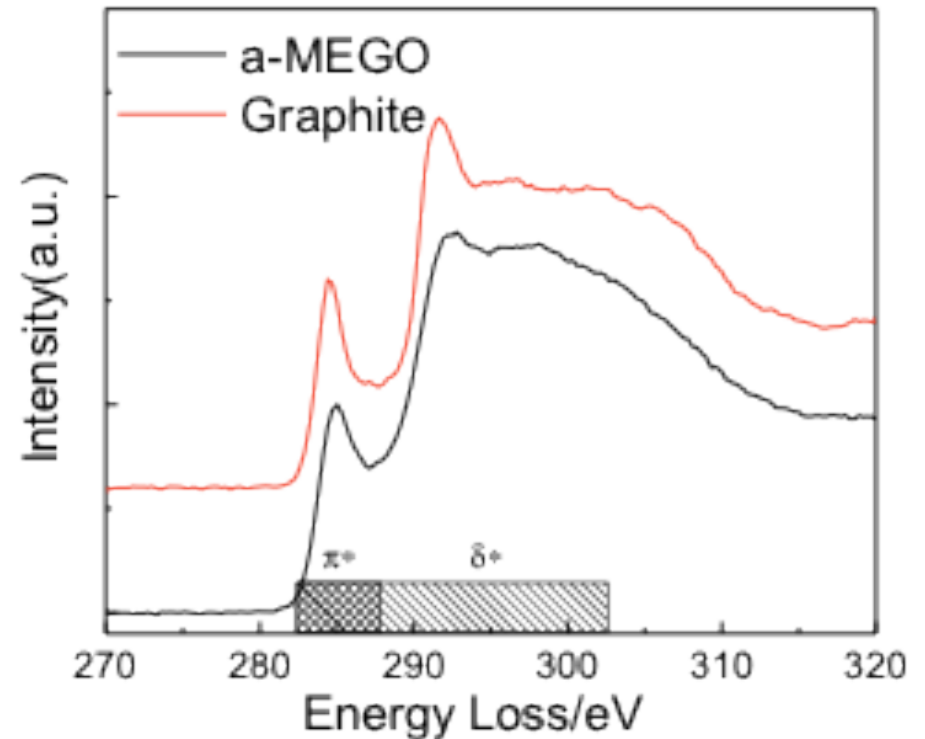


Low magnification SEM (A) and high magnification SEM (B) and ADF-STEM (C) images of a-MEGO. (B and C) are from the region 1 in (A). Larger pores of between 2 – 10 nm are clearly resolved. Very high magnification SEM (D) and ADF-STEM (E,F) images taken from the region 2 in (A). These images indicate that the entire microstructure is composed of very small pores, of order of ≈ 1 nm in size, as is evident in the magnified portion shown as (F).

