

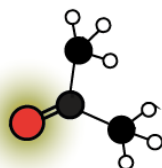
Directional cuts through potential energy surfaces



ANNETTE PIETZSCH
Helmholtz-Zentrum Berlin

IXS conference June 2019

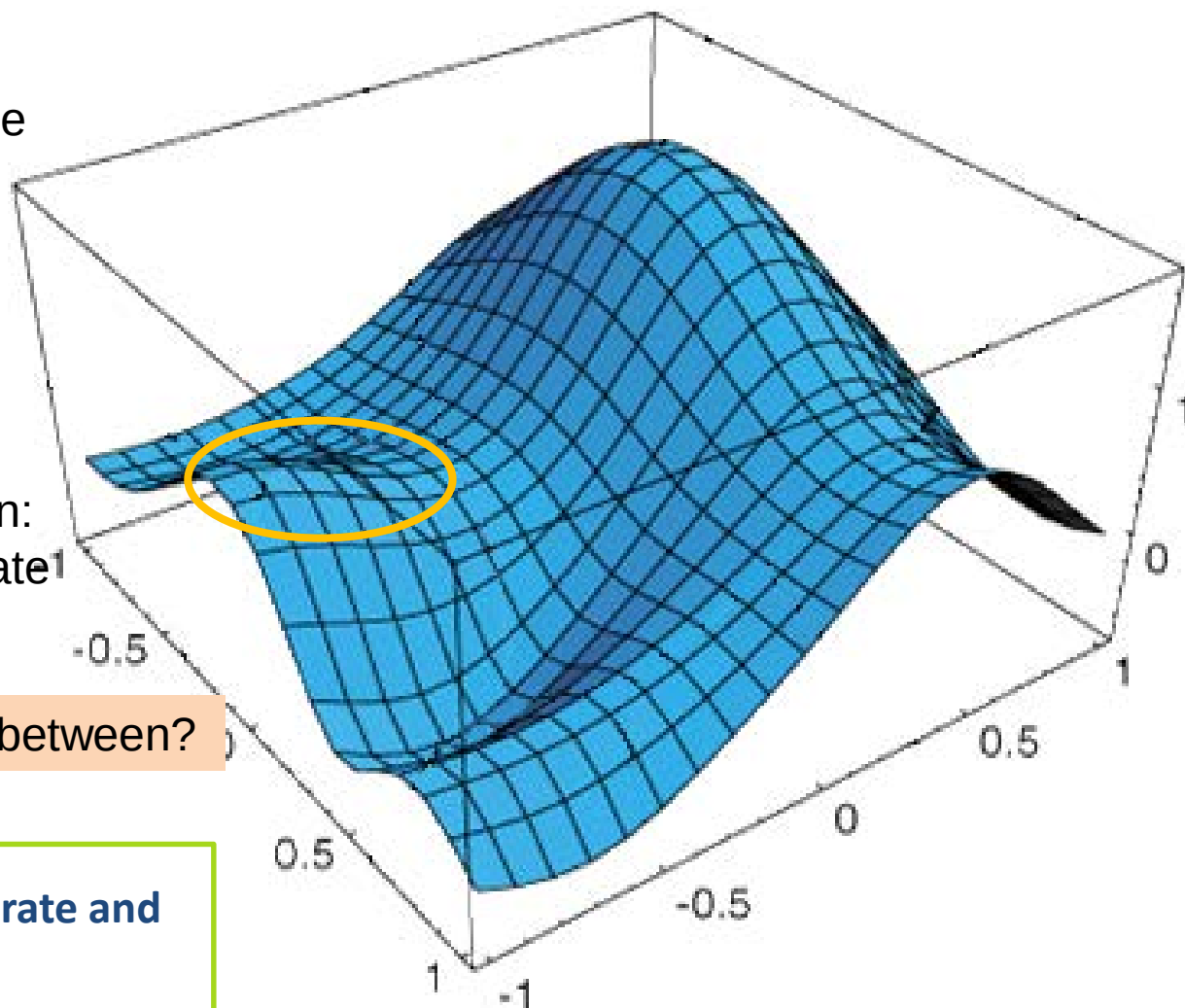
Describes energy of molecule
in terms of its structure



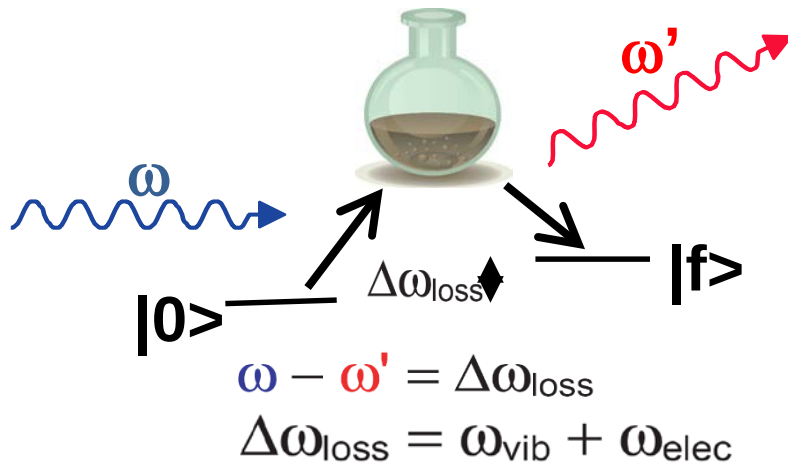
Equilibrium points well known:
Static molecule in ground state¹
(ball and stick model)

Excitation: what happens in-between?

- **Dynamics on PES control rate and selectivity**
- **Detailed info on PES needed!**

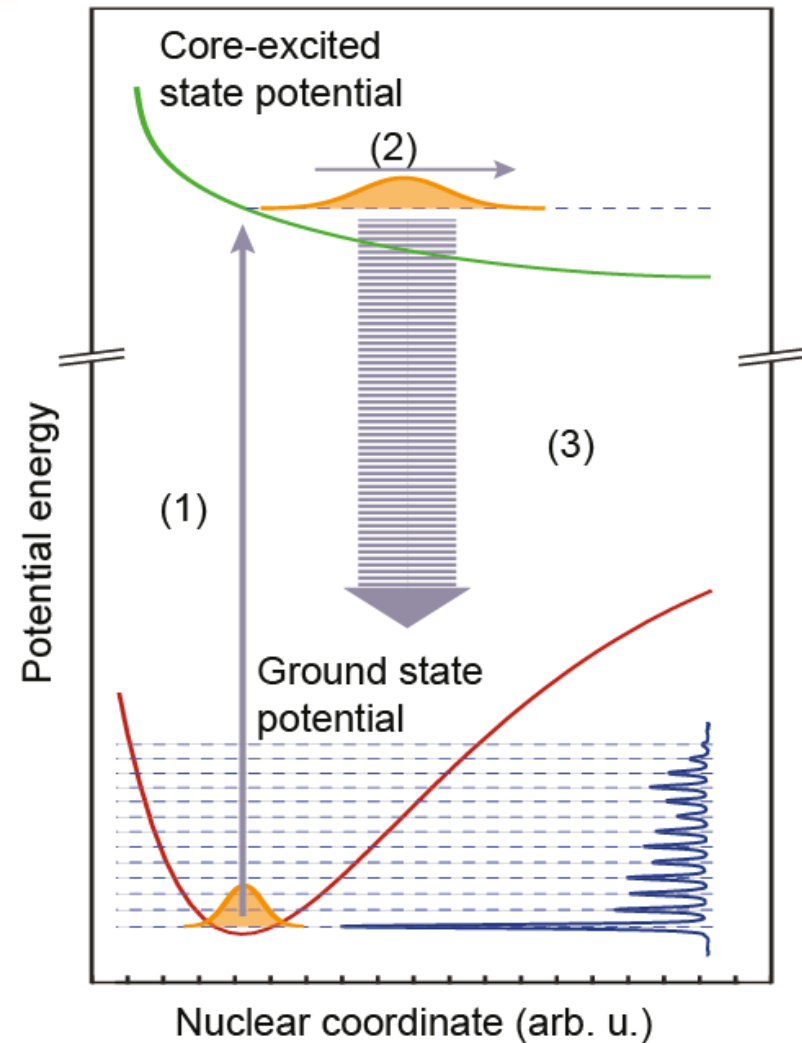


RIXS process



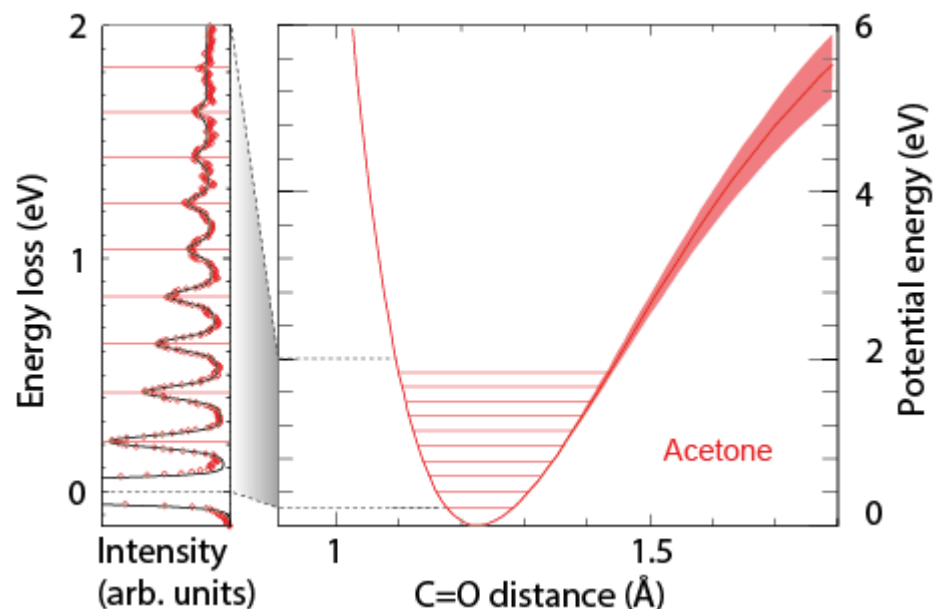
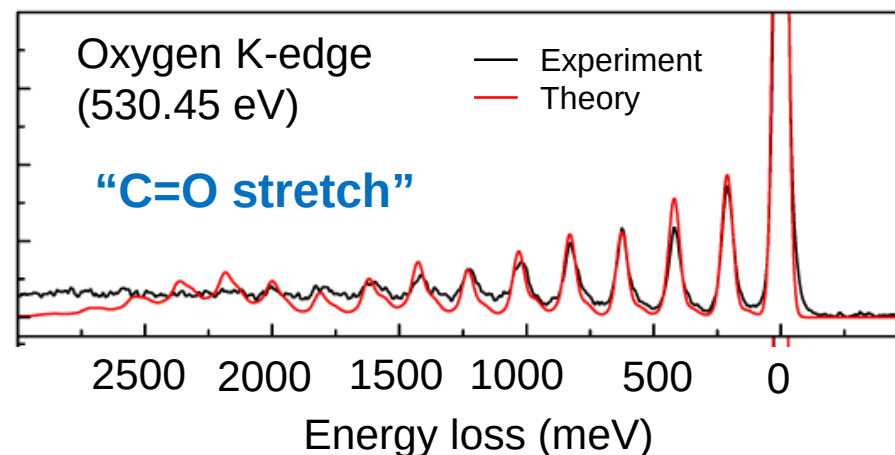
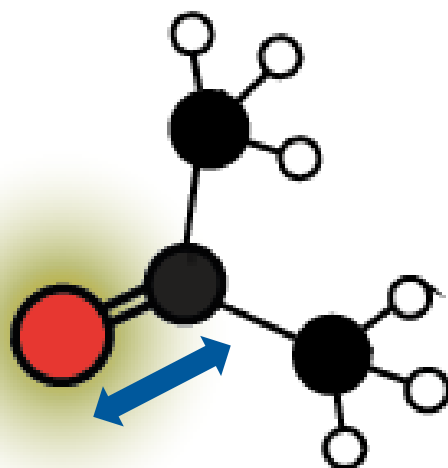
"Sub-natural linewidth Resonant inelastic X-ray scattering (RIXS)"

