

# **Powder Diffraction and Catalysis at NSLS and NSLS-II**

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Beam Line X7B

# Catalysis and NSLS II

Synchrotron radiation facilities provide unique and powerful tools for characterizing **the temporal and spatial evolution of working catalysts**. The properties of catalysts can be studied using a wide range of x-ray techniques, such as **x-ray powder** and/or single-crystal diffraction, small-angle x-ray scattering, and many x-ray spectroscopy methods. The high brightness and flux of NSLS-II light is for these techniques to be applied with the high spatial, energy, and time resolution necessary to fully characterize these complex catalytic systems.

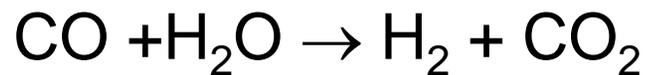
# Reason for talk

- Promote *in situ* powder diffraction studies of Catalysts at new beam line.
  - Sample studies at NSLS-I
  - Sample studies from other synchrotrons
  - Proposed studies for NSLS II
- Sample cells and sample positioning questions

# Overall Approach

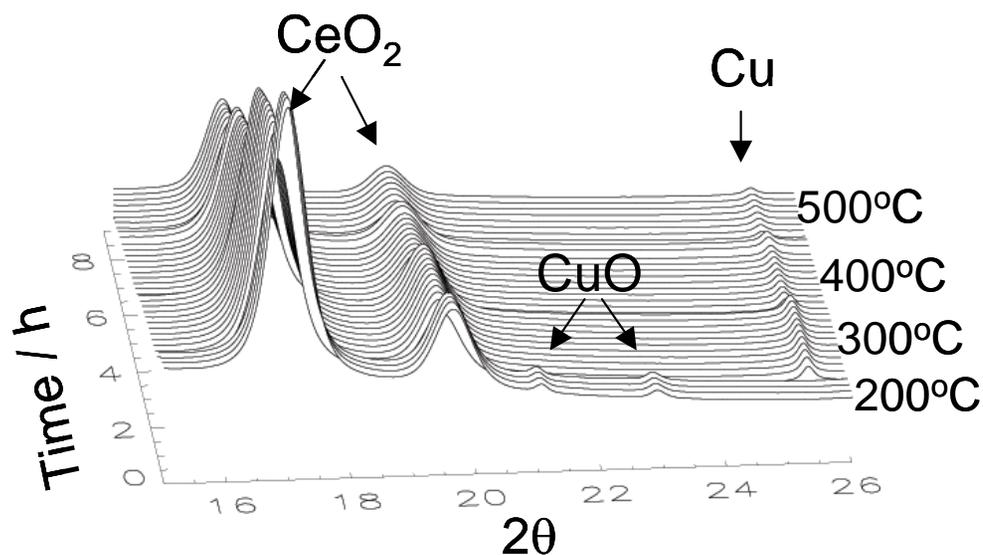
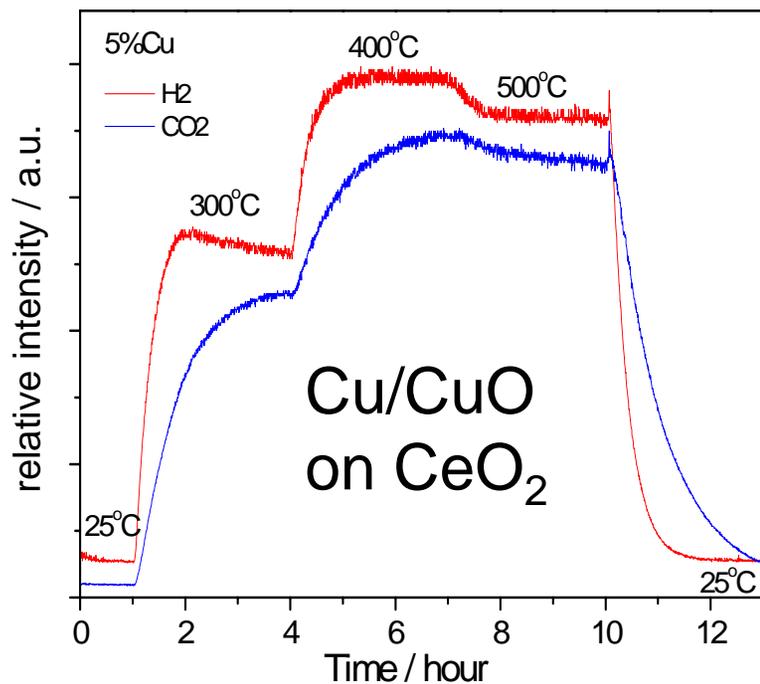
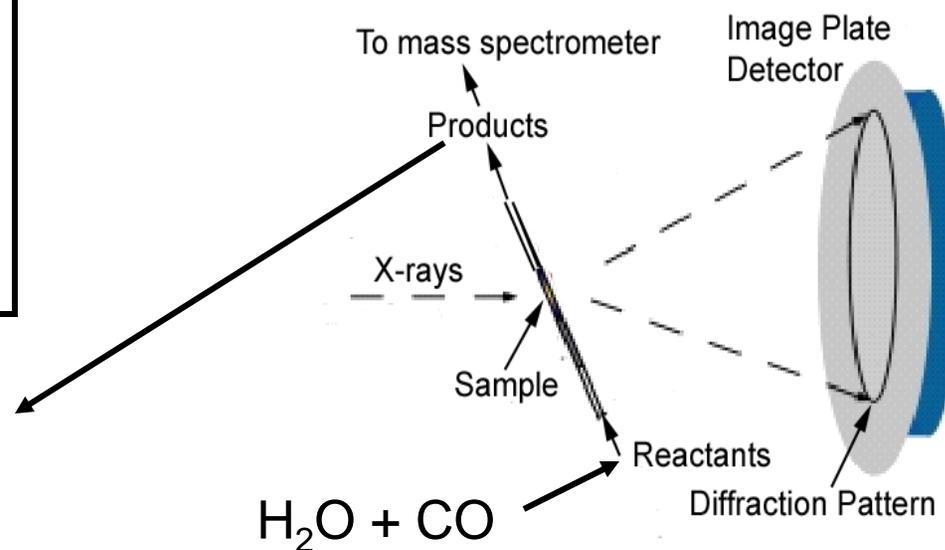
- Observe structure of active catalysts under “operando” conditions
- Observe structure changes during catalyst synthesis
- Methods
  - Powder diffraction
    - Crystalline (Profile refinement)
    - Amorphous (Pair Distribution Function)
  - Extended X-ray absorption fine structure

