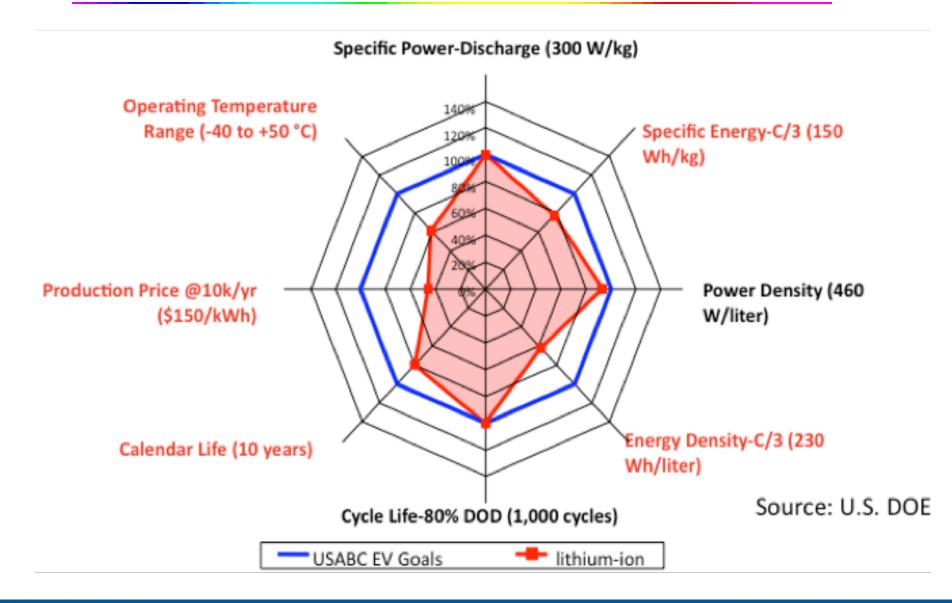


# Tinkering with Batteries: a multidimensional Problem!""



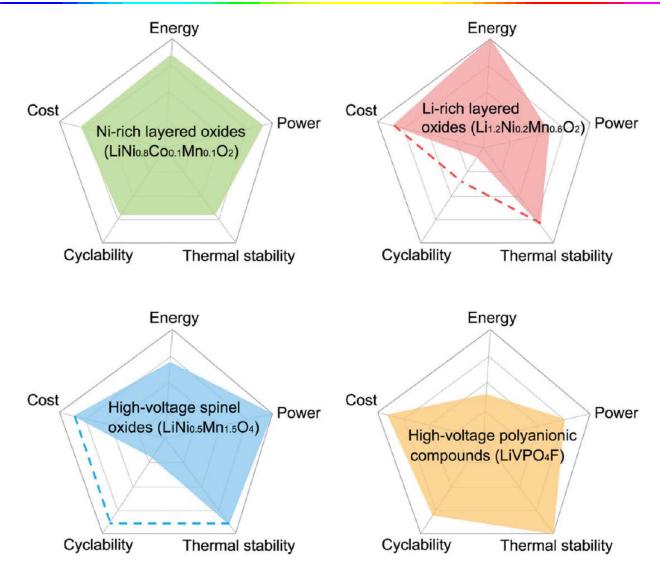






# e.g., "Co-free" electrode? – Sorry, NO good solution 🕾





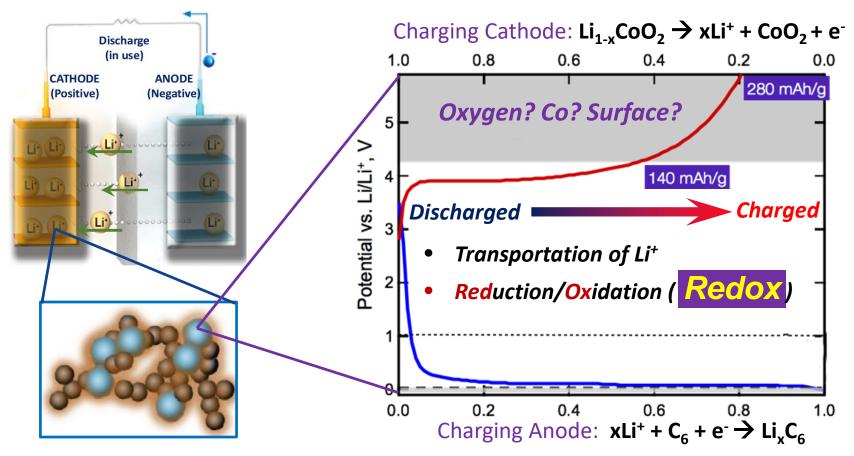
(Only cathodes shown here) Li et al., Chem. Soc. Rev. 46, 3006 (2017)