S. Schreck et al., *Sci. Rep.* 7, 20554 (2016)
Currently only at ADRESS/SLS



Vibrational progression – cut through potential energy surface!

liquid acetone: the potential along the C=O bond



Morse fit: potential along C=O bond

- Shows behavior further away from equilibrium
- 1D cut

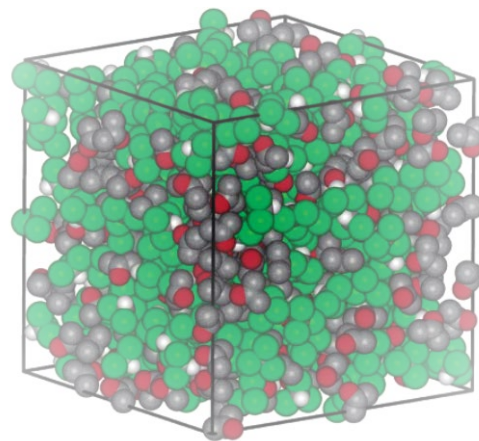
Y.P.Sun et al., *Phys. Rev. B* **84**, 132202 (2011)

Active sites interacting with the fluctuating solvent network

isolated small molecule



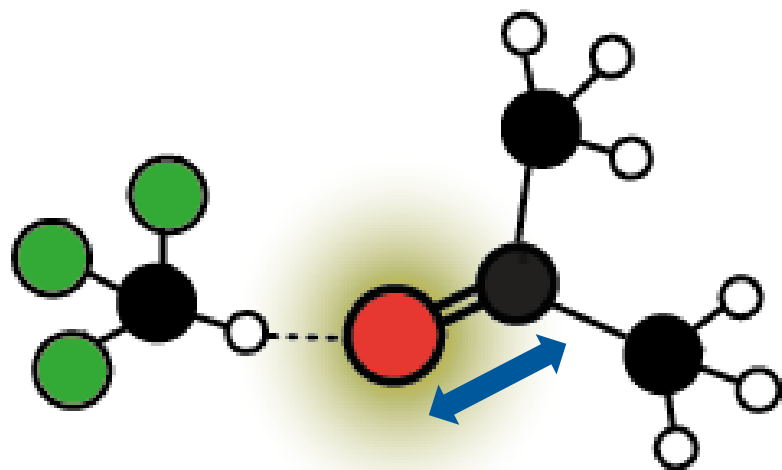
REALITY
(very often)



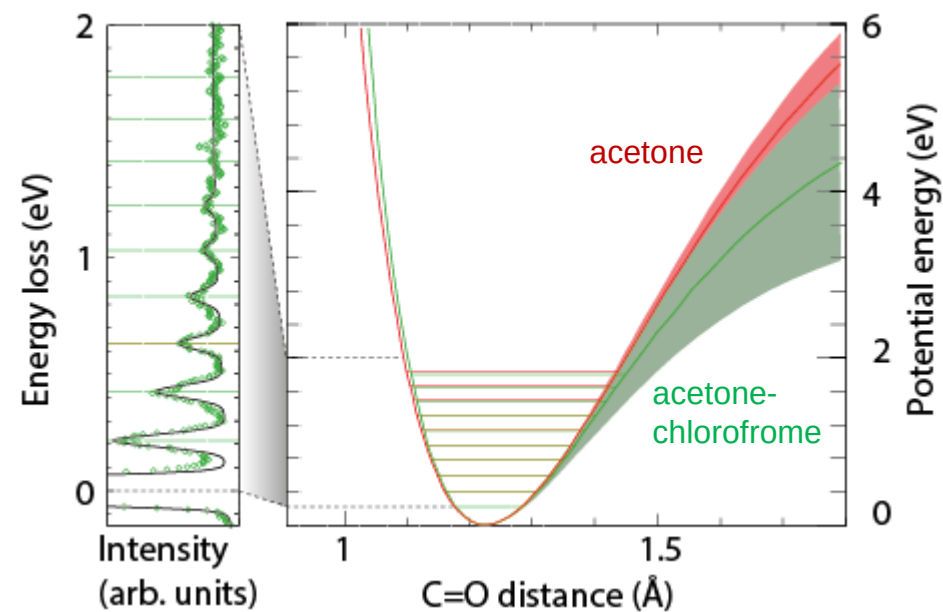
fluctuating bonding networks, solution....

? Potential energy surface along
interaction coordinate

liquid acetone: Adding one Hydrogen bond



Azeotrope acetone-chloroform mixture

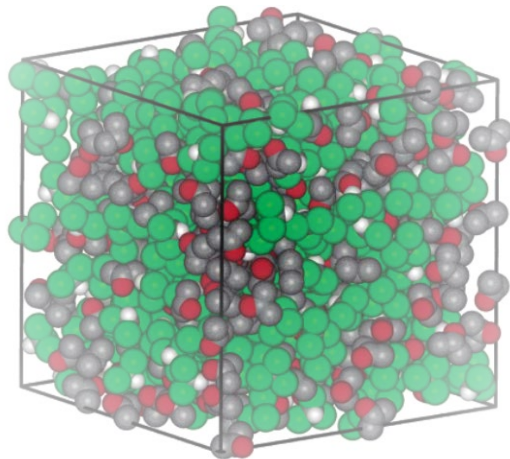


- ✓ Hydrogen bond weakens the C-O potential
- ✓ Quantitative plot

S. Schreck et al., Sci. Rep. 7, 20054 (2016)

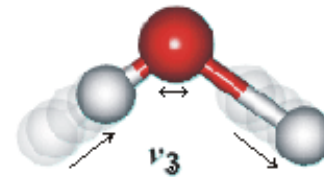
? Liquid water:

Bonding network
anomalies
coordination and correlation
structure: controversial

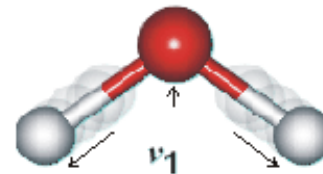


✓ The water molecule

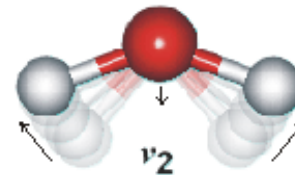
3-atomic molecule
3 normal modes



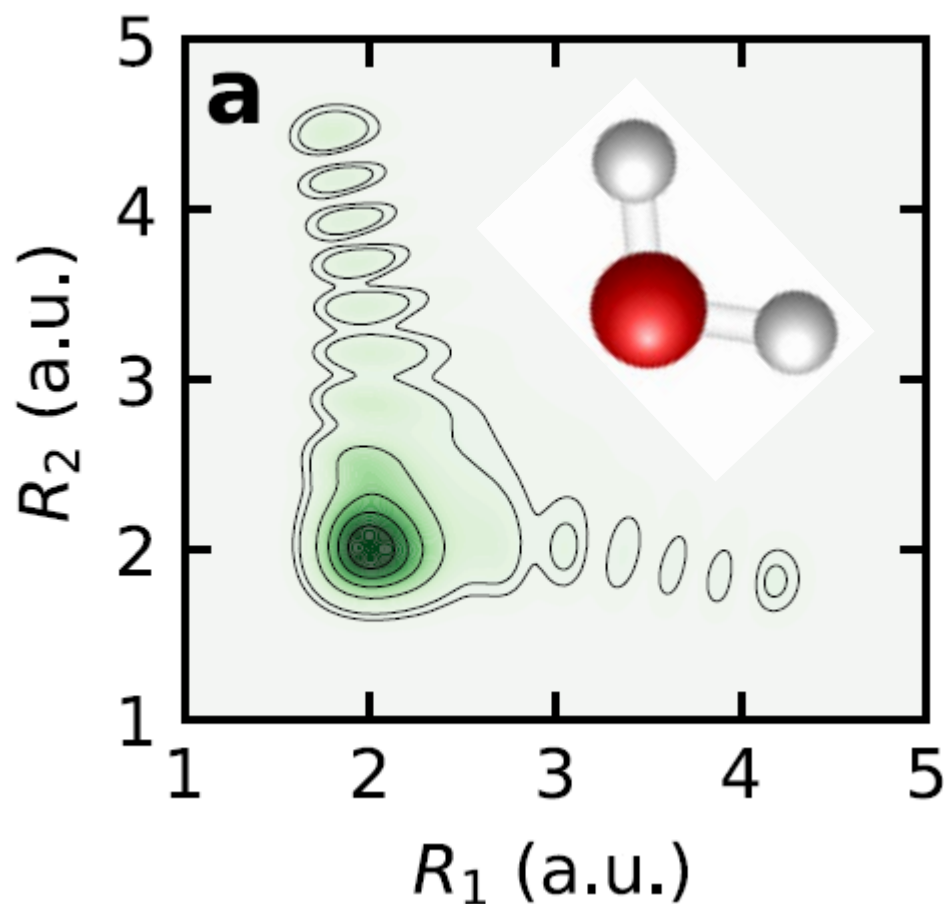
Asymmetric stretch



Symmetric stretch



Bend

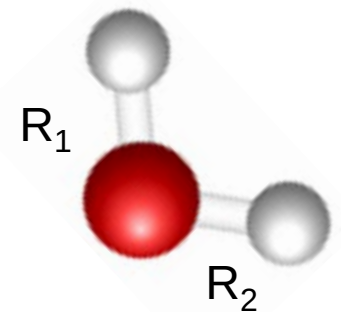


Coherent excitation of both OH bonds

Directional excitation within molecular H₂O

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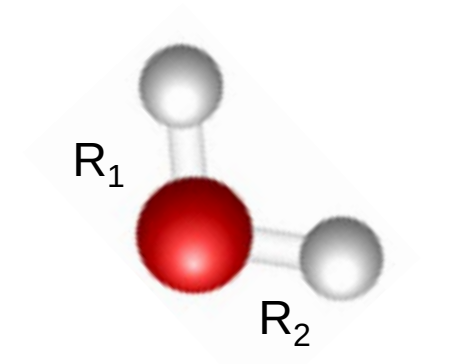
stretch

R. Couto et al, Nature Commun. 8, 14165 (2017)

Directional excitation within molecular H₂O

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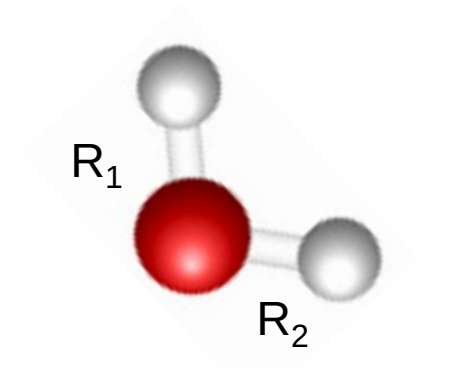
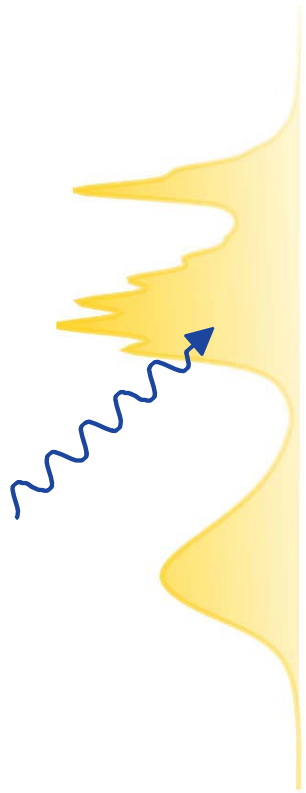
stretch

R. Couto et al, Nature Commun. 8, 14165 (2017)

Directional excitation within molecular H₂O

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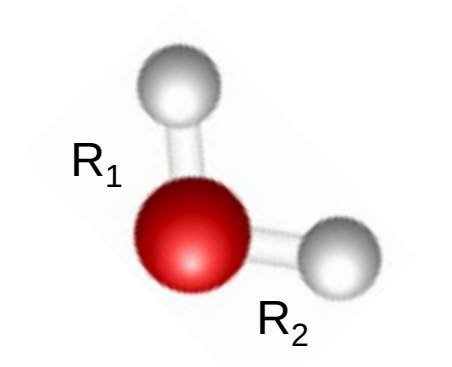
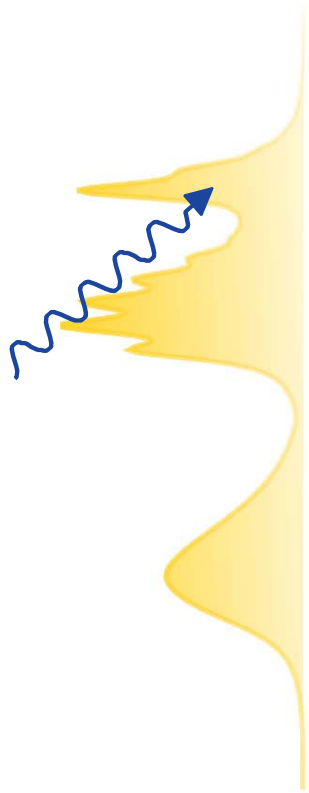
stretch

R. Couto et al, Nature Commun. 8, 14165 (2017)

Directional excitation within molecular H₂O

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stretch

R. Couto et al, Nature Commun. 8, 14165 (2017)

Directional excitation within molecular H₂O

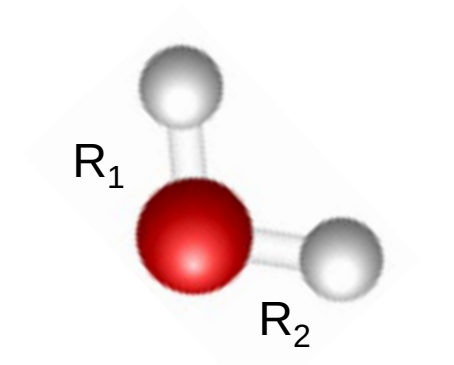
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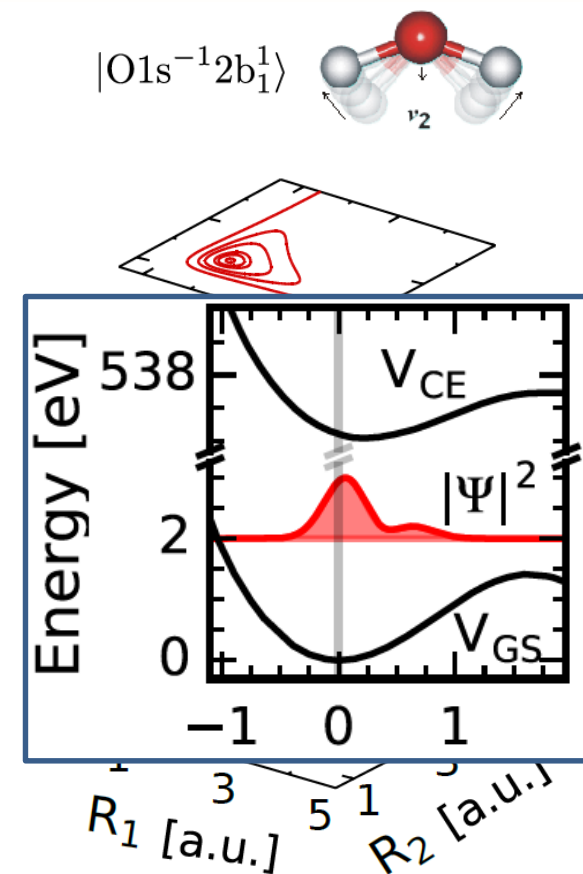
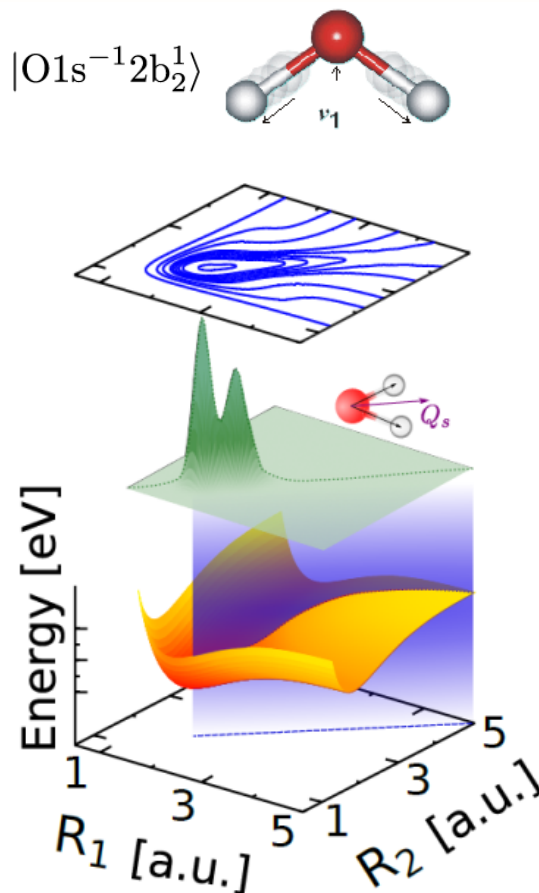
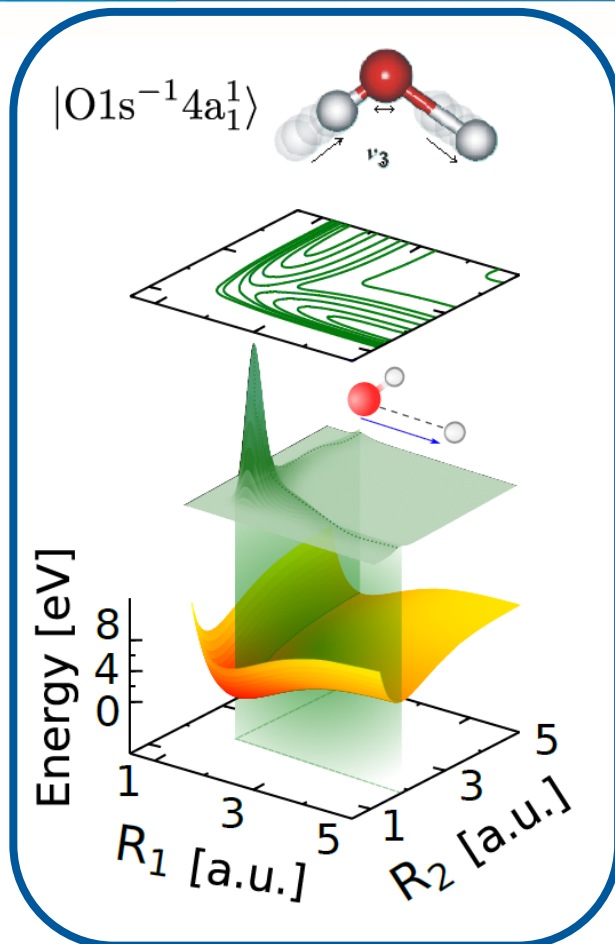
bend