Taken using CFN Hitachi instrument

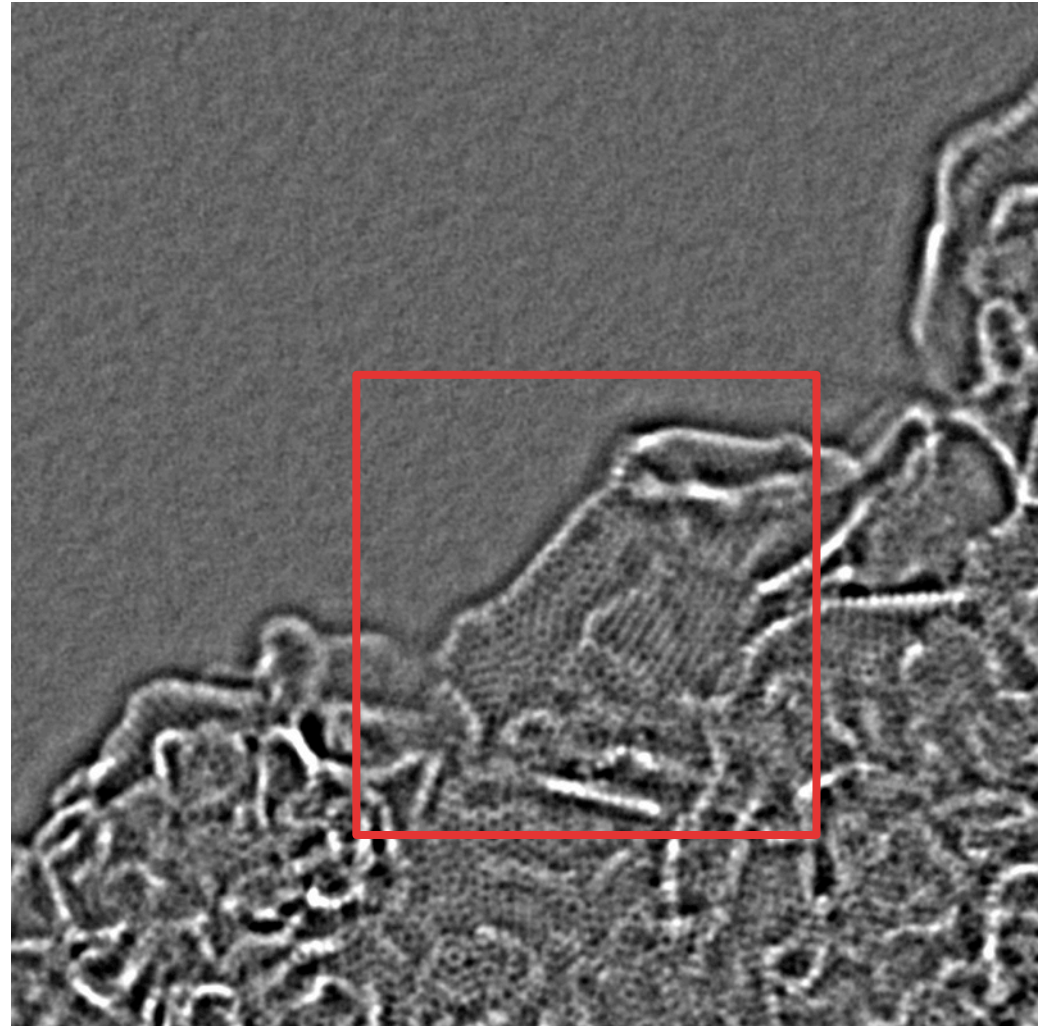
Energy loss spectroscopy from a-MEGO

- Carbon K edge shows a-MEGO to be sp^2 bonded
 - **Just like graphite**
- This means that the structure has returned to a form like the graphene
 - **At least as far as the electrons are concerned!**
 - **All the atoms are connected**