#### What kind of information matters for (practical) battery materials?





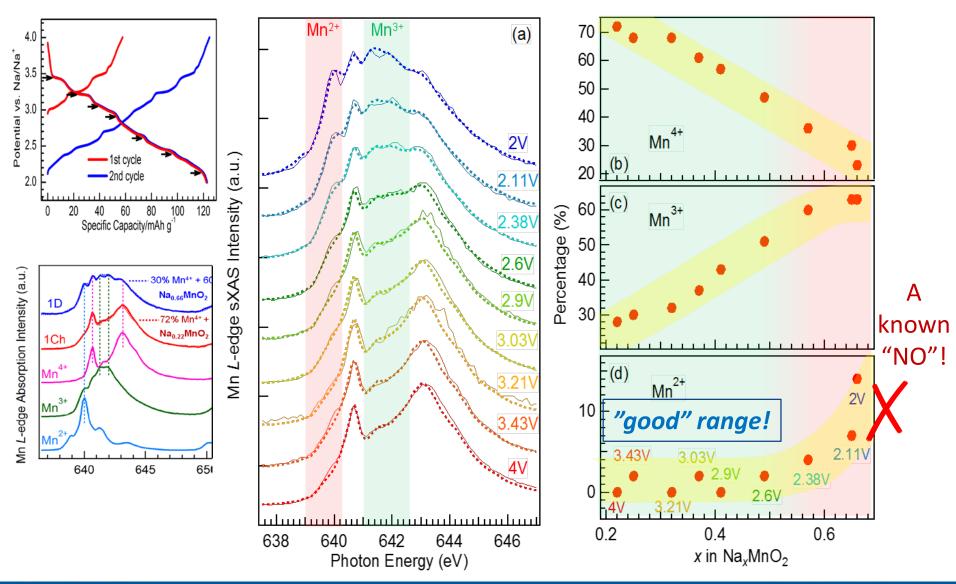
- Redox reactions in BULK electrodes
- Chemical activities on SURFACE / interface
  - Solvation shells in LIDUID (electrolyte)