# *In situ* Time-resolved XRD Schematic



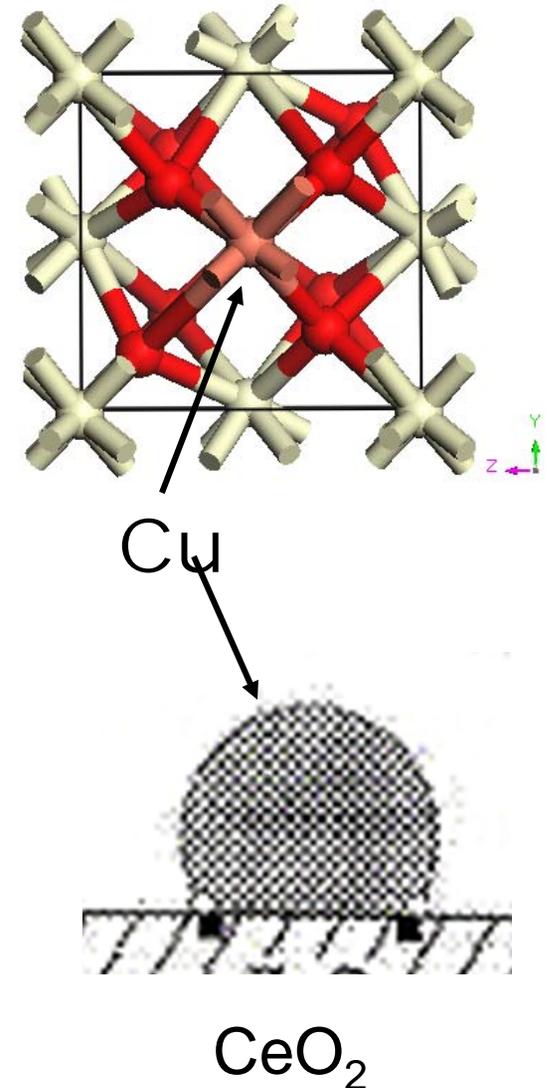
Reactants                  Products

Water Gas Shift Reaction

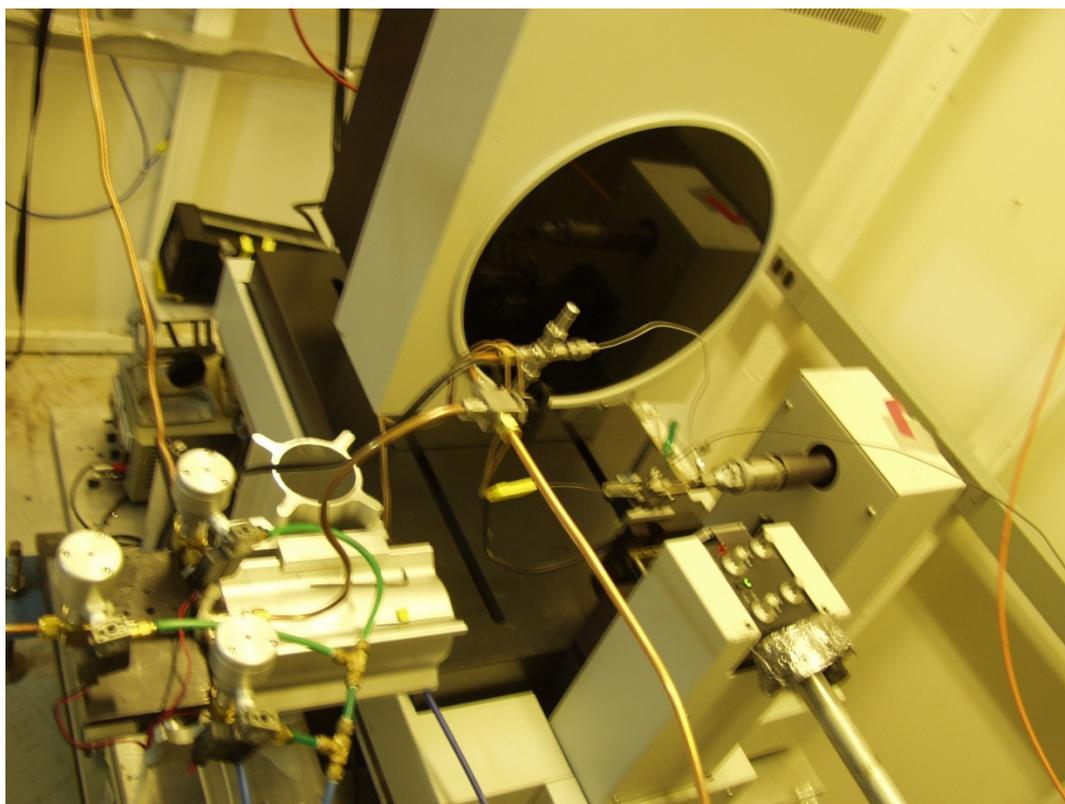


# New Low Temperature WGS Catalysts

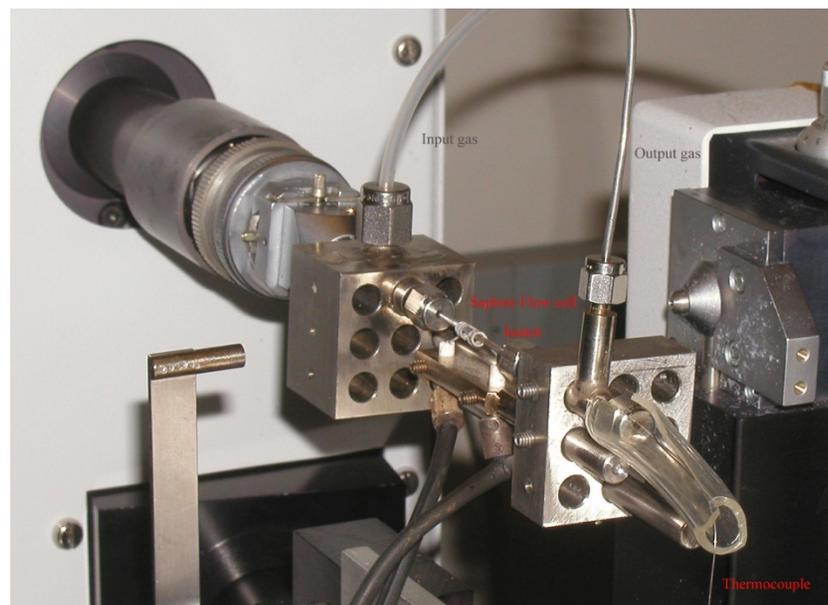
- Cu/CeO<sub>2</sub>:
  - doped: Cu on Ce sites (Cu migrates in and out of ceria structure)
  - impregnated: Cu/CuO on the CeO<sub>2</sub> surface
- Issues to understand mechanism
  - Oxidation state of the Cu during WGS
  - Location of metal on/in metal oxide
  - Vacancies in CeO<sub>2</sub>