stretch



R. Couto et al, Nature Commun. 8, 14165 (2017)

Directional potential cuts in isolated H₂O molecule



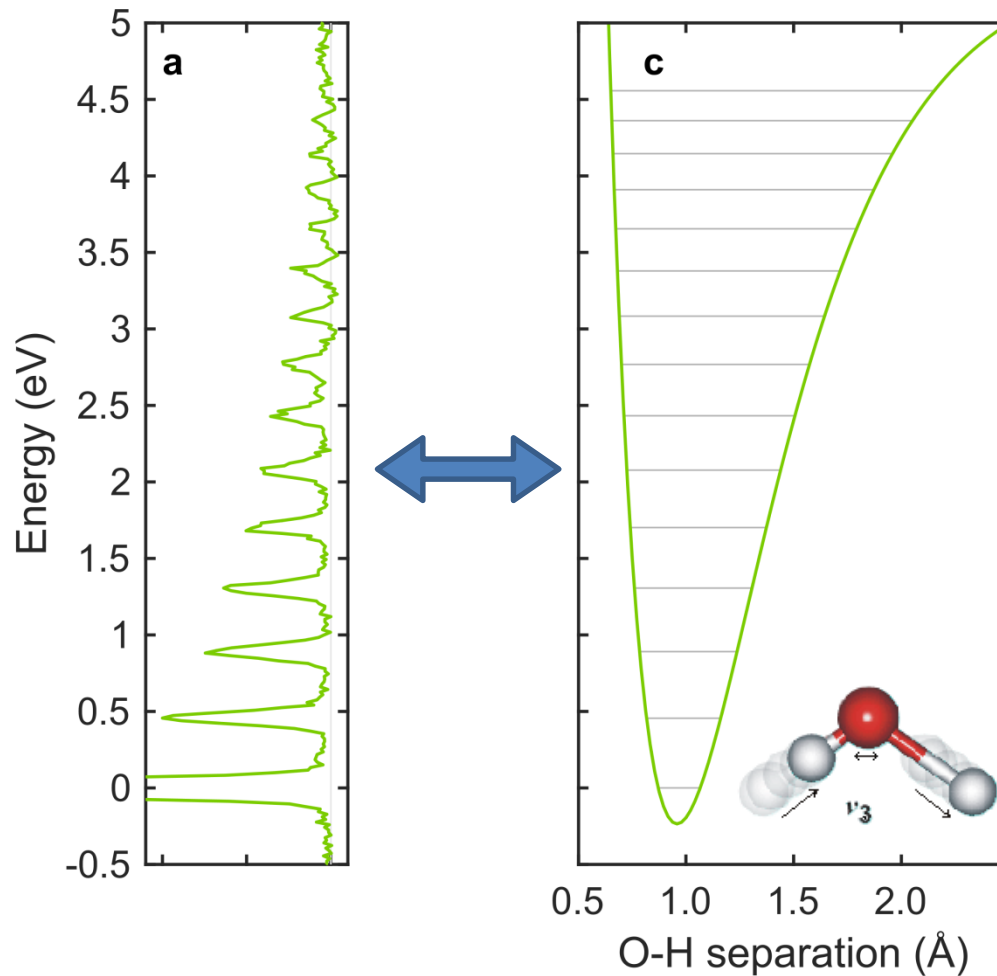
V. V. da Cruz et al, PCCP **30**, 19573 (2017), R. Couto et al, Nature Commun. **8**, 14165 (2017),
S. Eckert et al. Phys. Rev. A **97**, 053410 (2018)

3 different RIXS resonances – 3 directional cuts along the normal coordinates

How does the H₂O molecular potential change in the liquid?

Gas

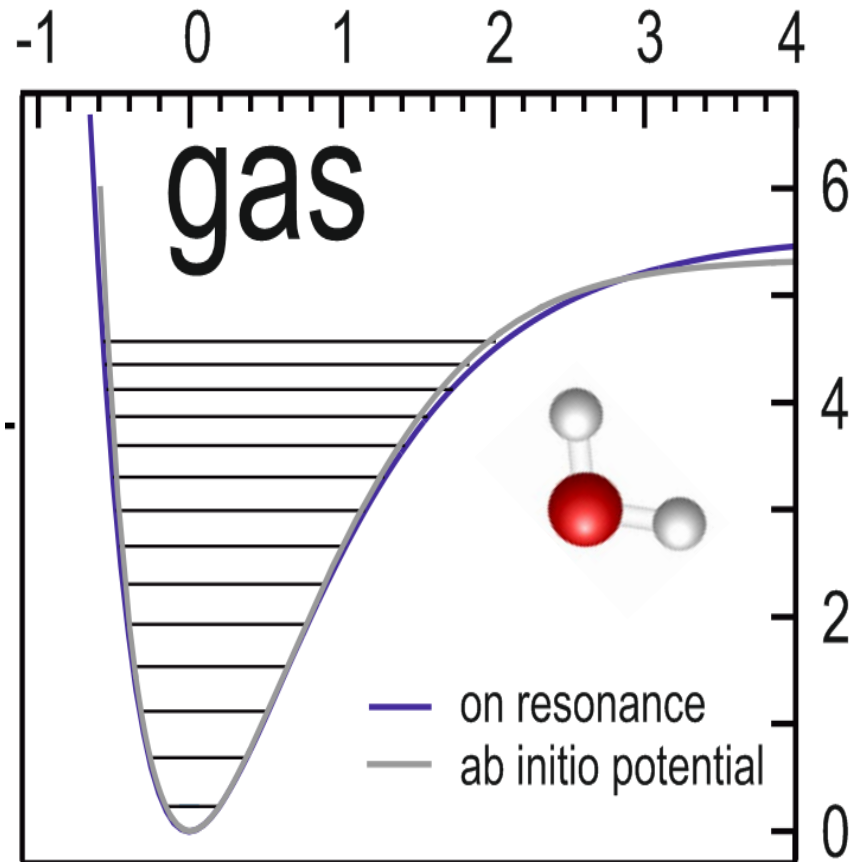
Liquid



single molecular potential

??

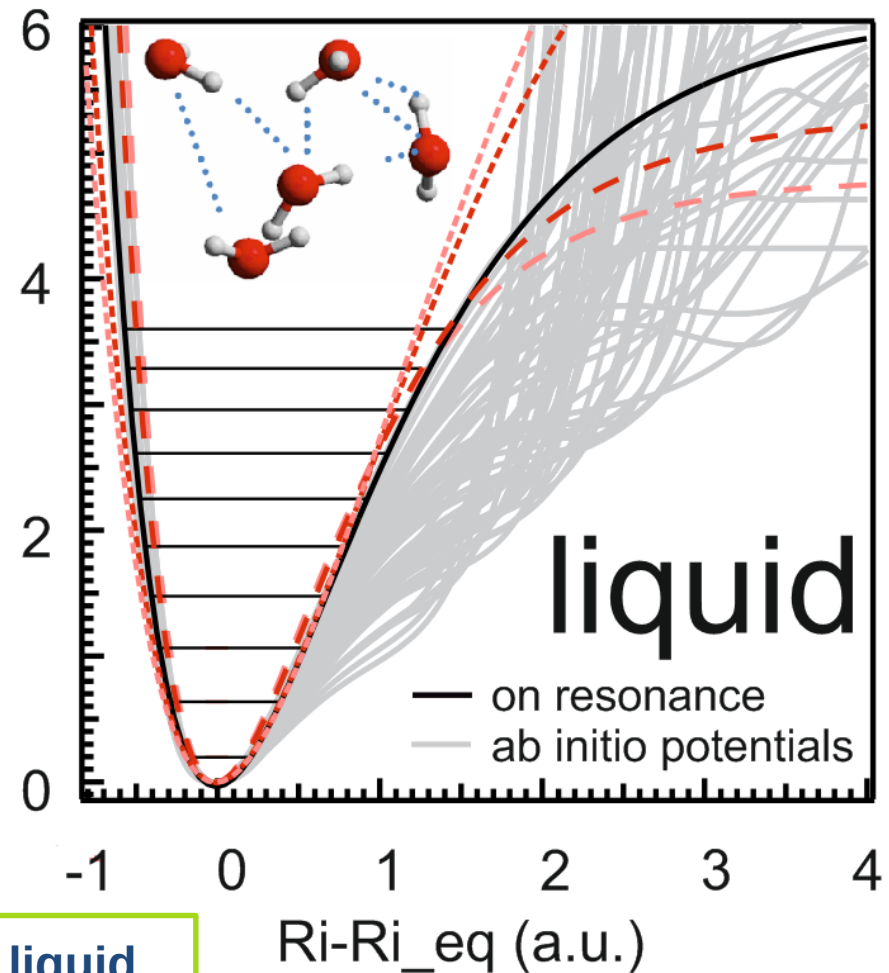
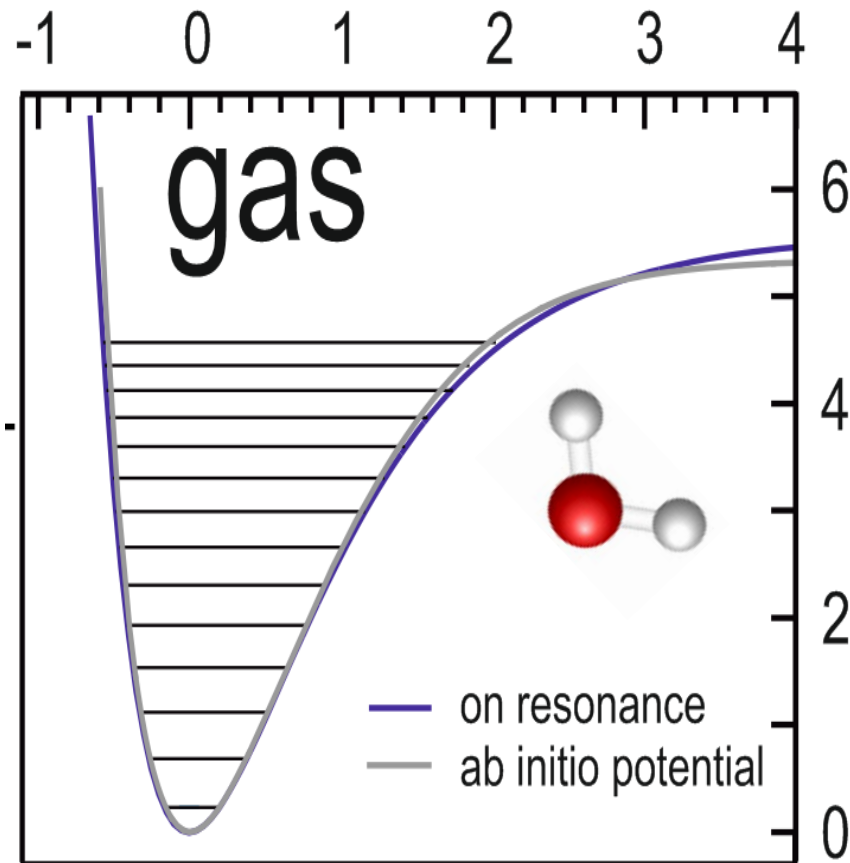
Compare to calculations