# $Na_{0.44+x}MnO_2$ : Mn states upon cycling and surface $Mn^{2+}$ evolution

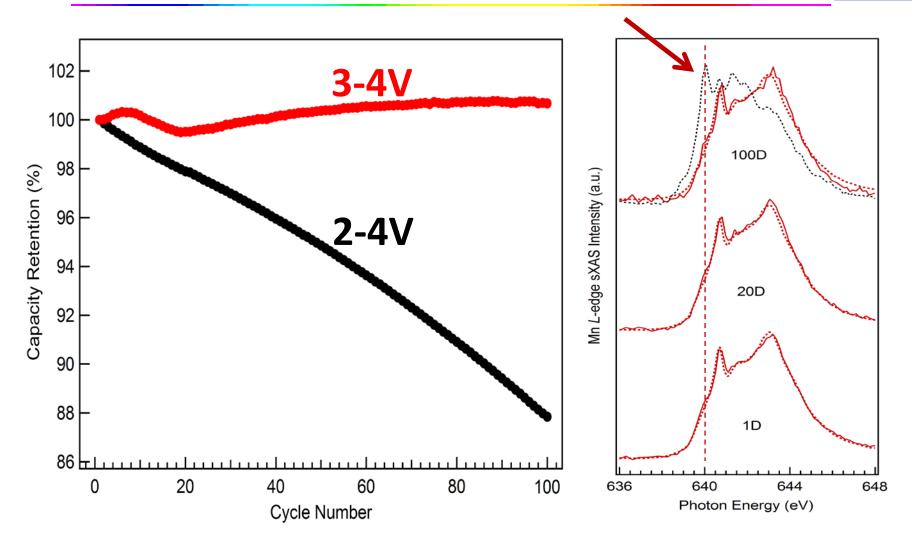








# A rational approach to improve $Na_{0.44+x}MnO_2$ cyclability!



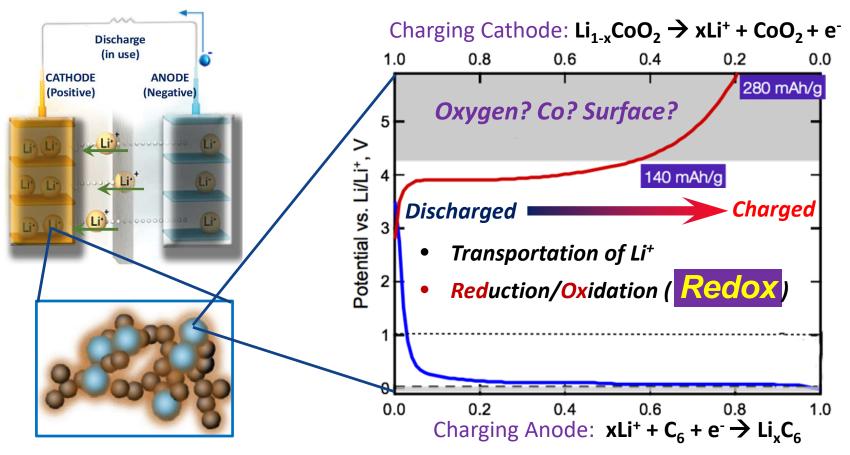
Improved Cycling Stability from Multiple effects (shallow cycling)!
- However, *one of the keys* is the suppression of *surface Mn*<sup>2+</sup> evolution





#### What kind of information matters for (practical) battery materials?





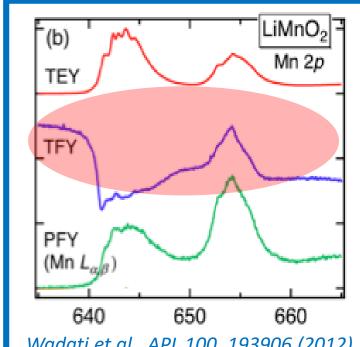
- Redox reactions in BULK electrodes
- Chemical activities on SURFACE / interface
  - Solvation shells in LIDUID (electrolyte)





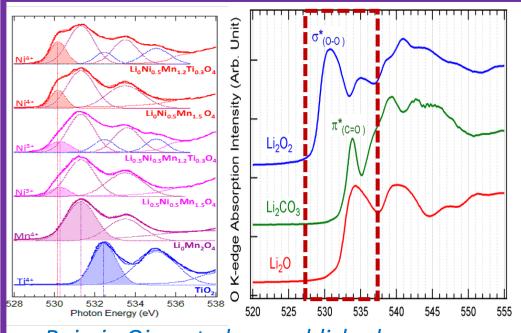
# Two problems of "bulk-sensitive" XAS of oxide electrodes

- Distortions of TM-L in photon-in-photon-out modes!
- Lack of chemical sensitivity for O-K!!



Wadati et al., APL 100, 193906 (2012)

- Material dependent
- Element dependent
- "self-absorption correct" -N/A

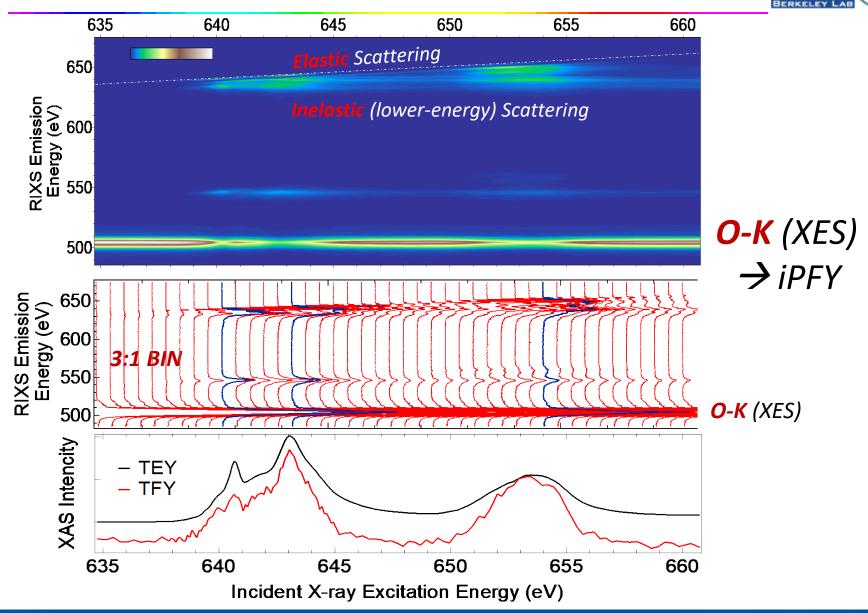


Ruimin Qiao et al., unpublished

- Strong 3d character in O-K "pre-edge"
- Chemical sensitivity lost due to the overlapping energy range ☺