# X7B *In Situ* Diffraction Apparatus

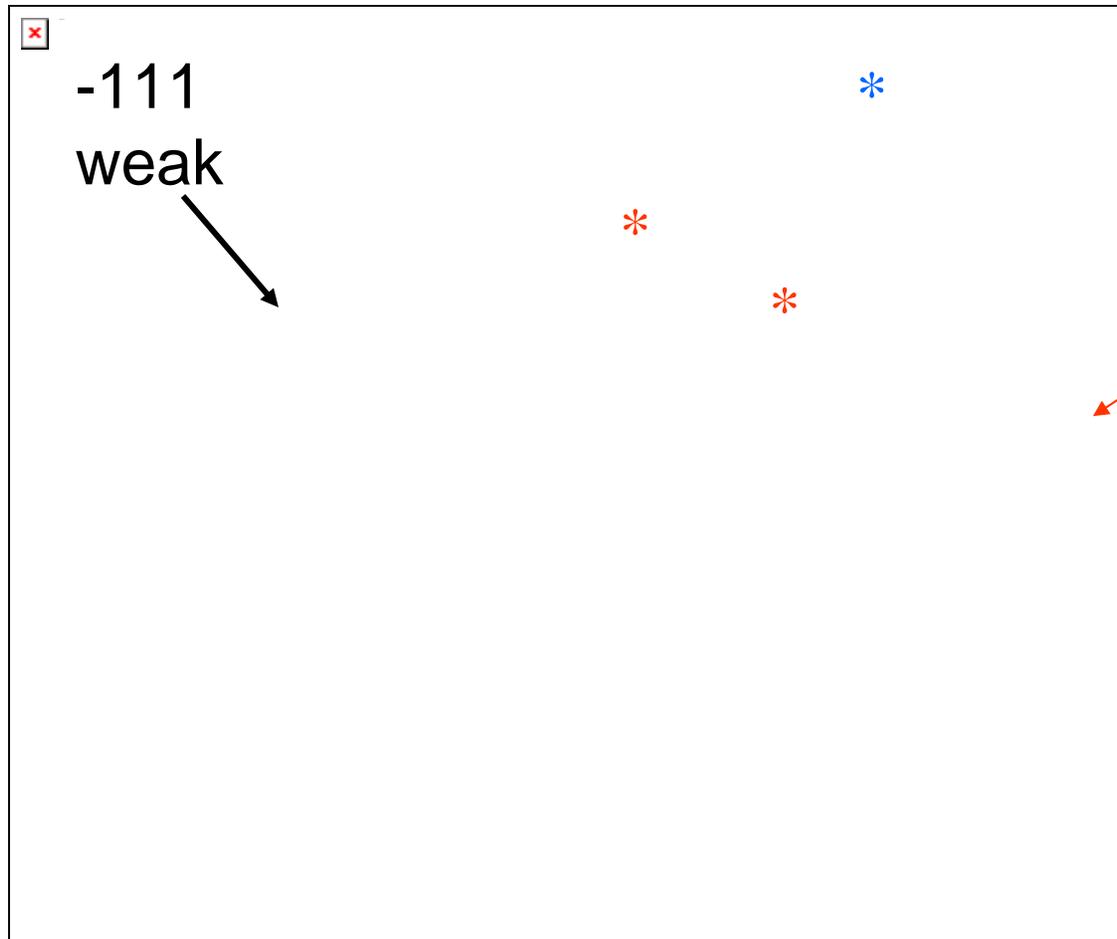


Mar345 Image Plate detector  
and flow controls



Flow Cell with heater  
and TC  
( $T < 1000\text{C}$ ,  $P < 45\text{Atm}$ )  
Clausen, Norby, Chupas