Reconstruct with Morse potential

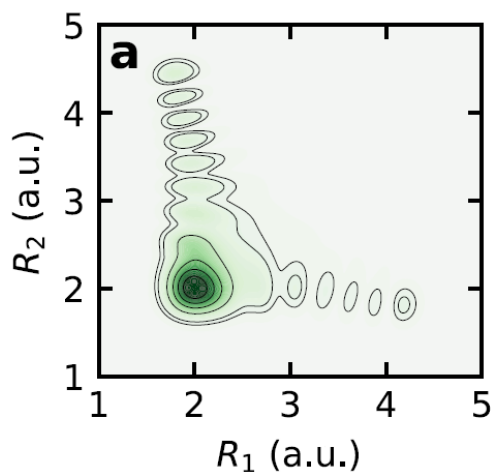
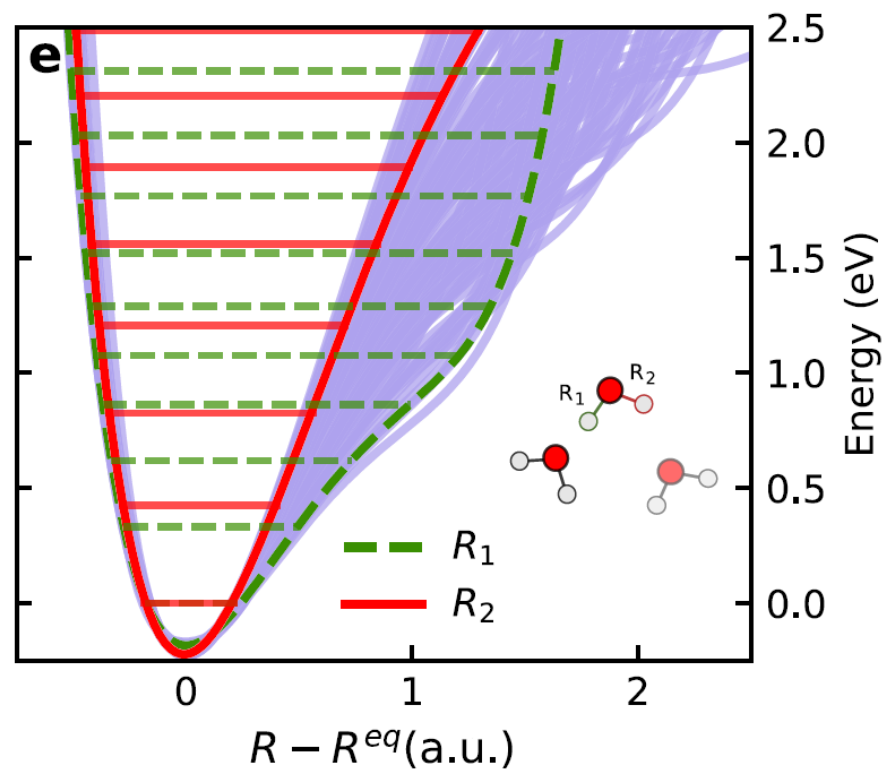
- ✓ Matches well
- ✓ Fits well to calculations

Compare to calculations



Morse reconstruction not working for liquid

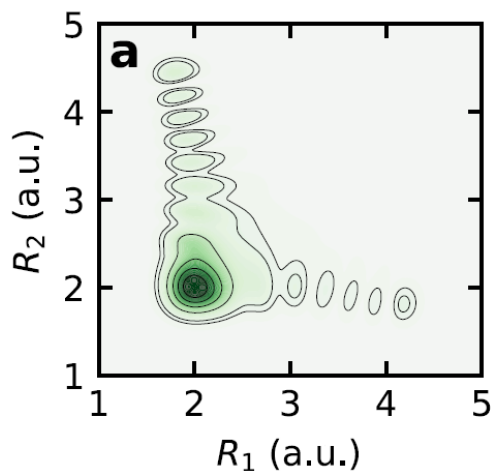
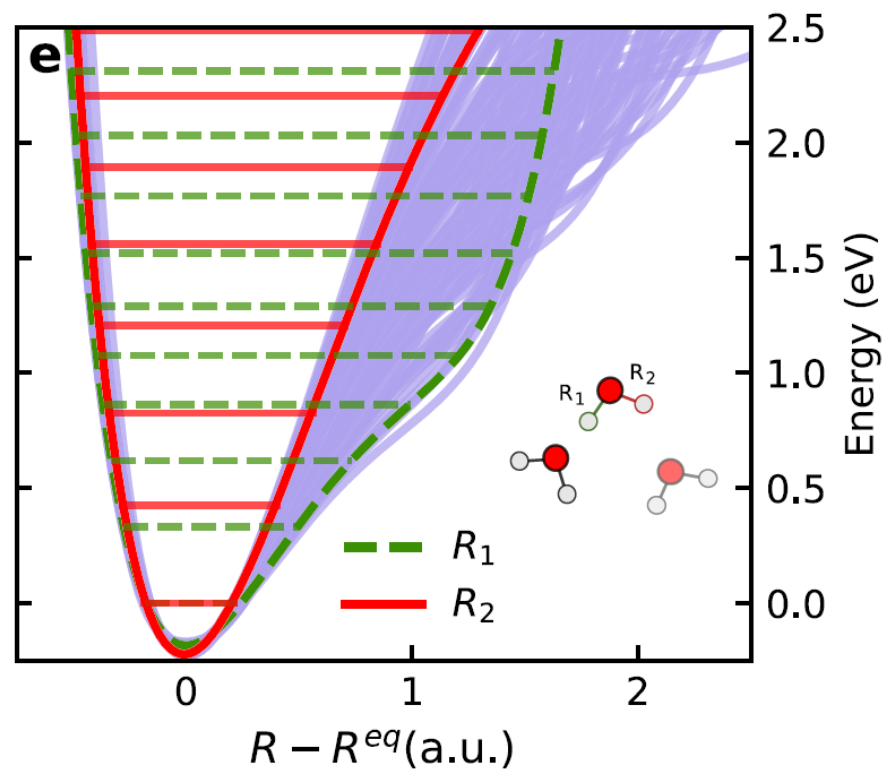
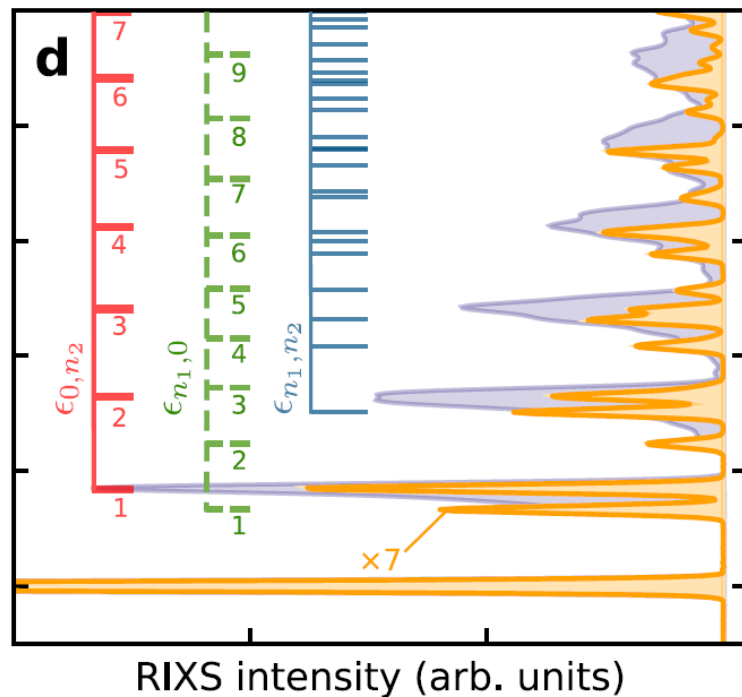
Excite both (different) OH bonds