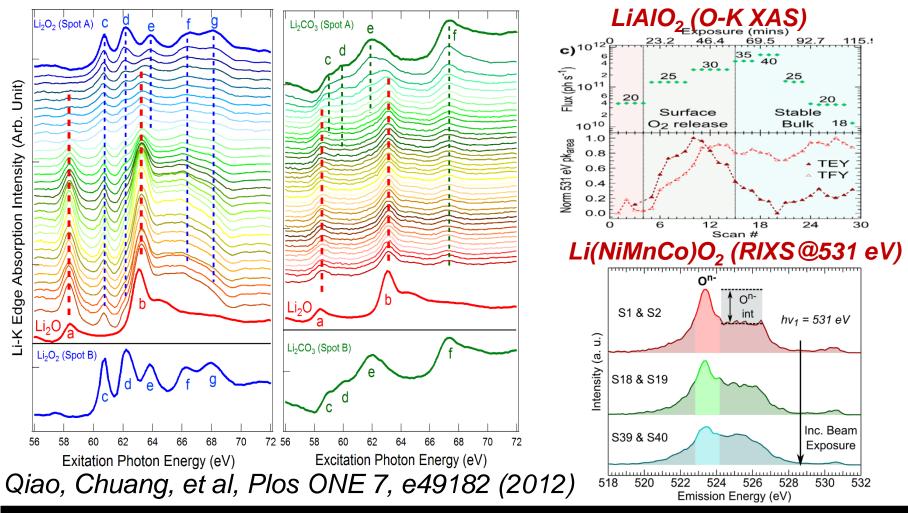






# The real challenge!





**Detection efficiency** is NOT a speed/quality issue; it is a **FEASIBILITY** issue for chemistry and material sciences!



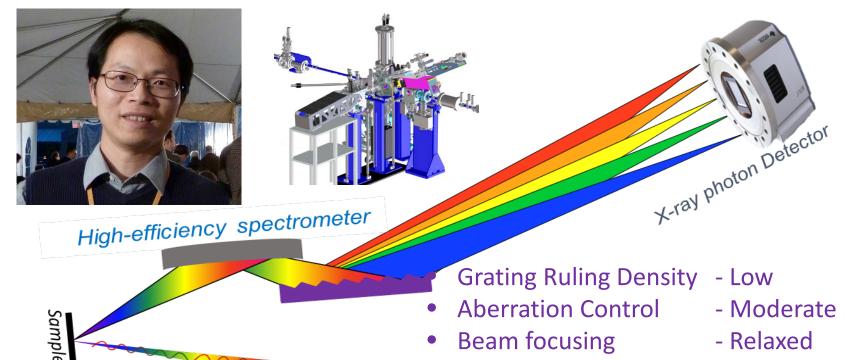




### A dilemma of resolution vs efficiency for Energy materials



## Qiao, et al., RSI 80, 063102 (2017) Chuang, et al., RSI 88, 013110 (2017)



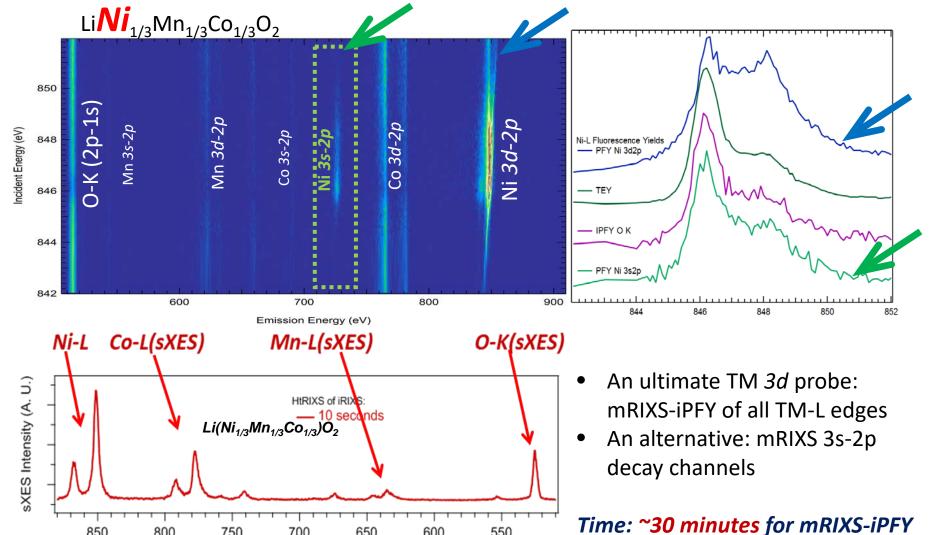
Of course, the true reason is: poor & out of real-estate!