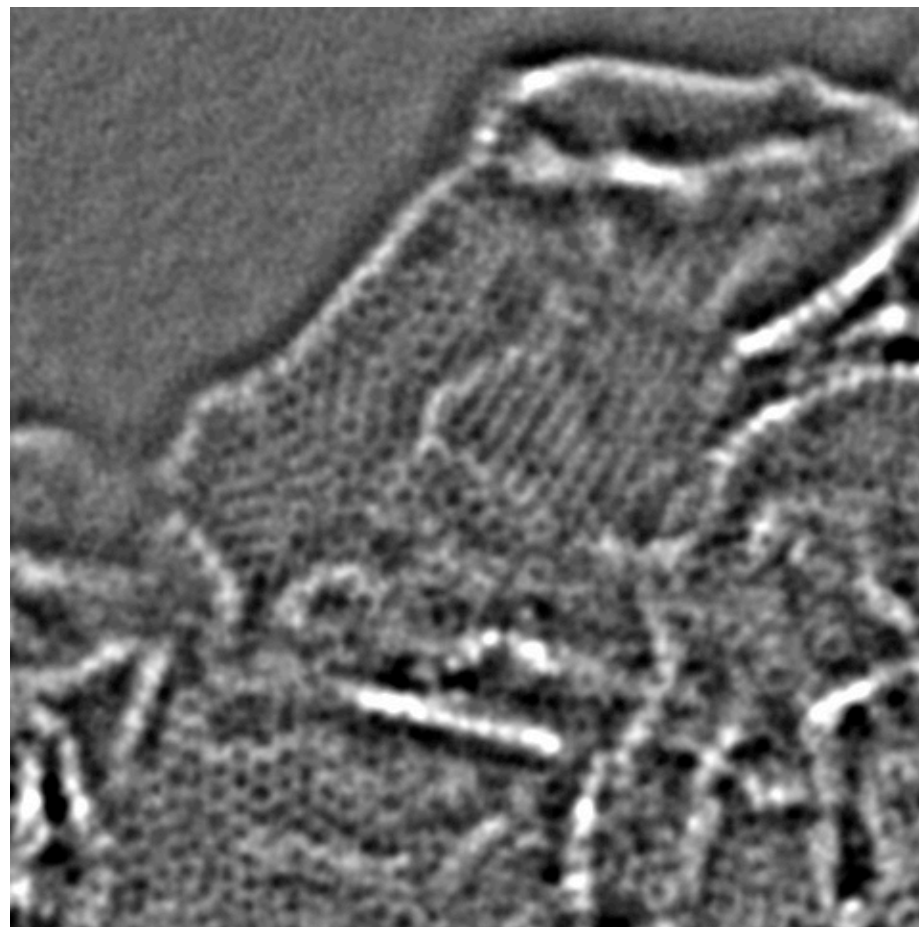
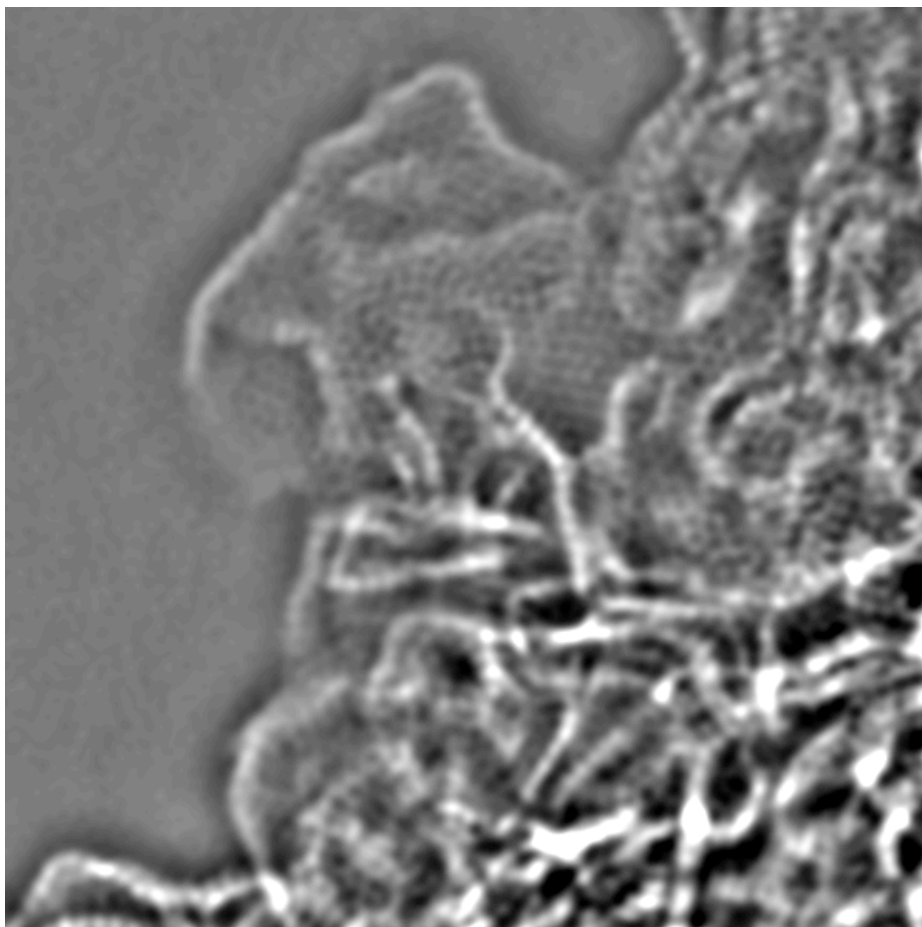