# Example of Catalyst Preparation



- Pre-treat  $\text{NiMoO}_4$  in He at  $500^\circ\text{C}$

- Flow  $\text{CO}/\text{H}_2\text{O}/\text{He}$

Catalysts I

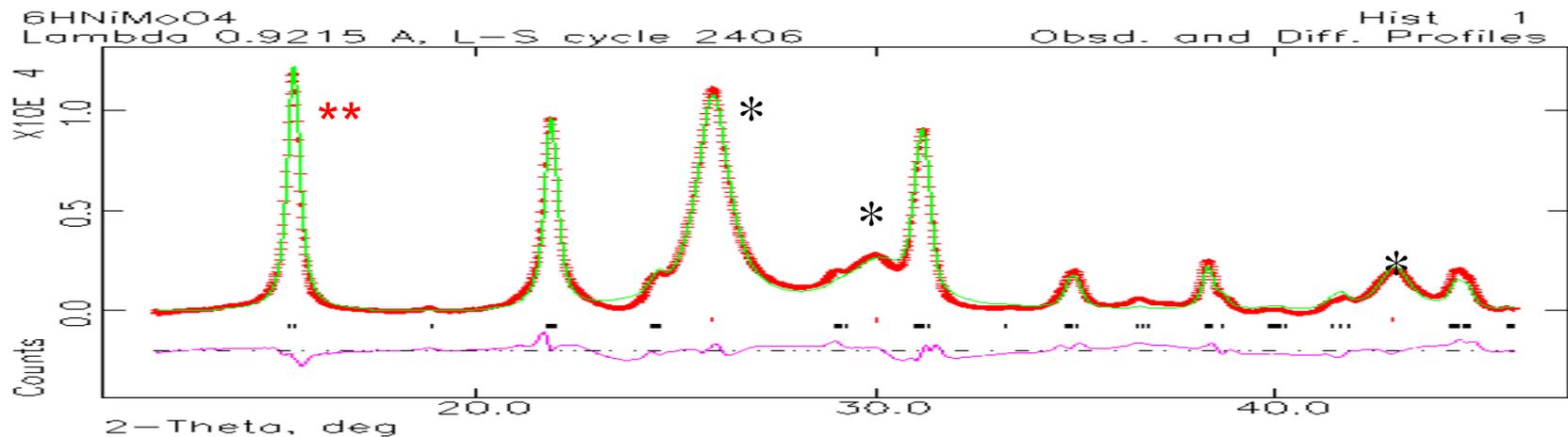
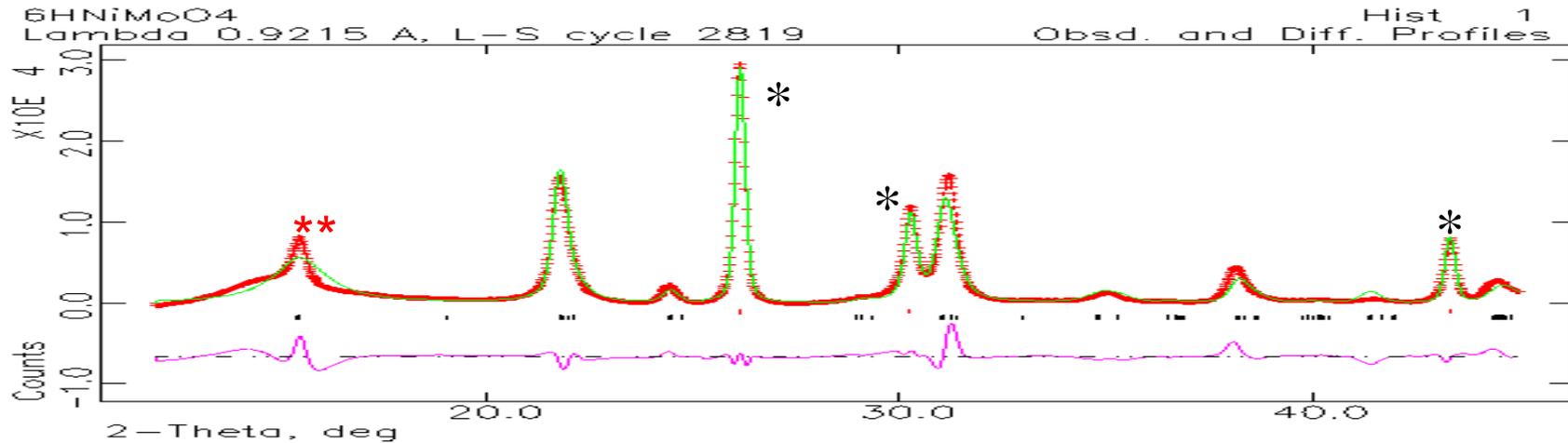
\* Ni