Detector Energy Window - Huge

Angles, etc... ...

## 2016: full-energy range ultra-high efficiency mRIXS





A **10-second** spectrum of a (discharged) NMC electrode

650

700

X-ray Emission Photon Energy (eV)







850

800

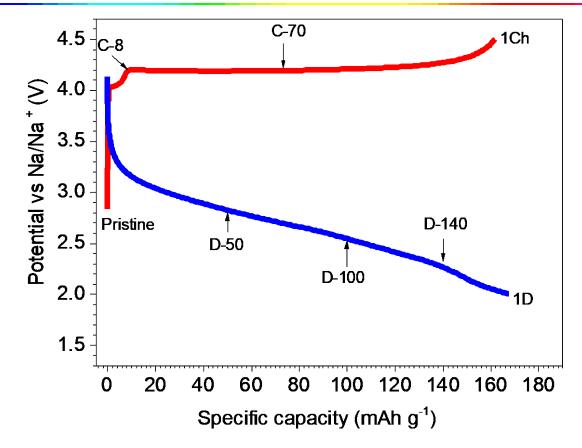
750

550

600

# $Na_{2/3}Mg_{1/3}Mn_{2/3}O_2$ : Complex states of evolving Mn & O



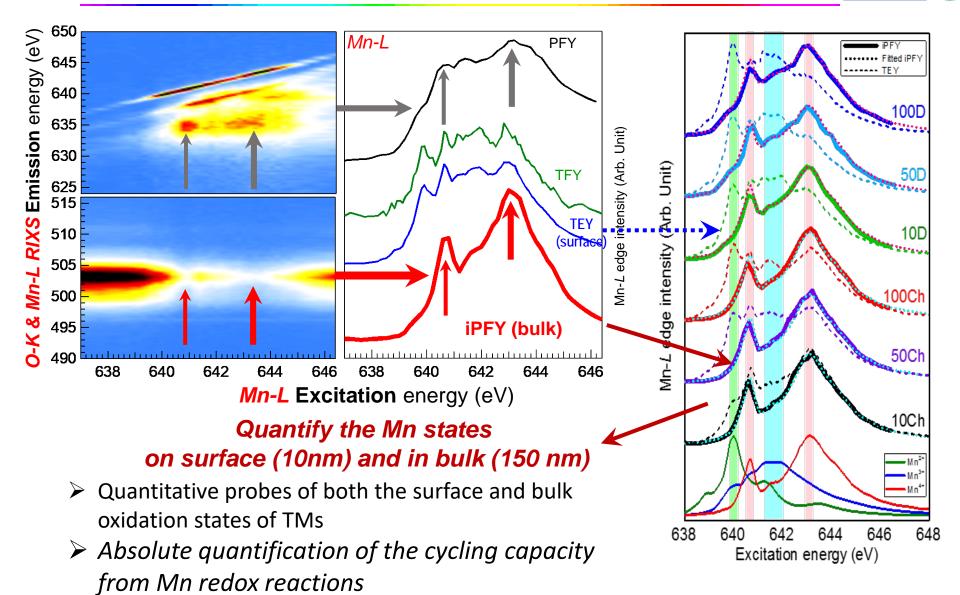


- Stoichiometry adjusted to drive (pure) high-valence Mn<sup>4+</sup>!
- Capacity reaches record high with "inactive" Mn?!
- Change of *Oxygen* states is expected to be strong?