Ultrahigh-resolution images of a-MEGO

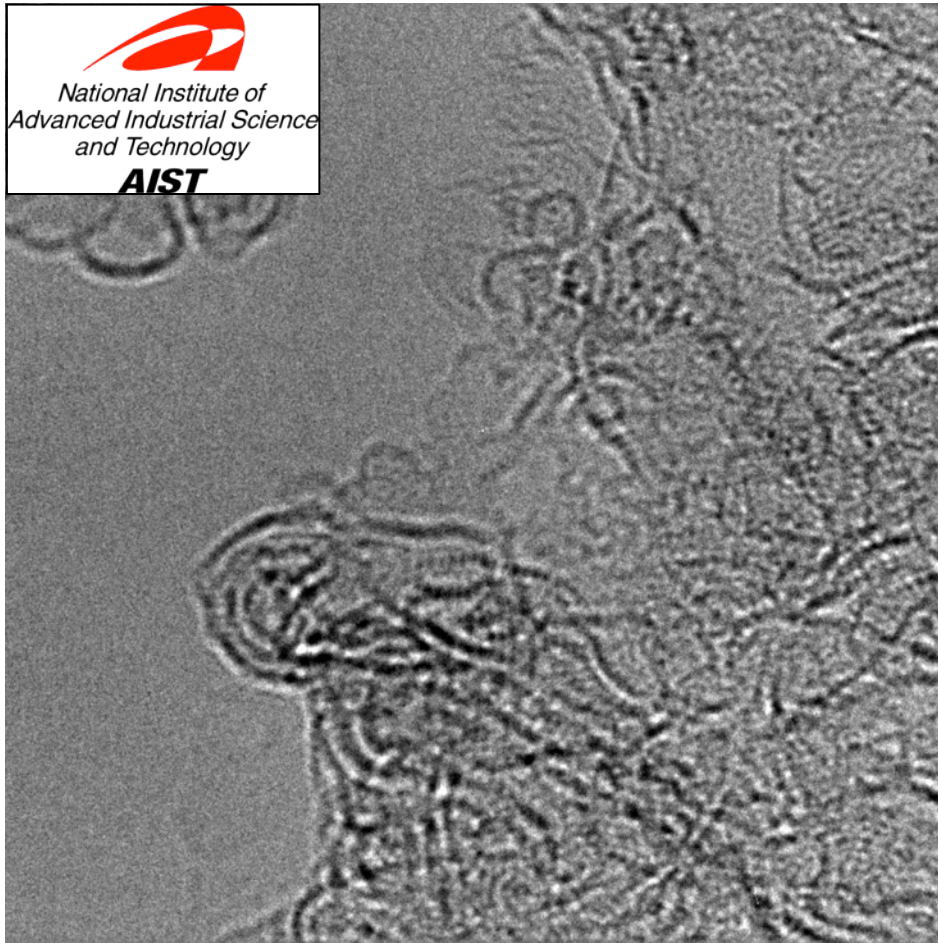
- Exit wave reconstructed image of a-MEGO from TEAM instrument
 - **At low voltage = little beam damage**
- A-MEGO is composed of highly curved, crystalline carbon
- Single sheet carbon, with only occasional 002 correlations