Coherent excitation of both OH bonds

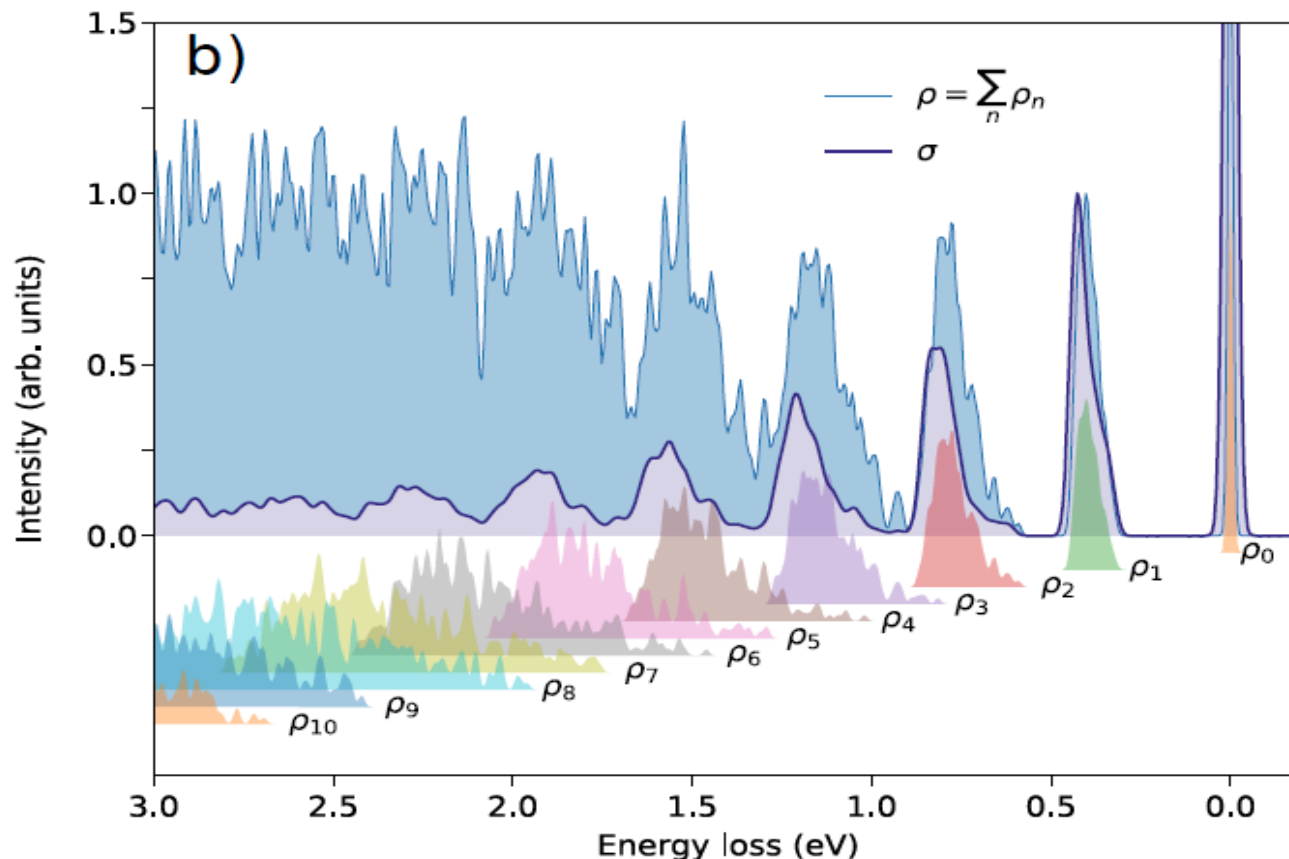
→ In liquid often very different bonds!

Excite both (different) OH bonds



Single molecule spectrum!

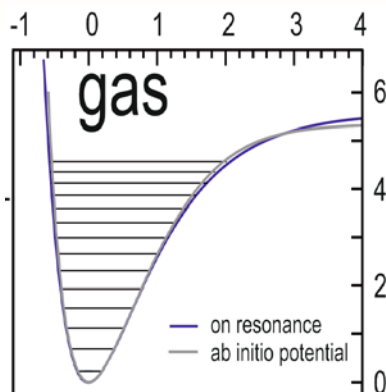
Many different molecules due to local environment (H-bond network)



Overlap of partial
density of states ρ

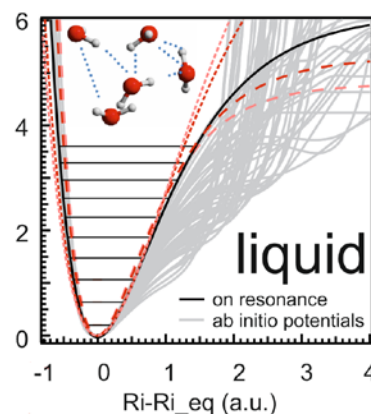
→ Breakdown 1-to-1 correspondence of eigenvalue to vibrational peak!

Liquid water is not straightforward!



Gas

- Isolated molecule
- Symmetric structure
- 3 normal vib modes
- 1 PES
- Morse reconstruction works



Liquid

- Surrounded by other molecules
- Assymetry due to local environment
- Coherent excitation of 2 very different OH bonds
- Set of PES
- Morse reconstruction gives steep limit
- Steep potential: weak H bond
- Shallow potential: strong H bond
- Breakdown 1-to-1 correspondence for eigenvalue to vib peak (width)

Sub natural linewidth RIXS

- ✓ Maps potential energy surfaces
- ✓ Cuts along interaction coordinates

In liquid water

- ✓ Coherent excitation into 2 different OH bonds
- ✓ Calculations give set of PES
- ✓ Morse reconstruction: limit of weak H-bond
- ✓ Startpoint for PES reconstruction where calculations too expensive

✓ Soon possible even at
METRIXS@BESSY!



Acknowledgements

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Dynamic Pathways in
Multidimensional Landscapes

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S. Eckert (former)

R. Jay

J. Niskanen (former)

V. Vaz da Cruz

M. Fondell

P. Miedema (former)

C. Weniger

S. Schreck (former)

A. Föhlisch



R. Couto (former)

N. Ignatova

S. Polyutov

V. Kimberg

F. Gel'mukhanov

PAUL SCHERRER INSTITUT



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IN MULTIDIMENSIONAL
LANDSCAPES**

ORGANIZERS

Prof. Dr. Alexander Föhlisch
Prof. Dr. Svante Svensson
Prof. Dr. Nils Mårtensson

LOCATION

Magnus-Haus
Am Kupfergraben 7
Berlin-Mitte, Germany

CONTACT

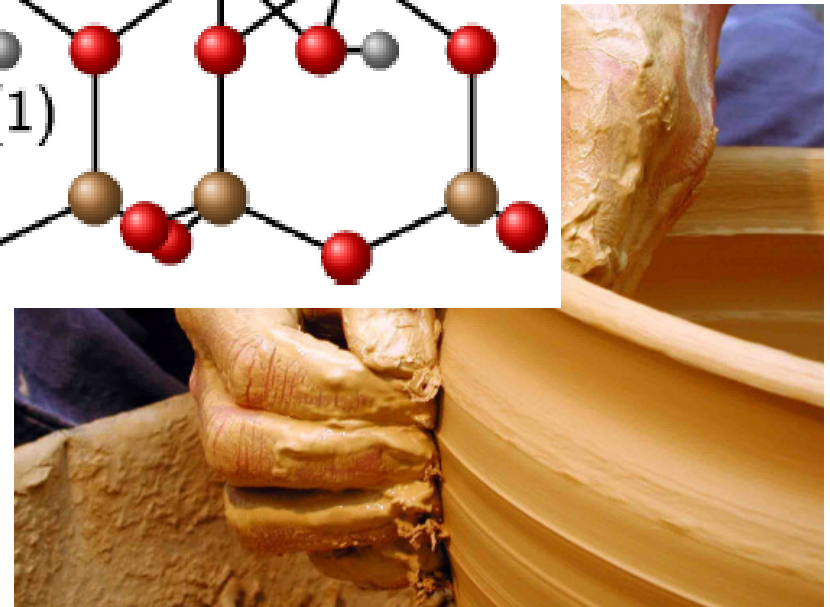
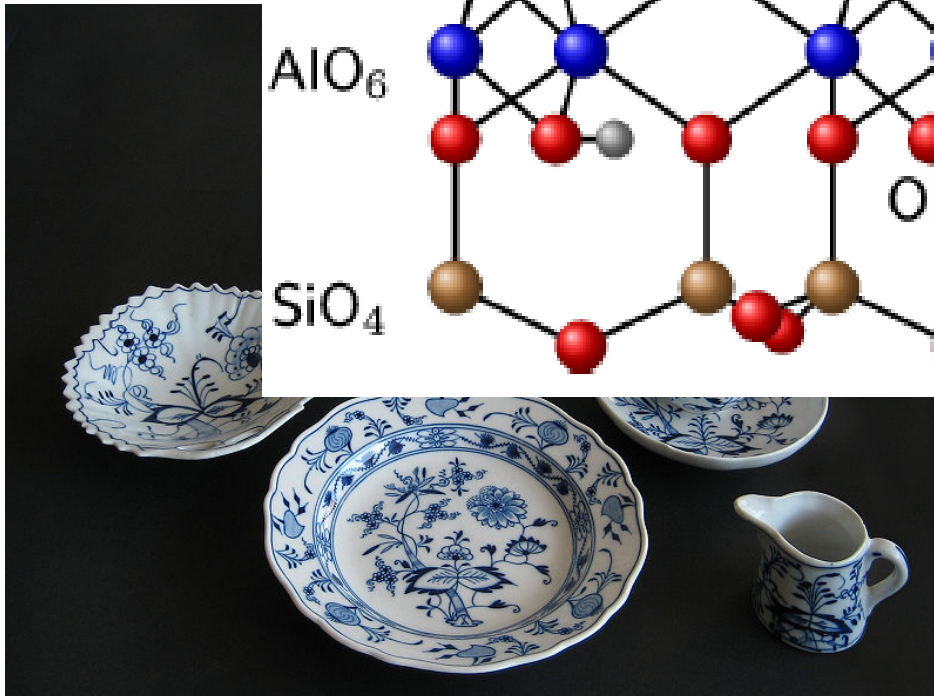
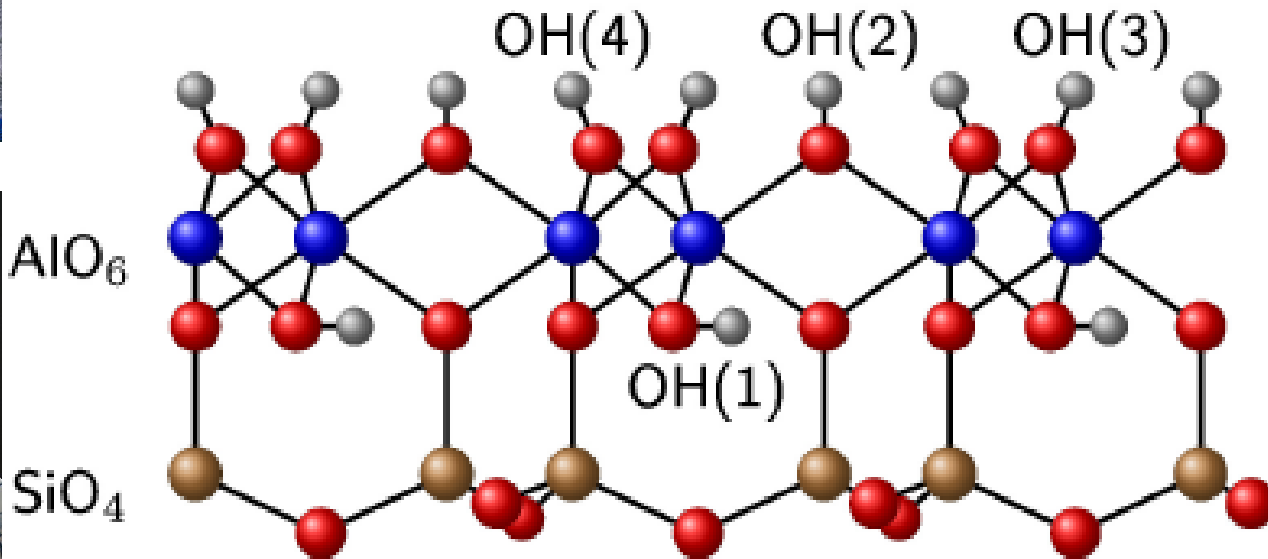
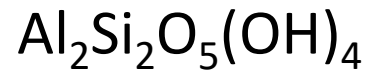
grunewald@helmholtz-berlin.de
hz-b.de/SXR2019



One center in a complex material?