### mRIXS-iPFY & TEY Mn-L: bulk and surface Mn redox quantification



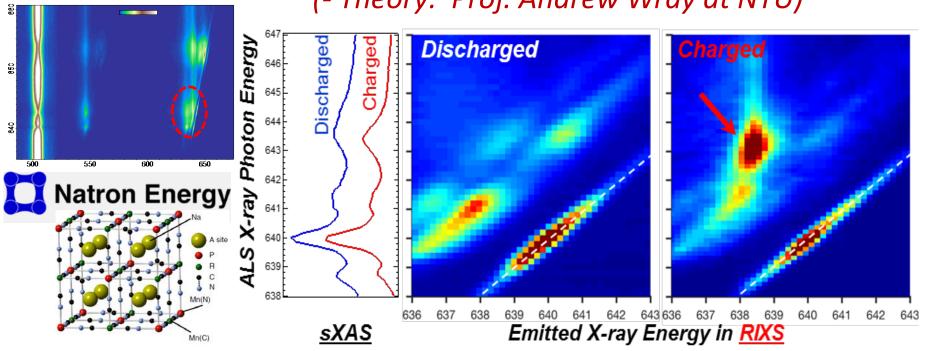




## Mn-L mRIXS reveals Mn<sup>1+</sup>: a 90-year speculation!



(- Theory: Prof. Andrew Wray at NYU)



In a recent breakthrough, Berkeley Lab scientists collaborated with Natron Energy and New York University to confirm a century-old chemistry speculation, a finding with broad-reaching implications for the future of battery technology. The researchers took advantage of two Berkeley Lab user facilities, the Advanced Light Source and the Molecular Foundry, to study an unconventional, but promising, new sodium-based battery design. They discovered a key to the battery's superlative properties was a novel chemical state of the element manganese. The revelation could lead to new classes of high-performance, low-cost batteries that can quickly and efficiently store and distribute energy produced by solar panels and wind turbines across the electrical grid.



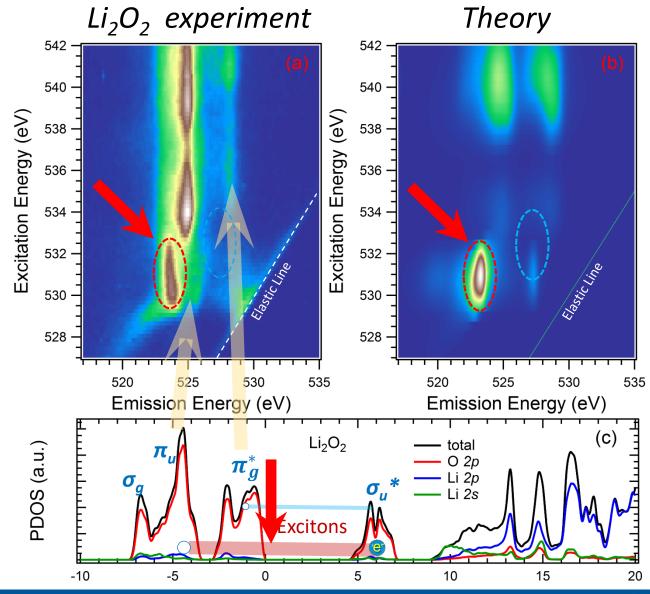
- Dr. Maxon's congressional testimony for U.S. National Labs (2018)





## Fingerprint non-divalent oxygen states through mRIXS



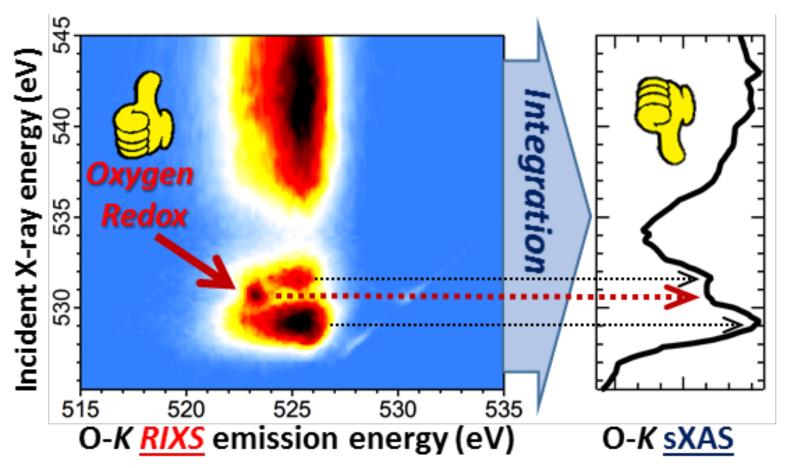






### mRIXS: a reliable probe of unusual bulk Oxygen states



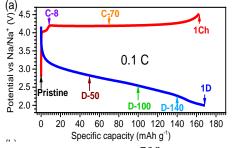


- TM-O hybridization is dissociated from oxygen redox reactions
- A full energy range mRIXS is necessary (signals buried in XAS)
- Theoretical calculations in process ...