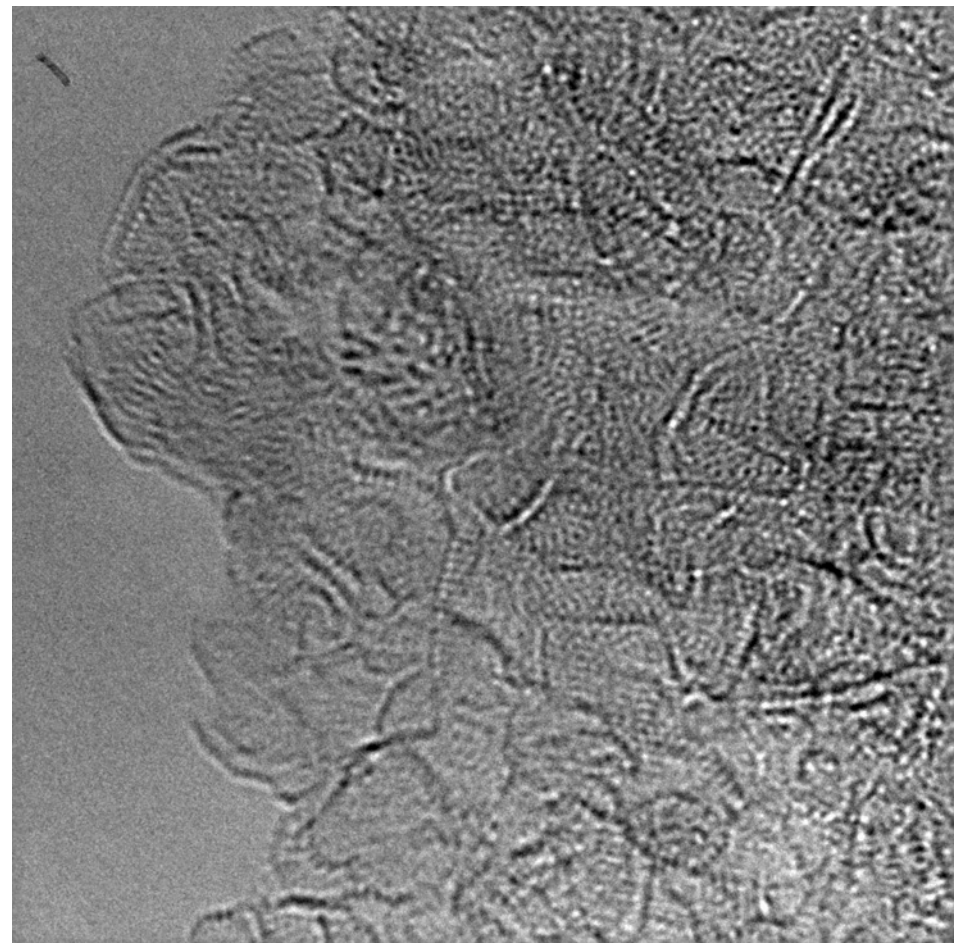
EWR images of a-MEGO



“Triple C” instrument: low dose / low voltage



20 summed images



100 summed images

With Kazu Suenaga

Summary of a-MEGO structure

- a-MEGO is composed of individual sheets of carbon
- This carbon is sp^2 bonded – just like graphite
- These sheets are crystalline, “in-plane”
- They are composed of 5, 6, 7 and 8 membered rings of carbon atoms
- Nearly all of the atoms are bonded to other carbon atoms
- These sheets are highly curved
- They form a broadly enclosed three-dimensional network of micro- (6Å - 1nm) and mesopores (2 to 10 nm)

**a-MEGO is a fundamentally new form of carbon,
with extraordinary properties**

Wu, et al., Science, 332, 1537, 2011.

> 650 Citations to date

Carbon-Based Supercapacitors Produced by Activation of Graphene

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Activated Graphene Makes Superior Supercapacitors for Energy Storage

New material combines high storage capacity with quick energy release and unlimited recharge

May 12, 2011