\*  $\text{MoO}_2$

←  $\text{NiMoO}_4$

-111 does not form when the other  $\text{MoO}_2$  peaks form

# Ni/MoO<sub>2</sub> catalysts with different activity



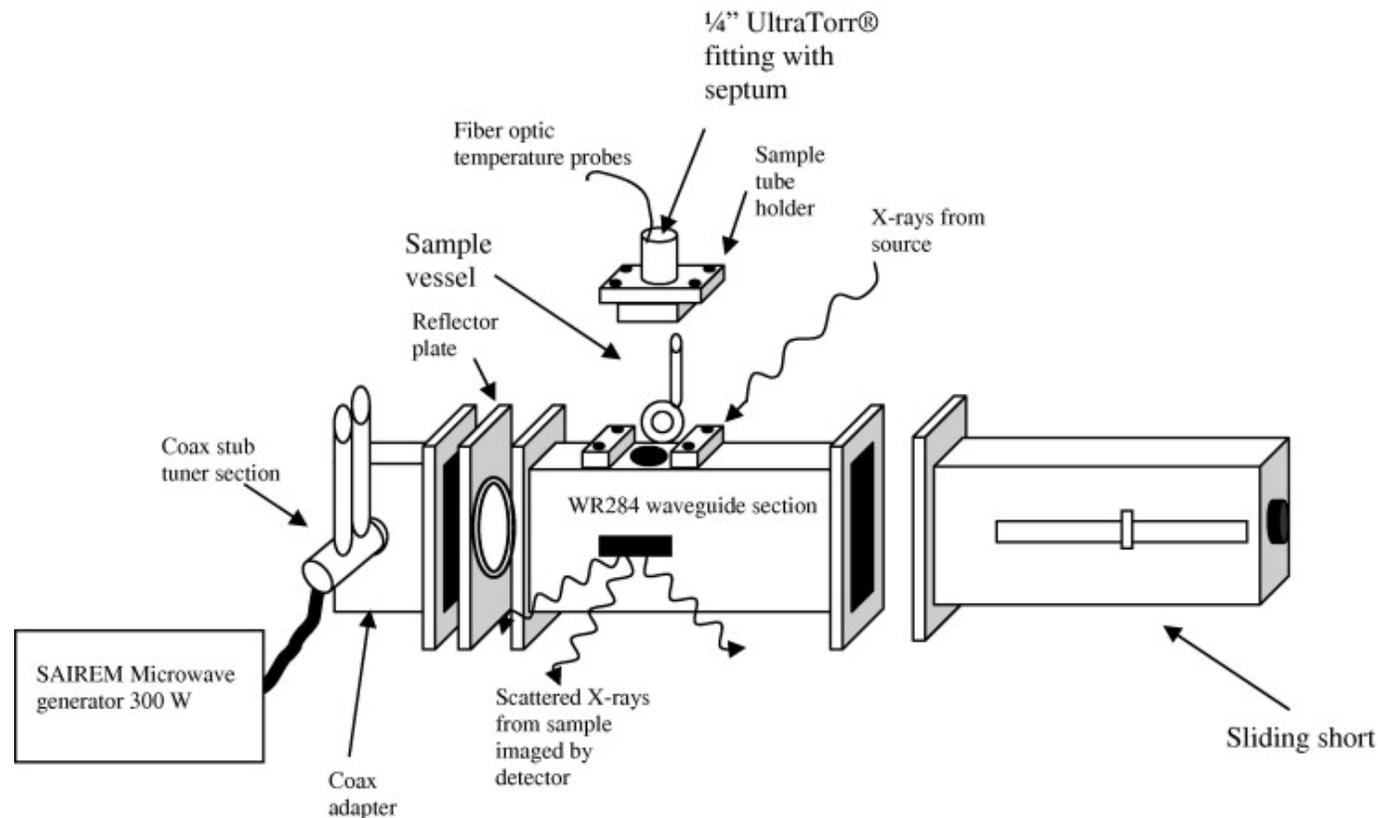
\* Ni

\*\* MoO<sub>2</sub> -111 peak

Stacking faults in MoO<sub>2</sub>

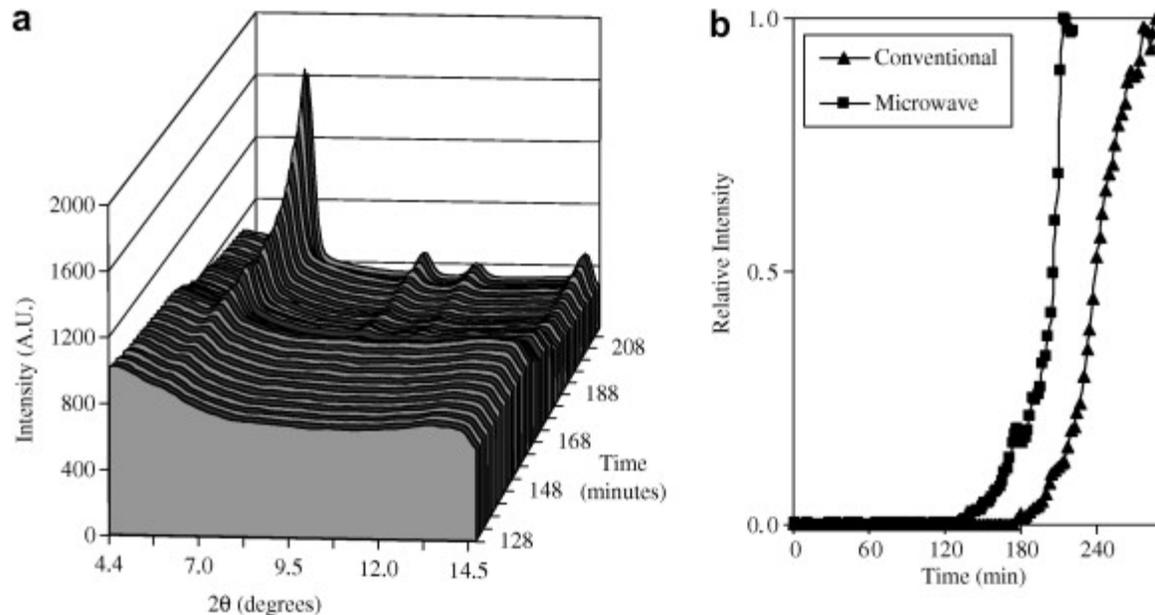
Particle size in Ni

# Microwave cell developed by Conner (UMass) at X10



G Tompsett, B Panzarella, W Conner, S Bennett, K Jones, In Situ SAXS and WAXS of Zeolite Microwave Synthesis, *Nucl. Instrum. Meth. B*, **261**: 863-866 (2007)