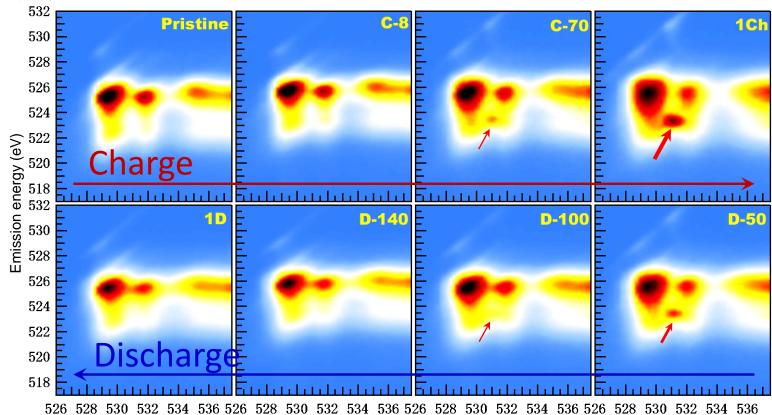






### $Na_{2/3}Mg_{1/3}Mn_{2/3}O_2$

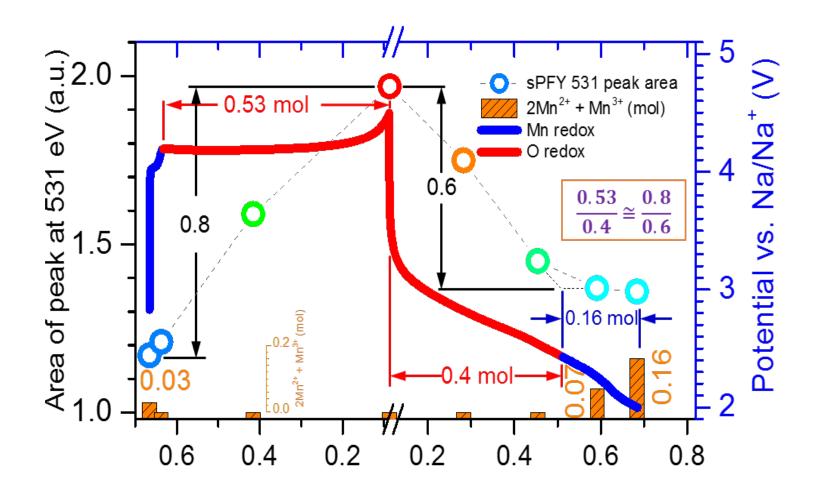
- O-K mRIXS feature follows tightly with electrochemical cycling!
- O-K mRIXS feature fingerprints the oxygen redox quantitatively!



Excitation energy (eV)







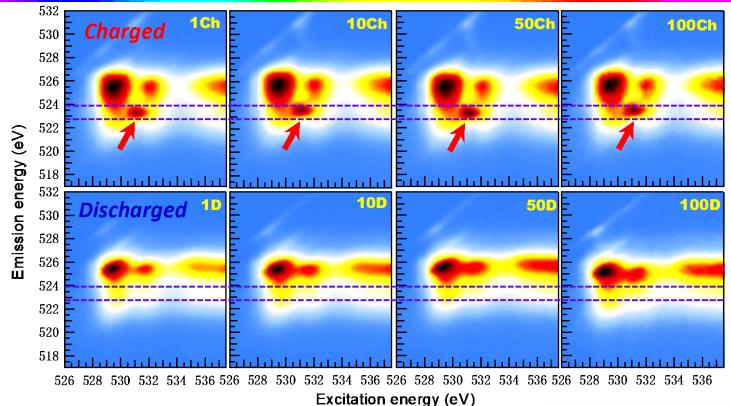
✓ mRIXS quantifications precisely match the electrochemical capacity/profile!





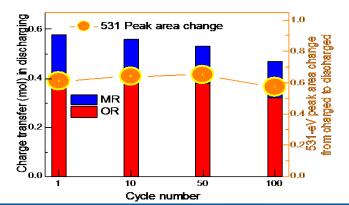
## Quantify the reversibility of Oxygen redox through mRIXS





- A strong O.R. system could also be highly reversible!
- Assessment of the reversibility of Oxygen redox reactions upon extended cycles!