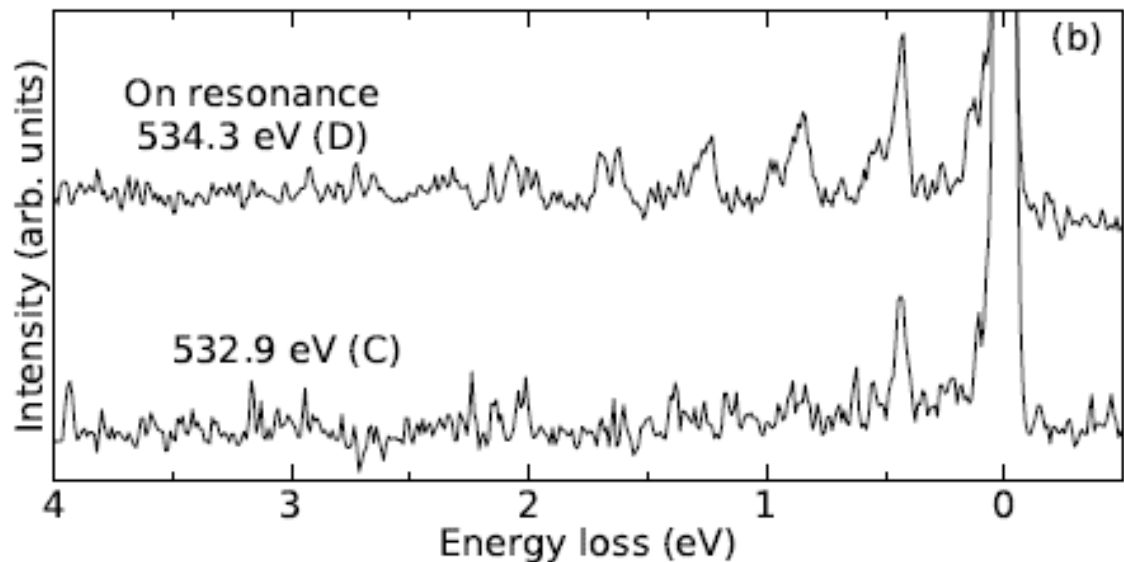
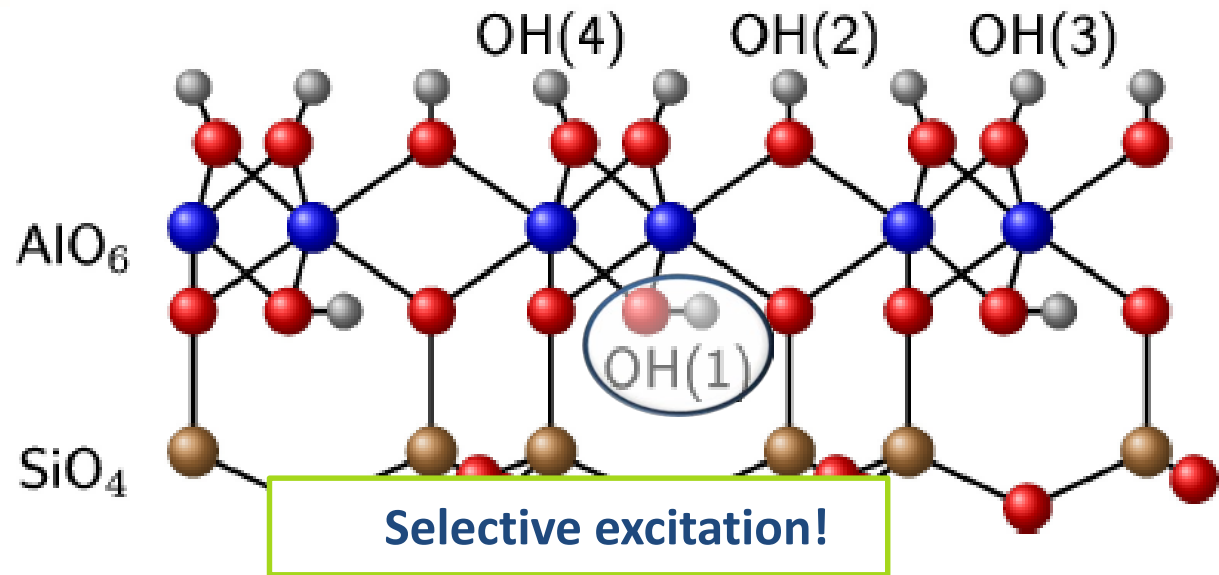
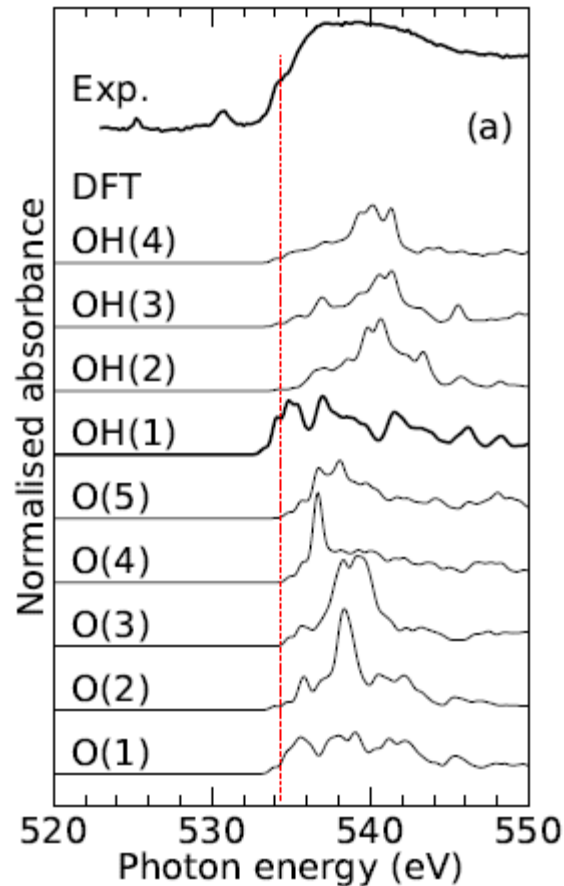
Kaolinite clay



Complex systems: O-H in Kaolinite

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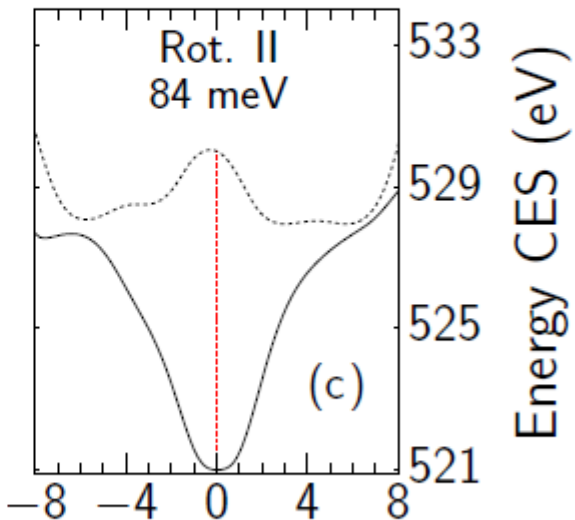
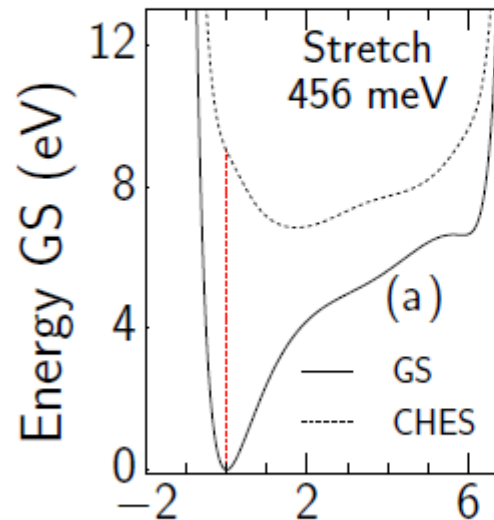
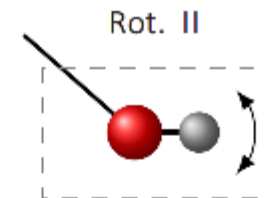
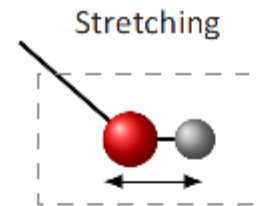
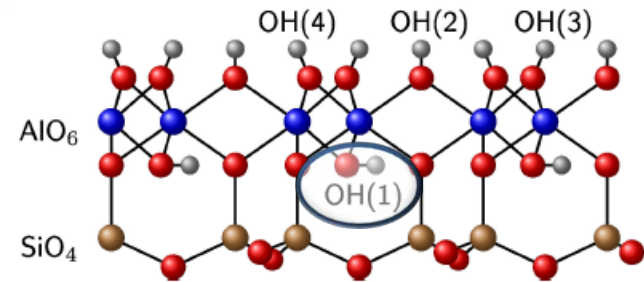
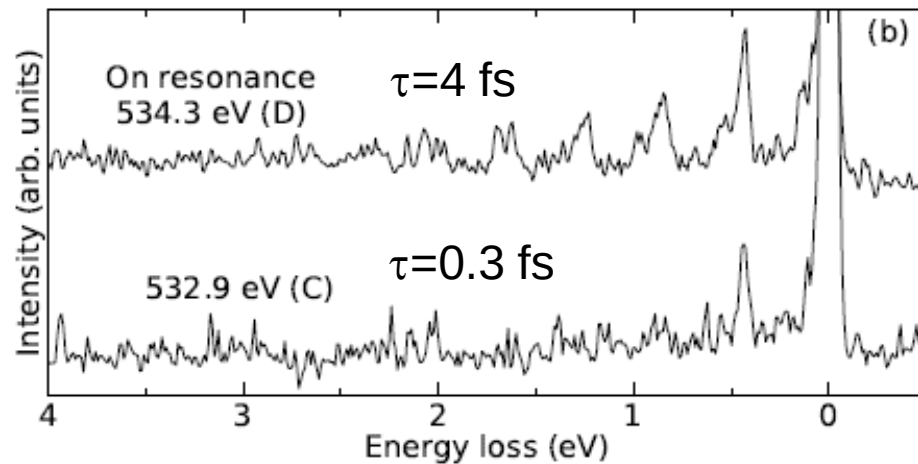
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Complex systems: O-H in Kaolinite