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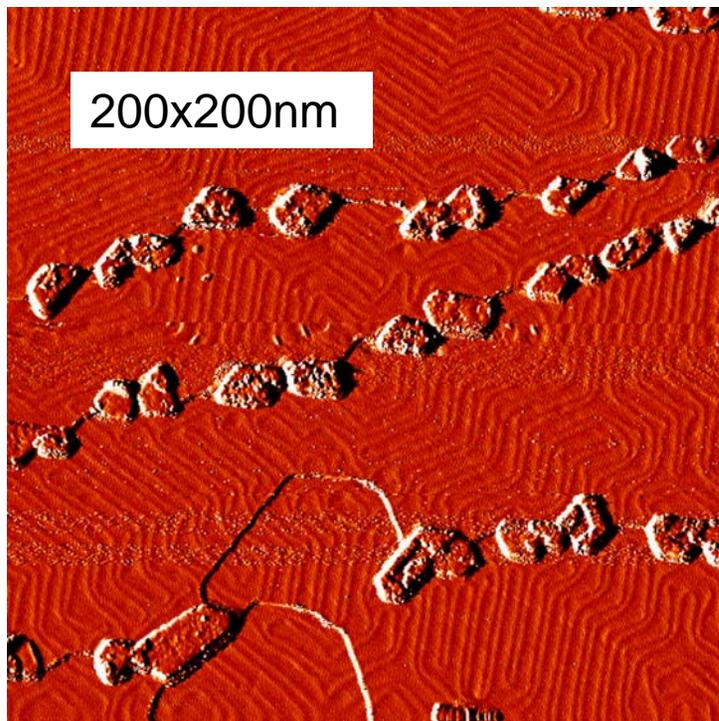
# Diffraction studies by catalysis groups at other synchrotrons

- Ressler @ DESY
  - Methanol reforming
  - Propane oxidation
- Sankar and Thomas @Daresbury
  - Zeolite synthesis
  - Redox studies

# *In Situ* Diffraction from supported nano particles

## Challenge for NSLS II

Inverse Catalyst:  
Ceria on Gold



J. A. Rodriguez et al. *Science* **318**  
1757-1760 (2007)

### Problems

1. Manipulate nano particles into 20nm beam
2. Preferred orientation
3. In situ cell to control/monitor reaction