**NOTE:** NOT all systems show such a perfect match between mRIXS quantification and electrochemistry

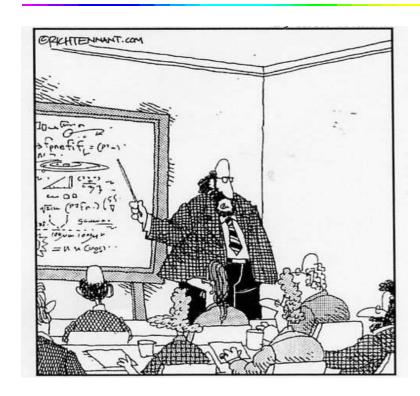






# Always #1 question: what matters for batteries?





"After the discovery of "antimatter" and "dark Matter", we have just confirmed the existence of "doesn't matter", which does not have any influence whatsoever!"

- the 5<sup>th</sup> wave

#### **Conclusions**

- High-efficiency mRIXS is truly needed and provides unique information on battery material performance, especially the novel chemical states of TMs & O.
- Instrumentational and especially theoretical developments are still needed.
- Improvements on <u>detection efficiency</u> has opened RIXS technique as almost the only "tool-of-choice" for a wide range of topics in energy materials!



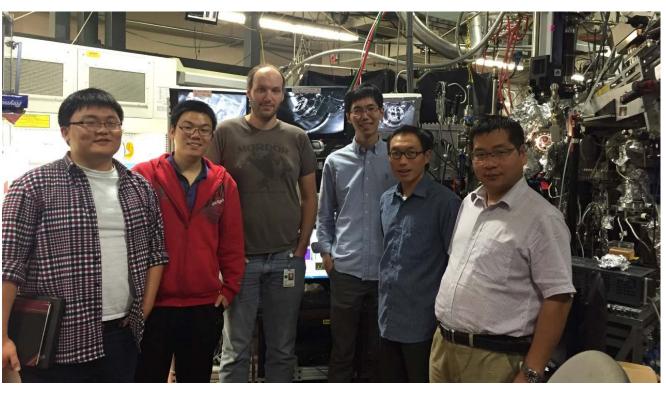


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- Zahid Hussain



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- ❖ Andrew Wray (NYU)
- Tom Devereaux (Stanford)

#### Industry:

Colin D. Wessells (Natron Energy)

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- **❖** Gao Liu (LBNL)
- ❖ Ning Li, Wei Tong (LBNL)
- Yuhao Lu (previously, Sharp USA)
- & many many others ...









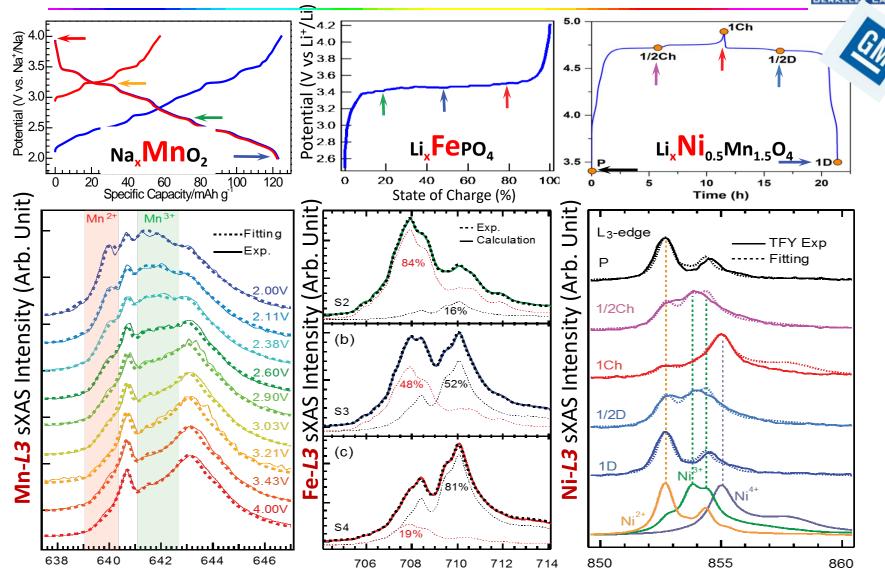
# **BACKUP SLIDES**





### Quantitative analysis of various TM-redox in battery cathodes





JACS 134, 13708 (2012) JPCC 119, 27228 (2015) Nano Energy 16, 186 (2015)



# 5 different XAS channels through full mRIXS