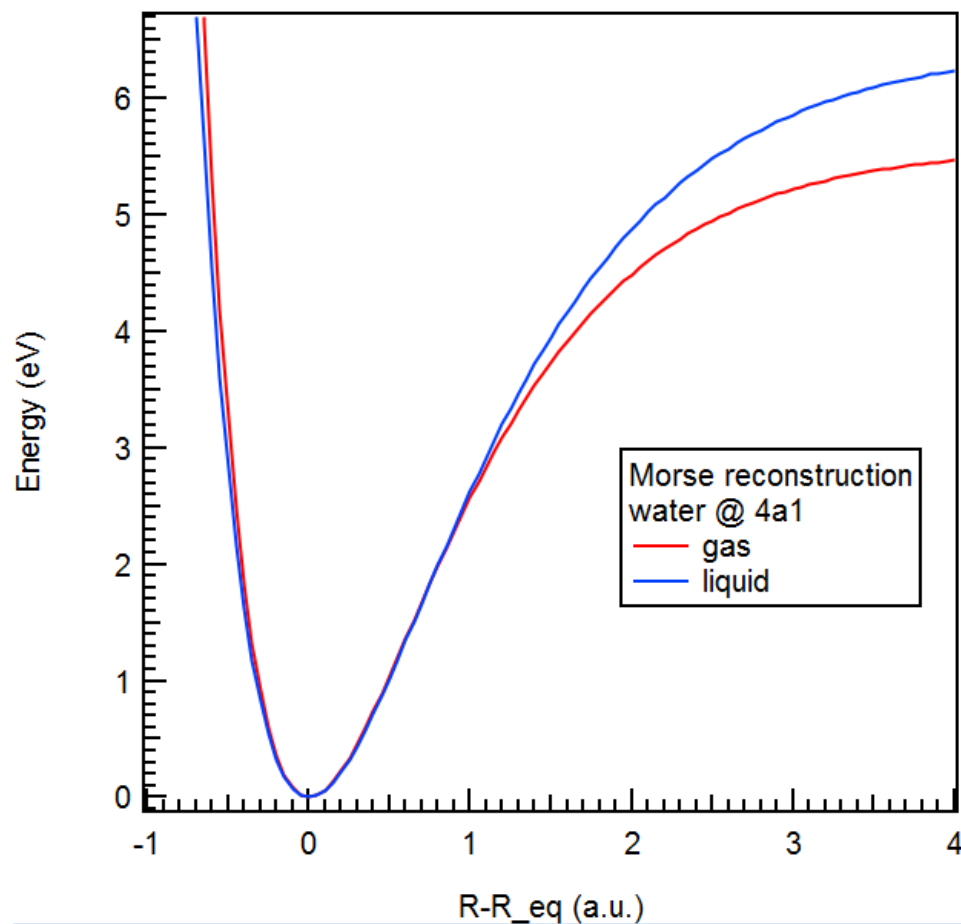
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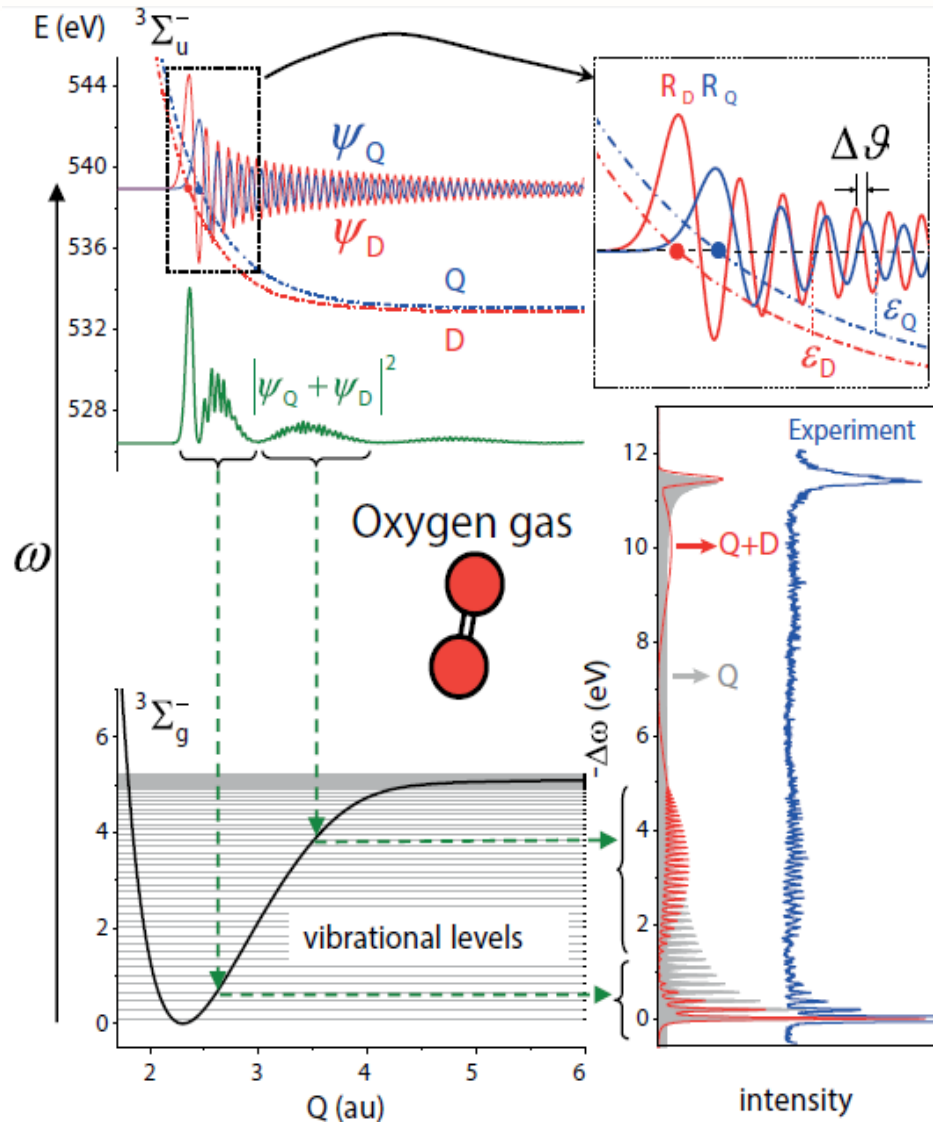
Distortion along normal mode (a.u)

Elongated bond induces rotation
→ Effect of environment



Liquid water PES
Steeper than gas

Core excited state dynamics ?



Pietzsch et. al., PRL 106, 153004 (2011)

RIXS Inherent temporal component

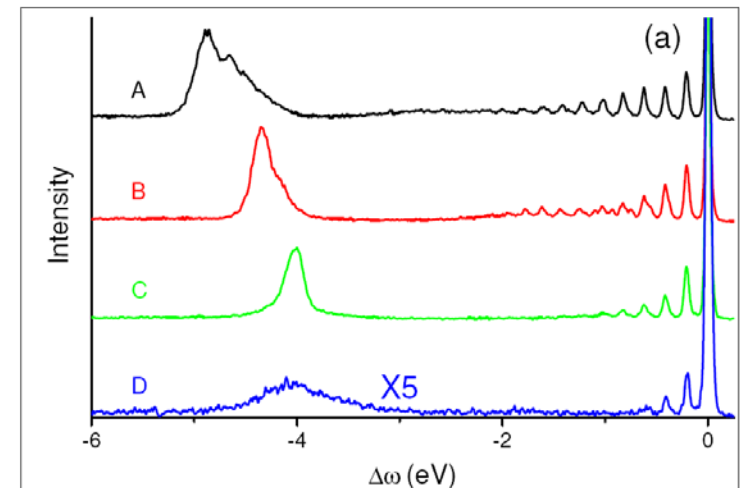
Scattering duration time:

$$\tau^2 = 1 / (\Omega^2 + \Gamma^2)$$

• Detuning

$$\Omega = \omega - \omega_{\text{res}}$$

→ Shortens scattering duration time



Liquid acetone O K-edge

Y.P.Sun et al., Phys. Rev. B **84**, 132202 (2011)

Where does it lead to?

? resolution

Interaction with surrounding
→ additional dipolar broadening γ_s :

$$\gamma_s \propto \Delta\mu_f \cdot \mu_s$$

(dipole moments of solute and solvent)

$$\Delta\mu_f = \mu_{final} - \mu_0$$

In sub natural linewidth RIXS:

Final state = ground state

$$\Delta\mu_f=0 \rightarrow \gamma_s=0$$

Set by

- ✓ x-ray optics
- ✓ Source properties

